









Vial (different coatings) Cartridge

Syringe (Dual chamber syringe)

Requirements of a formulation



Caveat for proteins: Influence on undesirable adverse events and clinical efficiency, immunogenicity and pharmacokinetic profile through product specific degradation products.

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Dose delivery

PDA



Lyo/cryo-protective excipients

Cryoprotectant

Stabilizes during the freezing process

- Excipients are preferentially excluded from the surface of the protein. This is an thermo-dynamically unfavored state. As the unfolded state of the protein would enhance this state, the protein is stabilized.
- (Timasheff 1993).

Lyoprotectant

Stabilizes during the drying process

 Water stablizes a protein in liquid solution by hydrogen bonding. The excipient replaces the hydrogen bonds of water during drying and thus stabilizes the protein.



protein



Lyo/cryoprotective excipients

Crystalline excipients

Ordered crystal structure



Amorphous excipients Glassy state



Eutectic temperature (defined melting point)

Bulking agent

• High eutectic temperature :

- Elegant cake appearance
- Fast drying
- In many cases no stabilization (e.g. for most proteins)
- Different morphologies dependent on excipient (Mannitol→ Annealing)
- Glass breakage (Mannitol at high fill)

Glycin, Mannitol, NaCl, ...

Confiecting reopie, science and negulation

Stabilzation of e.g. proteins

- Acceptable bulking agent at the same time
- Low glass transition temperatures
 → Cake structure?

Sucrose, Trehalose, PVP, Dextran, ...



Examples



Kadcyla 100 / 160mg

20 mg/mL ado-trastuzumab emtansine 10 mM sodium succinate pH 5.0 60 mM D-Sucrose 0.02% Polysorbate

Herceptin 150 / 400 mg

25 mg/mL Trastuzumab 5 mM L-Histidine/-HCl, pH 6.0 60 mM D-Trehalose 0.01 % Polysorbat 20



Analytical characterization

Product attributes for designing lyophilization cycles

- Differential scanning calorimetry: T_g, T_g, T_{eut}
- Freeze drying microscopy: T_{collapse}

Solid state characterization after lyophilization

- Residual moisture (Karl Fischer, NIR)
- Reconstitution time
- Thermodynamic state (Xray powder diffraction)
- Specific surface area (BET)
- Cake appearance at different levels (visual inspection, 3D scanning, PDMS embedding, SEM, μCT)

Other quality attributes of active compound

Differential Scanning Calorimetry (e.g. T_{a'})



Temperature



- Thermal analysis to detect physical transformation such as phase transitions (e.g. glass transition temperature T_{g'}/T_g, crystallization/melting point T_{eut} ...)
- Measurement of the difference in the amount of heat required to increase the temperature of a sample compared to a reference with well-defined heat capacity as a function of temperature
- Both the sample and reference are maintained at nearly the same temperature throughout the experiment

Differential Scanning Calorimetry (e.g. T_{g'})



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Vacuum pump

Freeze drying microscopy (T_{collapse})







(Intact) frozen sample

Onset of collapse

Complete collapse

$$\rightarrow T_{g'} < T_{collapse} !!$$

Residual moisture – Water content



PDA

Gravimetric analysis

Destructive





NR

- multivariate calibration and partial least square

Karl-Fischer Titration

- Two media are needed: Titrating agent and working medium consisting of the three components sulfur dioxide, alcohol, and organic base or/and water free vehicle.
- End-point detection occurs either by color change or potentiometrically via an indicator electrode (free I₂/I- redox couple).

Volumetric Karl Fischer Titration

lodine is added by a burette during titration. Suitable for samples where water is present as a major component: **100 ppm - 100%**





 $CH_3OH + SO_2 + RN \implies (RNH)SO_3CH_3$

 $H_2O + I_2 + (RNH)SO_3CH_3 + 2 RN \implies (RNH)SO_4CH_3 + 2 (RNH)I$

Redox reaction



lodine is generated electrochemically during titration. Suitable for samples where water is present in trace amounts: 1 ppm - 5%

- The working medium consists of the components sulfur dioxide, alcohol, and organic base or/and water free vehicle.
- Two electrodes are needed: One for lodine generation (anode), and one for potentiometric end-point detection via the indicator electrode (free I₂/I- redox couple).







Residual moisture - NIR



- Molecule vibrations (overtone and combinations)
- Near infrared: ~760–2500 nm or 13.000–4.000 cm-1

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Other quality attributes of active compound





- \rightarrow Water ideally flows along the side wall
- \rightarrow Avoid foaming if samples contain surfactants
- \rightarrow In case of long reconstitution times, shaking systems may be considered

Xray powder diffraction - Morphology



The constructive and destructive interference can be measured as different intensities in the X-ray beam at given angles.



- A crystalline powder contains many small crystallites, ideally randomly oriented
- Diffraction occurs when crystallites are oriented such that specific atomic planes are in the correct relationship with the incoming x-rays



Bragg's law: nλ=2dsinθ

Constructive interference is detected when the path-length difference is equal to an integer number of wavelengths

Mixture analysis



Specific surface area (BET)

S.Brunauer, P.Emmett, E.Teller Adsorption of Gases in Multimolecular Layers, J. Am. Chem. Soc., 1938, 60 (2), pp 309–319





- · Physical adsorption of a gas on the surface of the solid.
- Physical adsorption results from relatively weak forces (van der Waals forces) between the adsorbed gas molecules and the adsorbent surface area of the test powder. Thus, the determination is usually carried out at the temperature of liquid N2.
- Traditionally nitrogen is used as adsorbate gas.
- Based on the BET theory, the amount of adsorbed gas corresponds to a monomolecular layer on the surface.
- The amount of adsorbed gas is correlated to the total surface area of the particles including pores.



Sample preparation: degasing under vacuum and elevated temperature followed by measurement in liquid N2.



Visual inspection

Patel et al: Lyophilized Drug Product Cake Appearance: What Is Acceptable? Patel S, Nail S, Pikal M, Geidobler R, Winter G, Hawe A, Davagnino J, Rambhatla Gupta S. J Pharm Sci. 2017 Jul;106(7):1706-1721. doi: 10.1016/j.xphs.2017.03.014.



Intact cake



light collapse/melt-back severe collapse/melt-back complete collapse/melt-back

Cosmetic defects versus impact on product quality?



crack



dents



splashing



fogging





Dex0/Suc100 Dex60/Suc40 Dex100/Suc0





PDMS embedding

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Scanning electron microscopy (SEM)



PDA

Micro-computated tomography (μ CT) Set-up of a CT system and scheme of Data acquisition measurement High speed network Cone Beam Section in which Source 1 1000 Glass vial Sealed plastic cup Object Image Axis of reconstruction Rotation cluster Flat Panel Detector

- A micro-focus x-ray source illuminates the object and a planar x-ray detector collects magnified projection images.
- Based on hundreds of angular views acquired while the object rotates, a computer synthesizes a stack of virtual cross section slices through the object.
- · You can then scroll through the cross sections, interpolating sections along different planes, to inspect the internal structure.
- Selecting simple or complex volumes of interest, you can measure 3D morphometric parameters and create realistic visual models.



μ -CT - Interpretation of reconstructed volume







Pros and cons and applicability of different imaging techniques summarized in Häuser et al: Imaging techniques to characterize cake appearance of freeze-dried Products. J Pharm Sci. 2018.