



EXTRACTABLES & LEACHABLES FOR SVP-INJECTABLES

PDA TRAINING COURSE
EXTRACTABLES – LEACHABLES
BASEL
27 – 28 FEBRUARY 2020

Dr. Piet Christiaens

1. Typical Materials of Construction (MoC's) for SVP C/C

- Rubbers 101
- Glass & Glass related issues for E/L
- COP/COC

2. Container Closure Systems for SVP's

- Vials
- Prefilled syringes
- Cartridges

3. Rubber oligomers – safety concern

LIQUID SVP's

| Examples of Packaging Concerns for Common Classes of Drug Products | | | |
|--|---|---|---|
| Degree of Concern Associated with the Route of Administration | Likelihood of Packaging Components – Dosage Form Interactions | | |
| | High | Medium | Low |
| Highest | Inhalation Aerosols and Sprays | Injections and Injectable Suspensions ; Inhalation Solutions | Sterile Powders and Powders for Injection ; Inhalation Powders |
| High | Transdermal Ointments and Patches | Ophthalmic Solutions and Suspensions; Nasal Aerosols and Sprays | - |
| Low | Topical Solutions and Suspensions, Topical and Lingual Aerosols, Oral Suspensions and Solutions | - | Oral Tablets and Oral (Hard and Soft Gelatin) Capsules; Topical Powders; Oral Powders |

LYO

Revision of “Table 1” in USP <1664>, Originally Included into the FDA Guidance for Industry (1999): “Container/Closure systems for Packaging Human Drugs and Biologics”

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LYO

Remark:

1. the “**Medium**” Likelihood of Packaging - DP Interaction for **Liquid SVP's** is mainly based upon the observation that most Parenteral DP are Aqueous Based. For Non-aqueous based drug products: more caution is needed!
2. The “**Low**” Likelihood of Packaging - DP Interaction for **LYO SVP's** is mainly based upon the observation that:
 1. the interaction between a solid (Lyo cake) a material (eg rubber) is limited
 2. AND, there is limited direct contact between Lyo cake and Rubber closure

However the Mechanism of interaction for a LYO Cake and its MoC may not need always a direct contact.

BE CAREFUL when “rationalizing” a LYO application as being Non Critical!!!

1. Typical Materials of Construction for SVP Container/Closure Systems

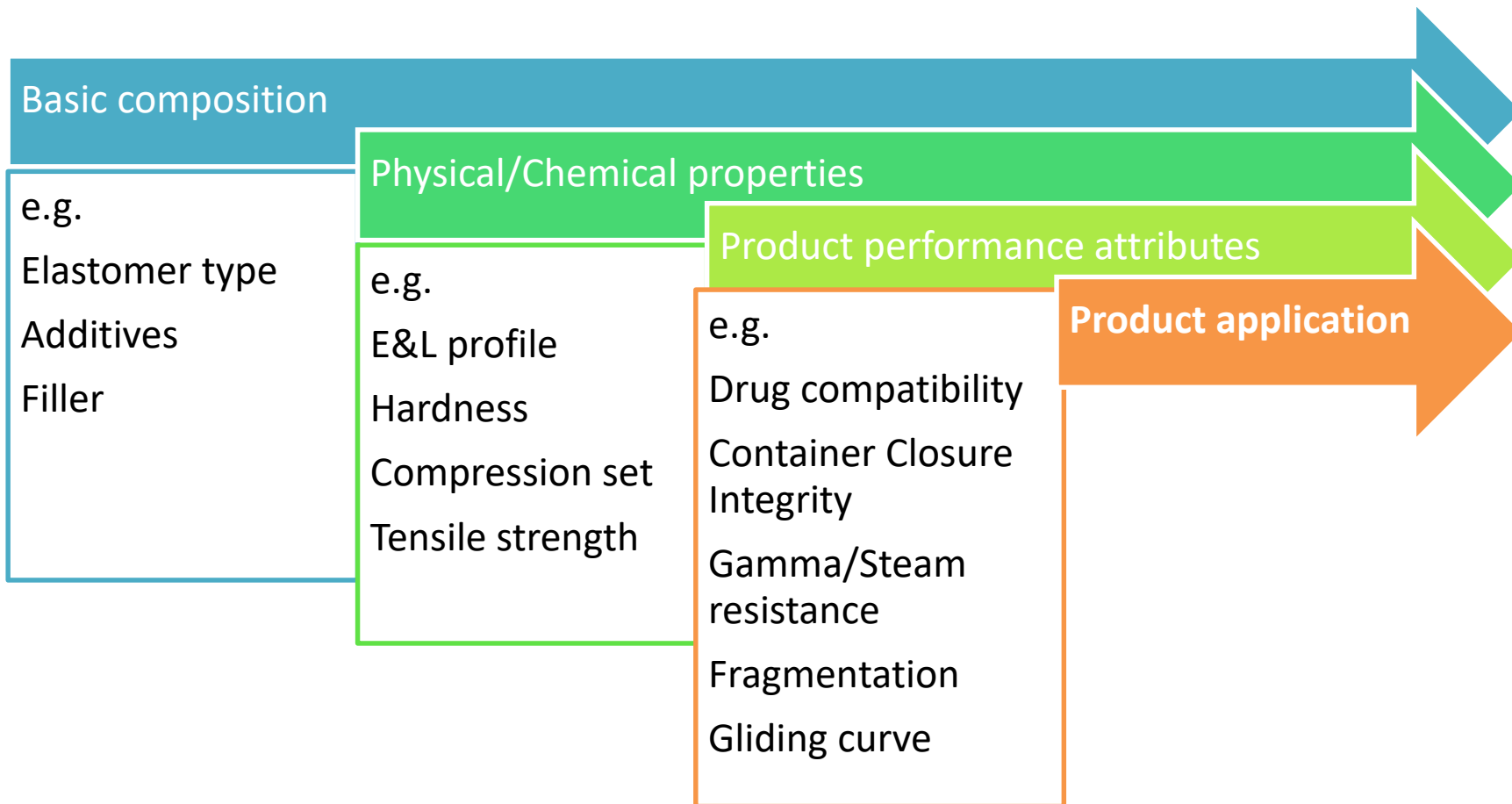


Elastomeric closures

Rubber 101

Supported by Datwyler

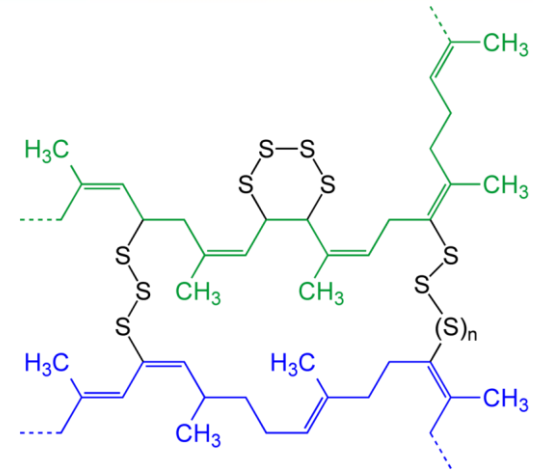




What is rubber?

- An **elastic** material
- A **compounded** material

- Basis of a rubber → polymer → **elastomer**
- **Elasticity via crosslinking** (curing, vulcanising) the elastomer
- Additional ingredients to “**tune**” the rubber



Compounded material of:

1. Elastomer



2. Filler



3. Cure system

4. Pigment

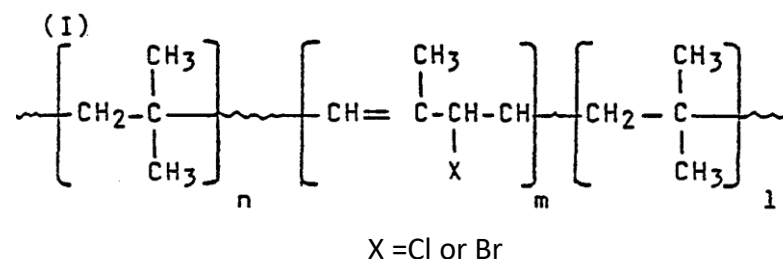
5. Other ingredients



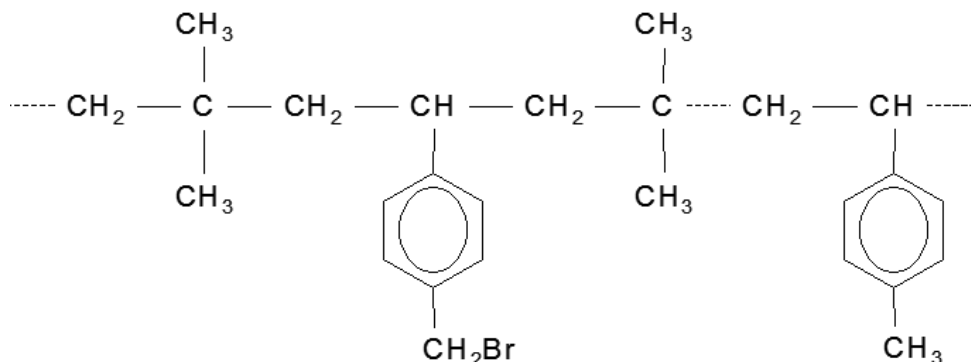
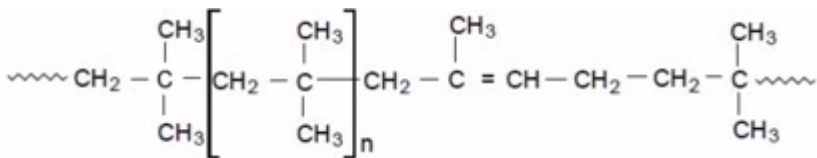
1. Elastomers

Halobutyl (BromoButyl, ChloroButyl)

- Cleanest curing system
- Lowest permeability
- High resistance to ageing



Regular **butyl** still on the market, and also newer types like **BIMS** (Brominated isobutylene para-methylstyrene)



Natural rubber / Polyisoprene

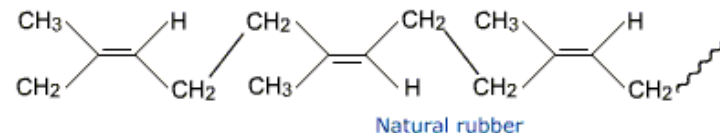
Natural rubber: latex allergy discussions

Historically the oldest elastomer type

Need complex curing systems

Good elastic properties

Polyisoprene (synthetic) replaces Natural rubber



SBR (styrene-butadiene rubber) $\left[\text{CH}_2 - \text{CH} = \text{CH} - \text{CH}_2 \right]_n \left[\text{CH}_2 - \text{HC} \right]_m$

Intermediate permeability

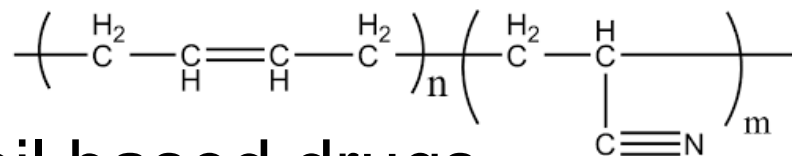
Typically used for pre-assembled EtO sterilized components (e.g. Needle Shields)



1. Elastomers

Nitrile rubber

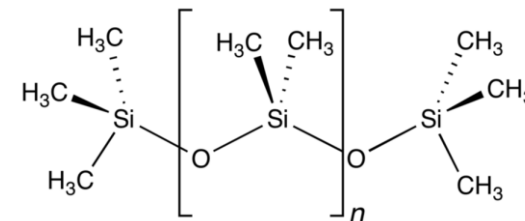
Typically used for mineral oil based drugs



Silicone rubber

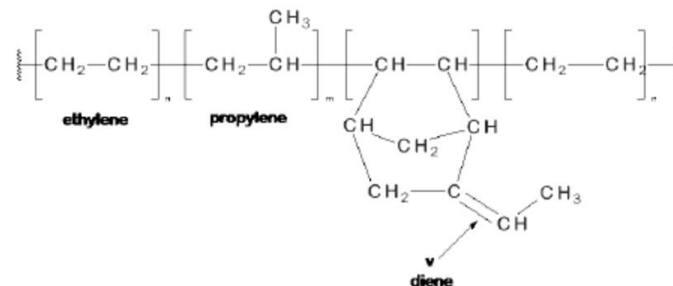
High permeability

Typically not used for parenteral applications



EPDM rubber

For niche applications



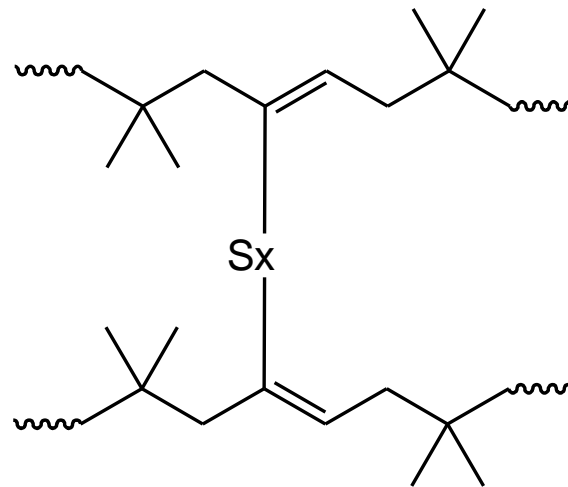
2. Fillers

- Fillers give **mechanical strength** (stiffness) to a rubber
- Attributes **physical properties** to a rubber compound
 - **More filler = Harder compound**
 - Better for **gliding** profile plungers
 - **Worse for stopper piercing** (coring!)
- Inorganic fillers ('white compounds')
 - Aluminum silicate (clay)
 - Magnesium silicate (talc)
 - Silicate
 - [Calcium carbonate]
- Carbon black ('black compounds')
 - Undesired for cleanliness reasons
 - May be associated with PNA's



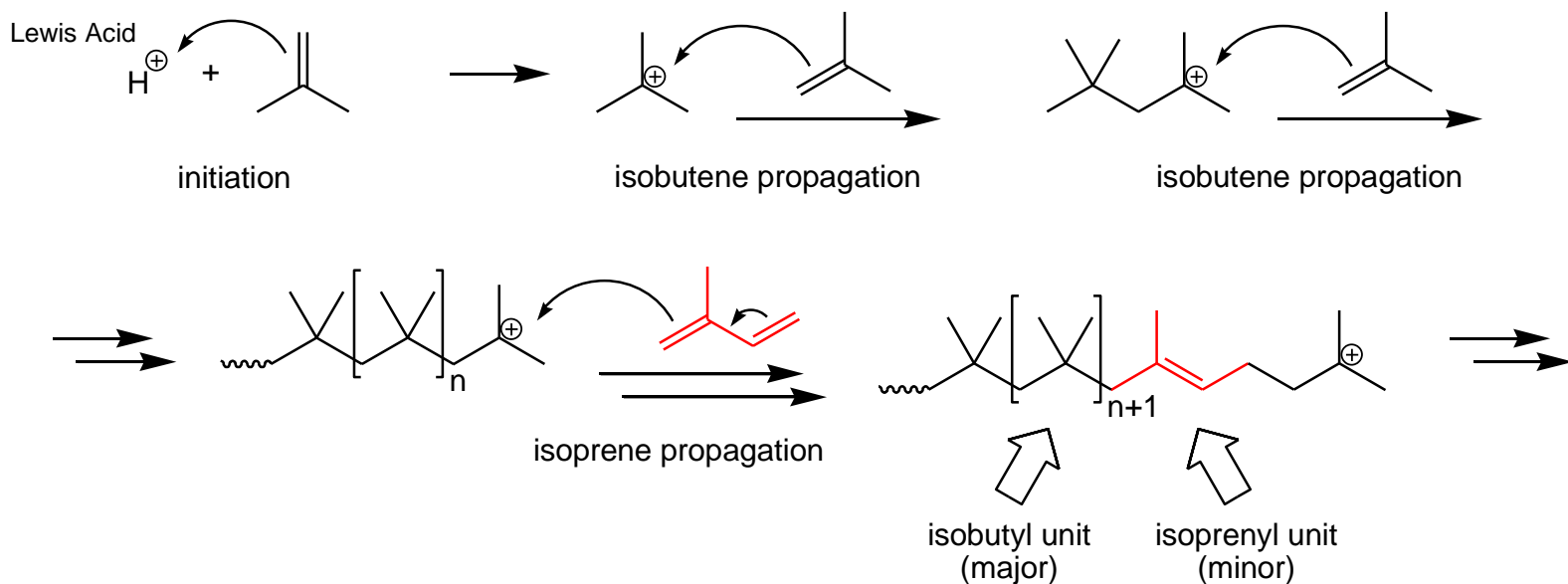
- **Cure system:**
 - Crosslinking agent
 - Activator: gives the onset of vulcanization
 - Accelerator: speeds up the vulcanization
 - Easily extractable organic molecules such as thiurams, sulfonamides, thiazoles, ...
- **Modern cure systems**
 - Aim at giving little extractables
- **Historic cure systems**
 - Use easily extractable organic accelerators

Rubber Curing / Vulcanization:



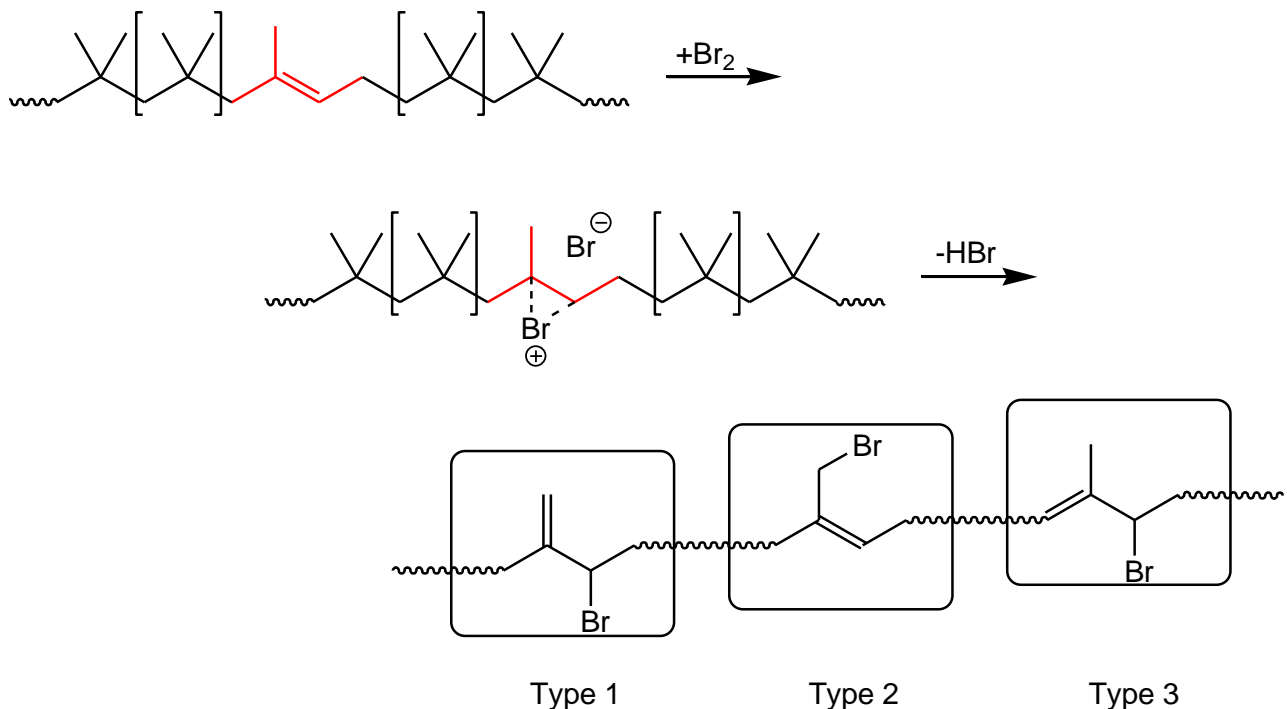
Rubber crosslinking requires S-Donors, activators, accelerators
Activator: ZnO / Stearic acid

Formation (polymerization) of a Butyl Elastomer (IIR): Cationic Polymerization



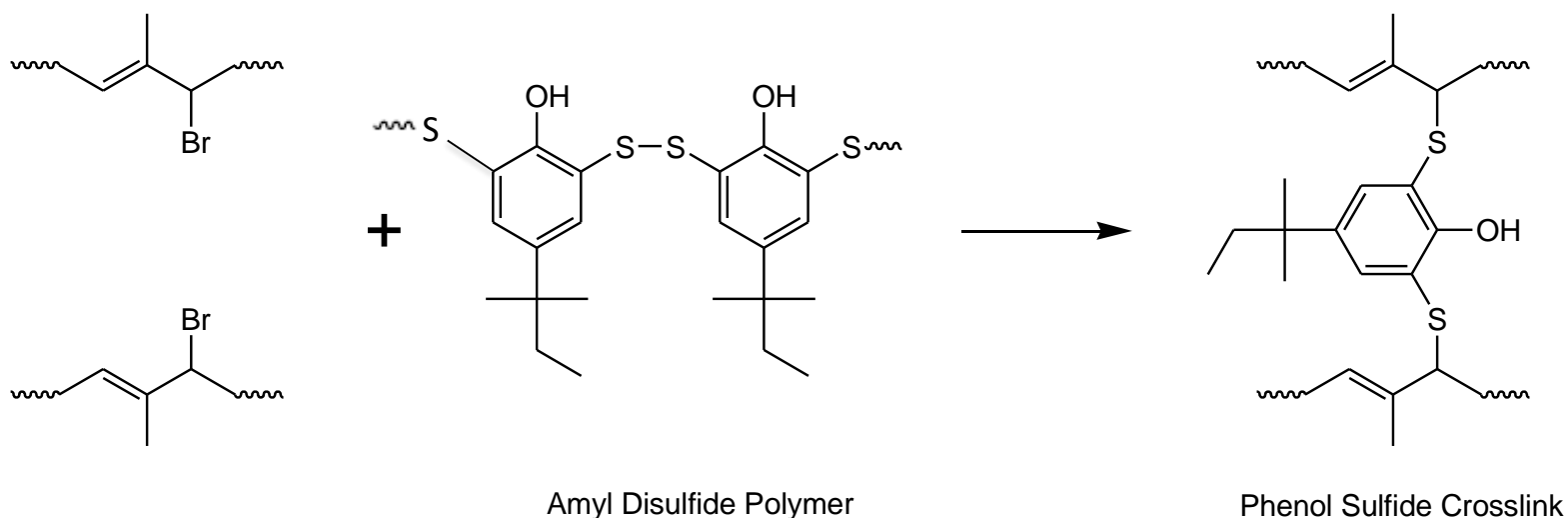
- Note: the Polymerization Starts with a Isobutene Unit (present in high excess!!)
 - 98 – 99 mol% is isobutylene
 - 1 – 2 mol% is isoprene

Bromination of a Butyl Elastomer (BIIR)



***Bromination of the Backbone makes Elastomer
(with a relatively Low N° of double bonds in backbone)
more reactive in vulcanization/cross linking***

Vultac Curing of (Halobutyl) Elastomers

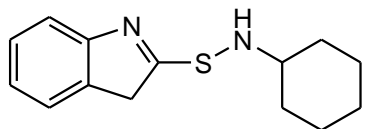


Bromide: good leaving group!

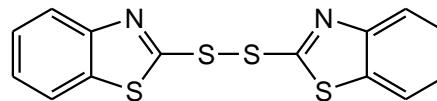
Bond Energy C-H 413 J/mol ⇌ C-Br 209 J/mol

Explains Br⁻ release from bromobutyl rubbers

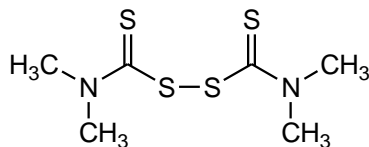
Rubber Curing - Accelerators:



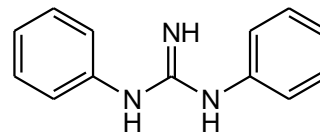
Cyclohexyl benzothiazole sulfenamide



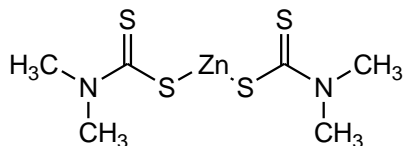
Mercaptobenzothiazole disulfide



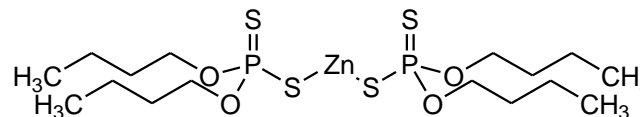
Tetramethylthiuram disulfide(TMTD)



Diphenyl guanidine

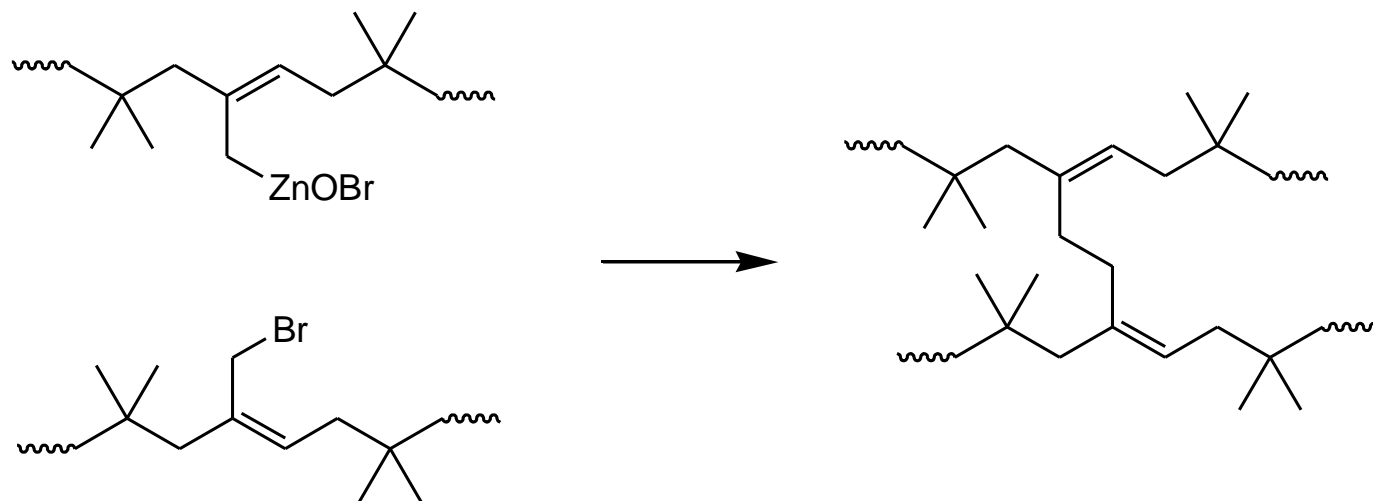


Zinc dimethyldithiocarbamate



Zinc dibutylphosphorodithiate

ZnO as Cross-Linking Compound in Halobutyl-Rubbers:



4. Pigments

- **Inorganic pigments**
 - Titanium dioxide
 - Traces of carbon black
 - Oxides of iron

- **Organic pigments**
 - Avoided in modern compounds

- **Halobutyl polymer stabilizers**
(to prevent dehydrohalogenation during processing)
 - Calcium stearate
 - Epoxydized soybean oil
- **Anti-oxidants**
 - Already present in halobutyl elastomer
 - Hindered phenol type anti-oxidants
 - Additionally added to improve environmental stability (ageing)
- **Plasticizer, Waxes, Oil**
(introduce softness, anti-"coring")
 - High polymeric weight plasticizers, Paraffinic oil
 - To tune a formulation (e.g. reduce coring)
- **Processing aids**



Smart selection of ingredients can tune a rubber compound

E.g. recipe based on hydrophobic ingredients will show better E-profile with aqueous drugs

E.g. blend of halobutyl and SBR can tune the permeability

E.g. MgO replaces ZnO to avoid Zn-ion extraction

E.g. low water absorption compounds for lyo applications



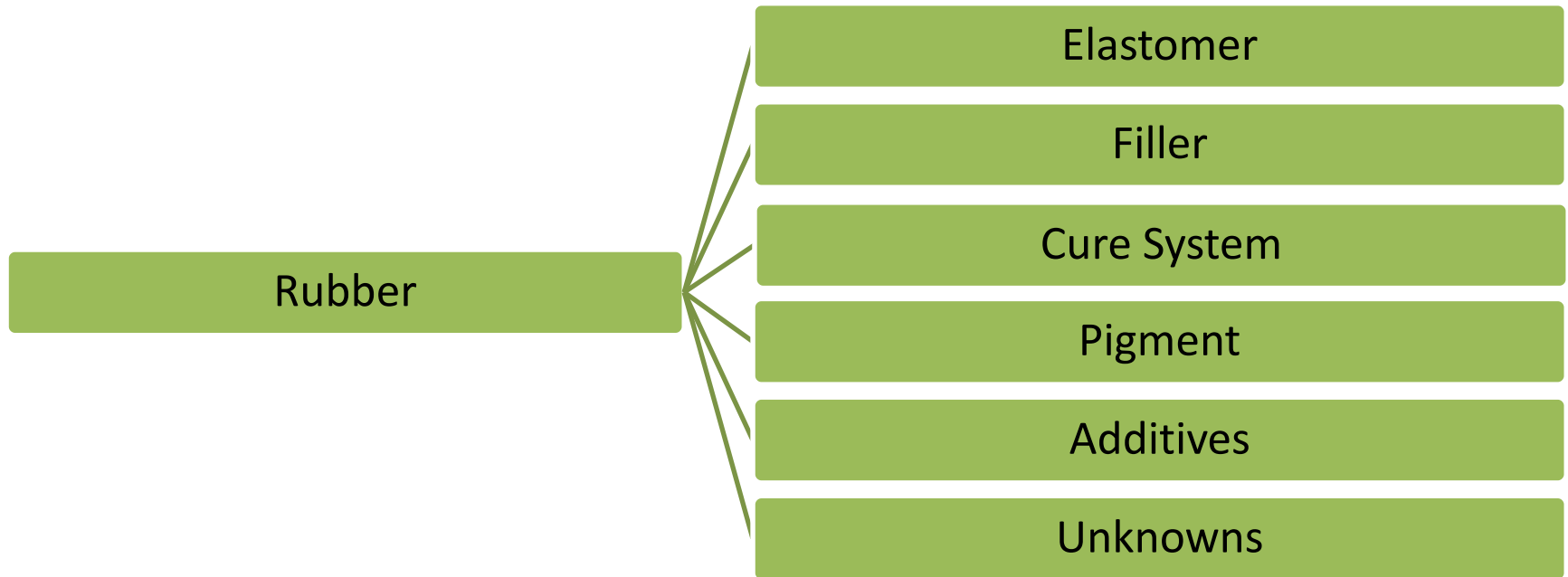
But in general too many ingredients should be avoided: negative impact on E-profile

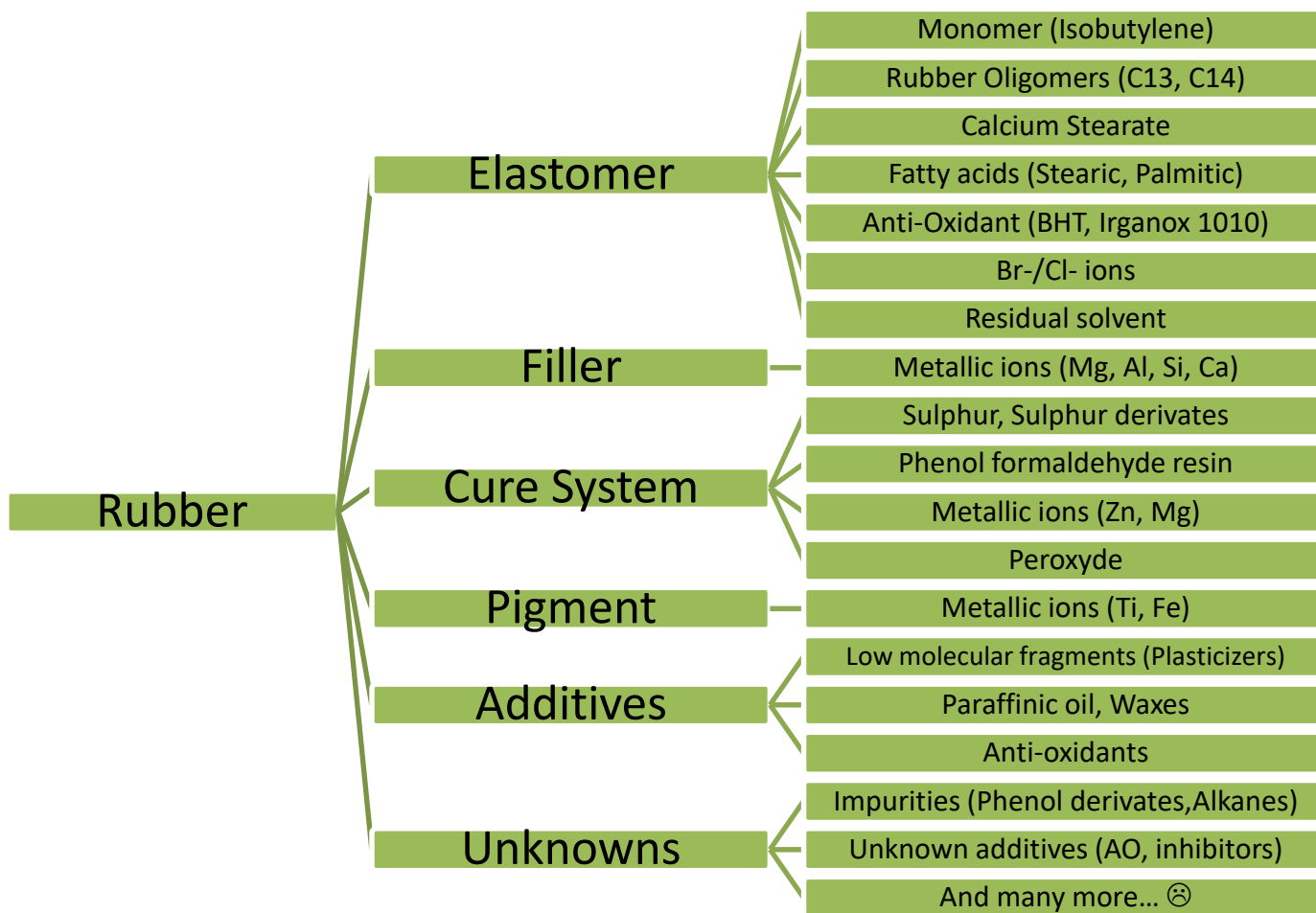
→ *“what you don't put in, can't come out”*

THE COMPOSITION OF RUBBERS CAN BE VERY COMPLEX!!

RUBBER EXTRACTABLES: SUM OF

1. **INITIAL INGREDIENTS** OF THE RUBBER FORMULATION
2. **IMPURITIES** OF THESE INGREDIENTS
(e.g. Residual Solvents, **Oligomers in Elastomer**, Halides in Halobutyl Rubber...)
3. **REACTION/DEGRADATION PRODUCTS** DURING RUBBER PRODUCTION

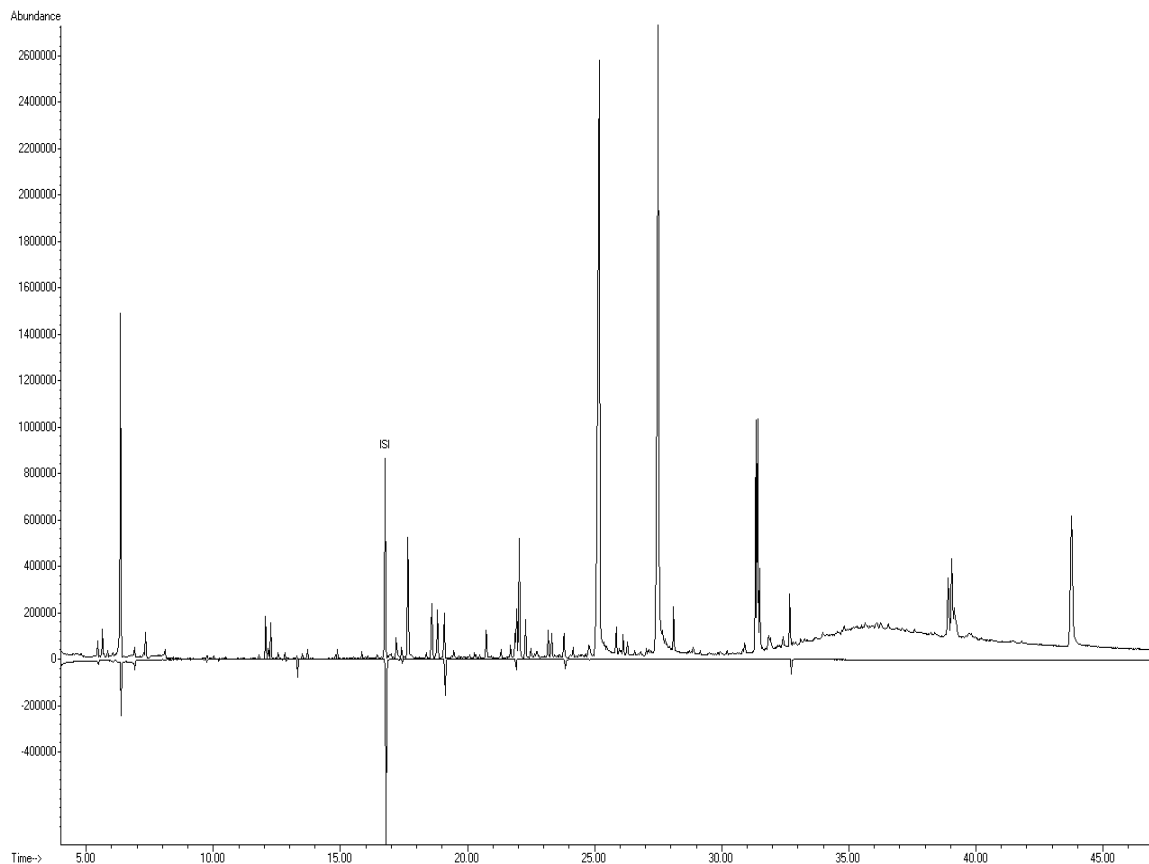




Number of Leachables from rubbers in PFS is determined by:

- The Type of **Rubber Formulation**
- The **Number of Ingredients** in the Rubber
- **Type** of Ingredients (type of vulcanisation, type of AO, stabilizer....)
- **Coated/Non-coated** rubbers
- The **composition of the Medicinal Product (MP)**
- The **type of contact** between the rubber and the MP
(*e.g. exposed surface area*)
- The **Storage Temperature**
- The **Storage Time** (Expiration Date)

Difference in Extractable Results for an **OLD** vs **NEW** rubber
(IPA Extract; GC/MS analysis)



“OLD” RUBBER

“NEW” RUBBER

COATED RUBBERS

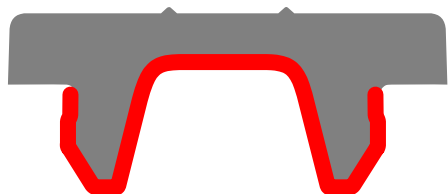
Coated closures: significant improvement in E&L terms

Key attribute: barrier effect from the fluoropolymer!

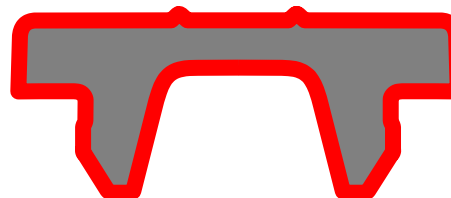
Simplified extractables profile

Improved compatibility with drugs/excipients

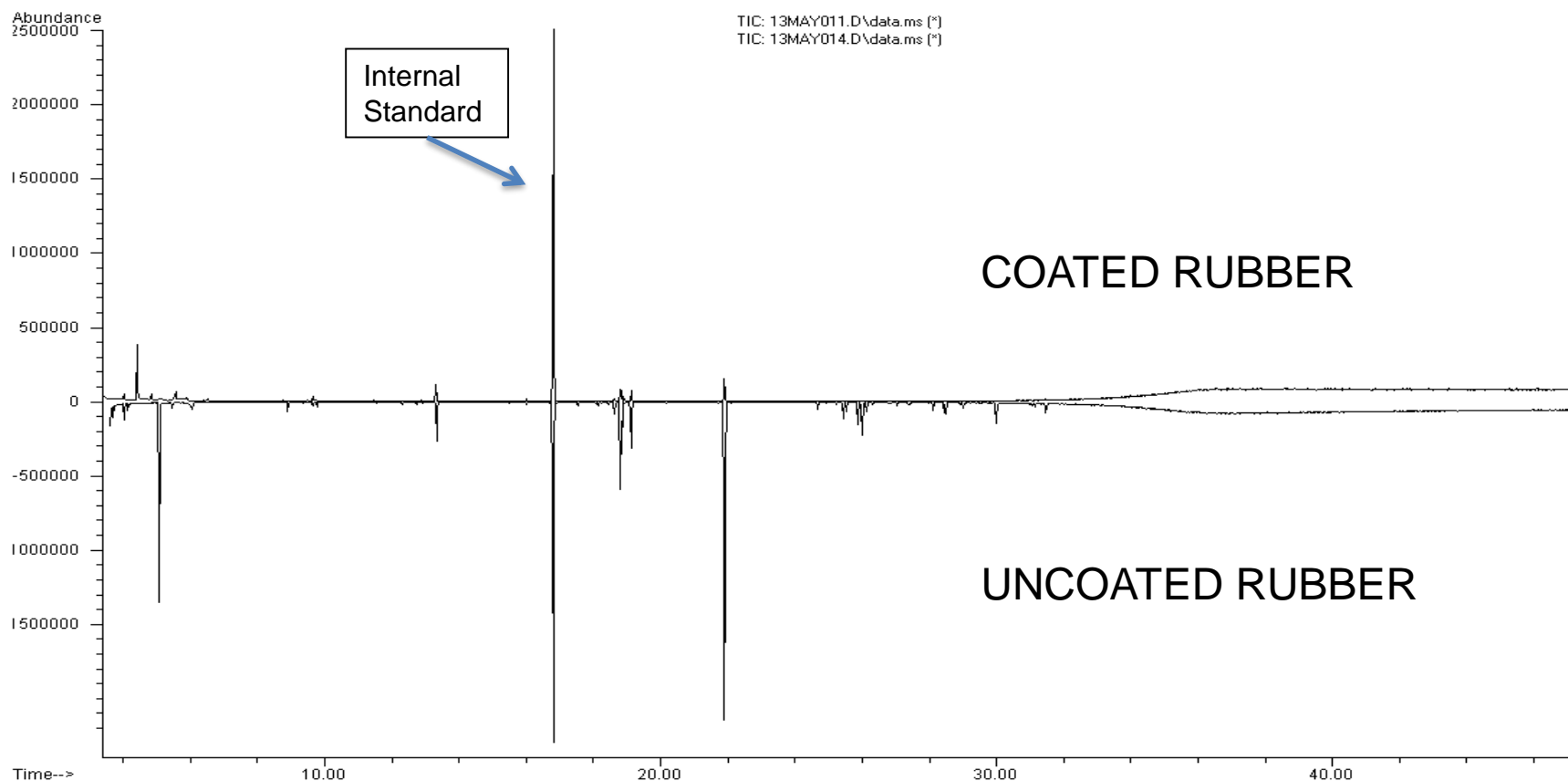
Film coating technology



Spray coating technology



Difference in Extractable Results for a **Coated vs Uncoated rubber**, for the same rubber grade (*IPA Extract; GC/MS analysis*)



Glass

&

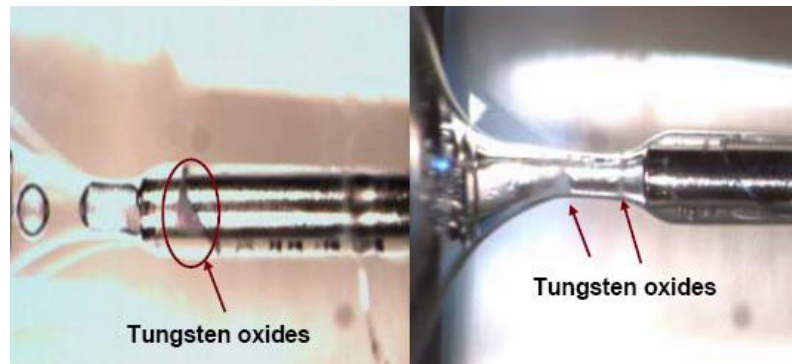
Glass Related Issues

Vials, Prefilled Syringes, Cartridges

Glass as Barrel Material

TUNGSTEN RESIDUES

- Tungsten pin used in the production of glass pre-filled syringes to open the syringe hub (cavity where staked needle is glued in)
- Tungsten Oxide Residues are known to cause protein degradation (protein oxidation causing aggregation)



Glass as Barrel Material

GLUE RESIDUES

- *Glue is used to glue in the staked needle into the PFS-system*
- *Prolonged contact with a drug product may release glue components*
- *Target compounds may depend upon the glue used (e.g. Loctite 3345, Loctite 3081, or other grades)*



Glass as Barrel Material – Related Compound

1. PRODUCT AND COMPANY IDENTIFICATION

Product name: Loctite 3345
Product type: Ultraviolet adhesive

IDH number: 256930
Item number: 33417
Region: United States

Company address:
Henkel Corporation
One Henkel Way
Rocky Hill, Connecticut 06067

Contact information:
Telephone: 860.571.5100
MEDICAL EMERGENCY Phone: Poison Control Center
1-877-671-4608 (toll free) or 1-303-592-1711
TRANSPORT EMERGENCY Phone: CHEMTREC
1-800-424-9300 (toll free) or 1-703-527-3887
Internet: www.henkelna.com

3. COMPOSITION / INFORMATION ON INGREDIENTS

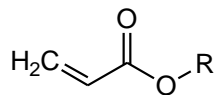
| Hazardous components | CAS NUMBER | % |
|--|-------------|---------|
| Polyurethane Methacrylate Resin | Proprietary | 30 - 60 |
| Tetrahydrofurfuryl methacrylate | 2455-24-5 | 10 - 30 |
| Hydroxyalkyl methacrylate | 27813-02-1 | 10 - 30 |
| Acrylic acid | 79-10-7 | 5 - 10 |
| High boiling methacrylate | 7534-94-3 | 5 - 10 |
| Propylidynetrimethyl trimethacrylate | 3290-92-4 | 1 - 5 |
| Gamma-glycidoxypropyl trimethoxysilane | 2530-83-8 | 1 - 5 |

Glass as Barrel Material – Related Compounds

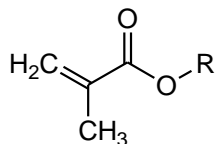
EXTRACTABLES RELATED TO GLASS BARRELS:

GLUE RESIDUES

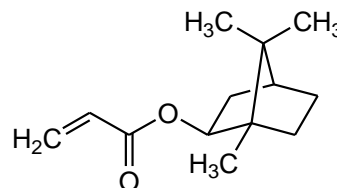
Base Polymer



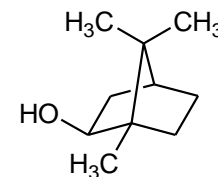
Acrylate



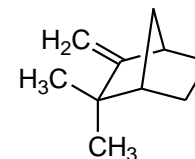
Methacrylate



Isobornyl acrylate

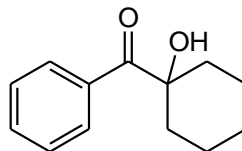


Isoborneol

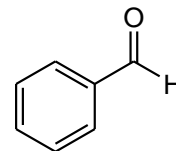


Camphene

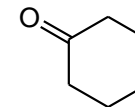
UV curing / activation



Irgacure 184



Benzaldehyde



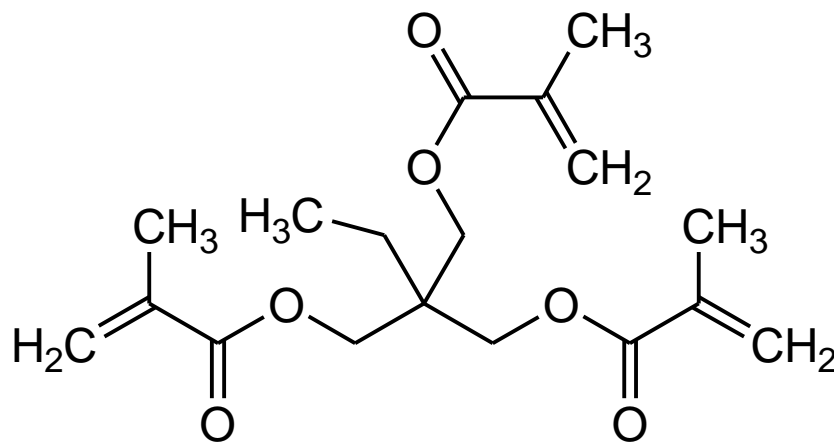
Cyclohexanone

Glass as Barrel Material – Related Compounds

EXTRACTABLES RELATED TO GLASS BARRELS:

GLUE RESIDUES

The key indicator compound TMPTMA



Trimethylolpropane trimethacrylate

Glass as Barrel Material – Related Compounds

SILICONE OIL RESIDUES

- *Glass surfaces are siliconized a.o. to reduce potential interactions with aqueous contact solutions*
- *Hydrophobic surface / reduced wettability*
- *Reduced alkali release*
- *Silicone oil remainders become leachables*

Barrel Materials

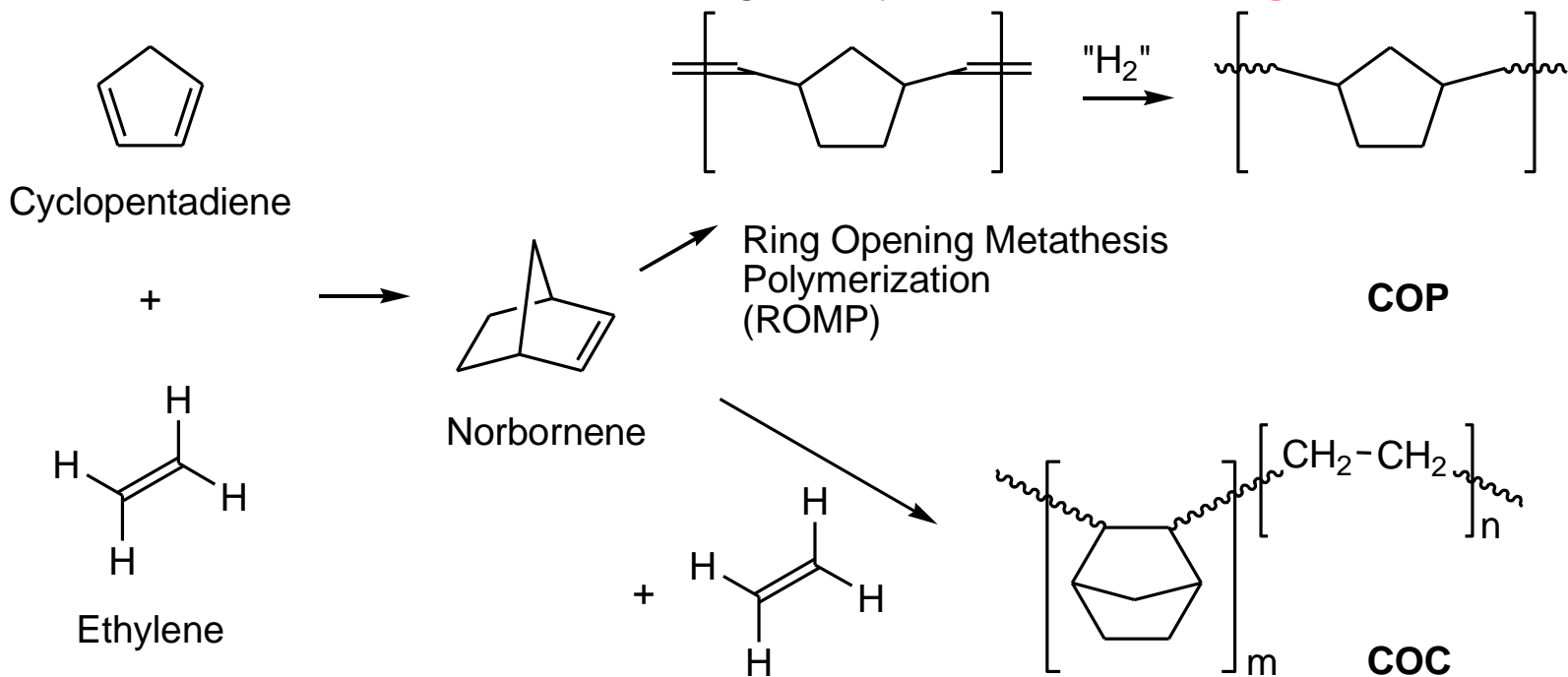
Polypropylene (PP)

Cyclic Olefin (Co-)Polymer COC/COP

Glass

COP: Cyclic Olefin Polymers COC: Cyclic Olefin Copolymers

- Relatively **Clean** Materials
- **High Tg, rigid** materials
- However, **low gas barrier** (O₂) properties
- **Risk for diffusion**: potential (regulatory) risk for **label migration**



CRITICAL PARTS OF A **POLYMER SYRINGE** WRT E/L

PRIMARY PACKAGING (Direct Contact between DP and Material):

- The Barrel: COC, COP, PP
- The Piston: Rubber
- The Tip Cap: Rubber
- The Needle

Same Concern as for Glass PFS

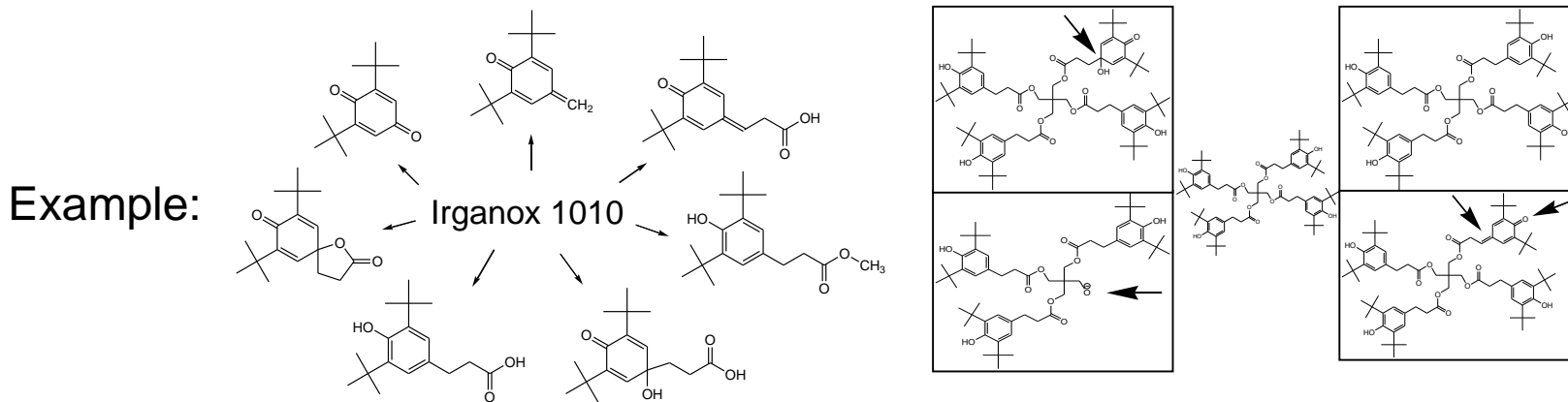
SECONDARY PACKAGING (No Direct Contact between DP and Material):

- *The Needle Shield (should it be considered as primary or secondary?)*: Rubber
- **The Label**: Adhesive, Ink, other Label Components
- In some Cases: The Lacker
- In some Cases: The Packaging of the Syringe (Overwrap, Tubs,...)

Specific for
Polymer PFS!

TYPICAL COMPOSITION OF **COMMERCIAL POLYMERS**, e.g. For Barrel Manufacture

- Additives (*BHT, Irganox 1010, Stearates, Pigments, Clarifiers...*)
- Residues (*Monomers, Solvent Residues, Processing Residues..*)
- Oligomers (*Mainly for PP*)
- Potential Degradation Compounds from Polymers
 - *Organic Acids, Aldehydes, Ketones, Alcohols, Chain Scission Fragments...*
- Degradation Compounds from Polymer Additives



SECONDARY PACKAGING

➤ **Label**

- Adhesive
- Paper
- Ink
- Varnish

Typical extractable compounds:

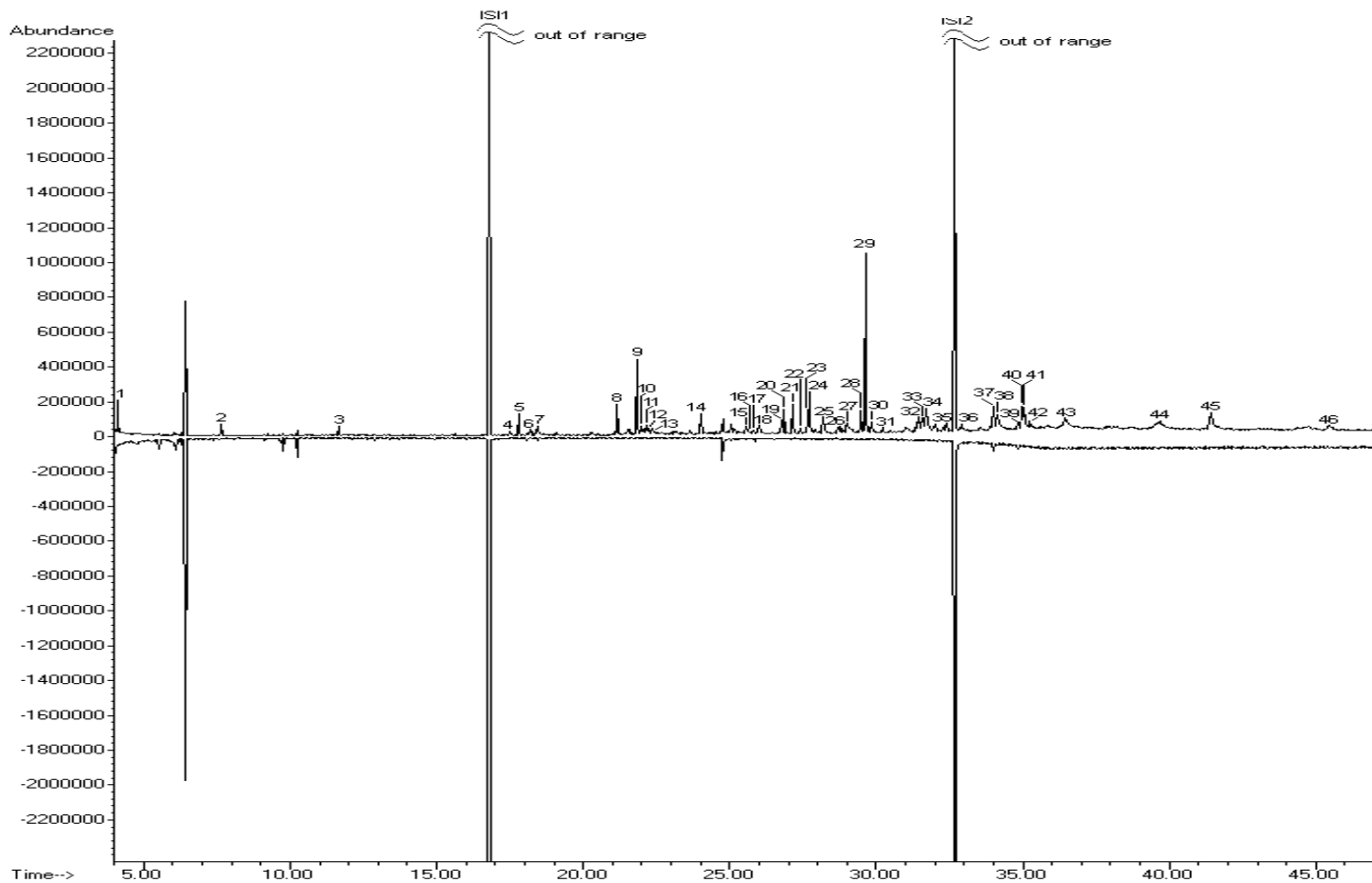
Curing agents (e.g. Benzophenone, Irgacure 184,...)

Solvent residues (e.g. Toluene, acetone)

Adhesive residues (e.g. Acrylates)

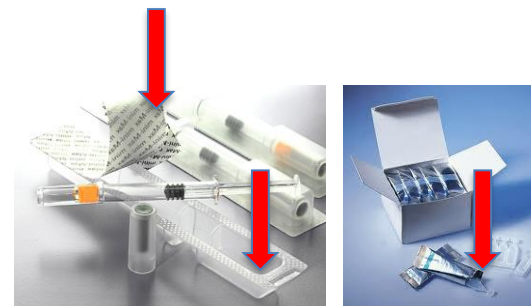
Paper residues (e.g. (dehydro)abietic acids, abietates)

Example GC/MS Chromatogram of a Label Extract (IPA)



SECONDARY PACKAGING

➤ Overwrap/Overpouch/Blister



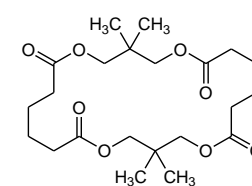
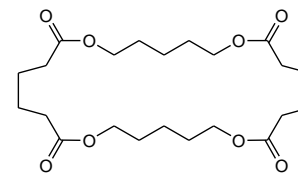
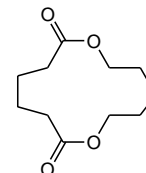
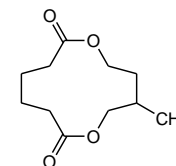
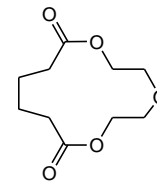
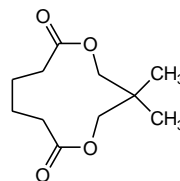
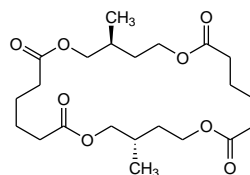
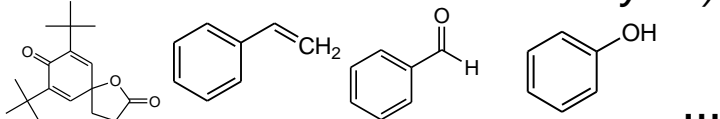
(to compensate for potential lower barrier properties of the Polymer)

- Multilayer System
- Aluminum as barrier layer
- Tie-layers to keep the different layers together

Typical extractable compounds:

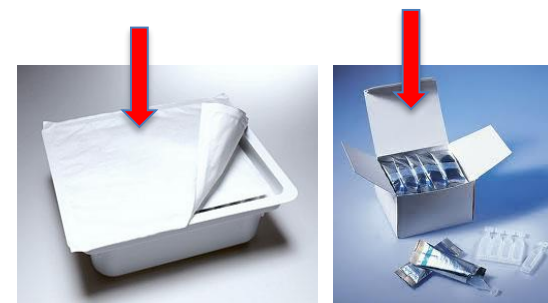
Bislactone Compounds from Tie-layer

Residues from other layers (depends largely on selected materials of the multilayer!!)

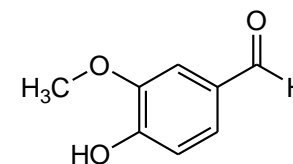
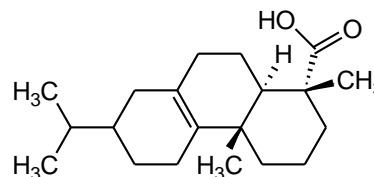
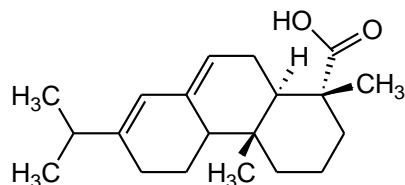
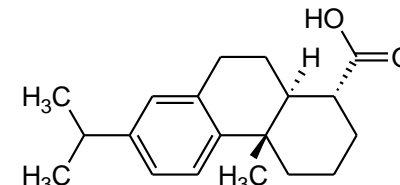
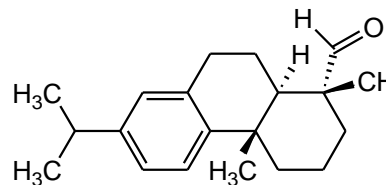
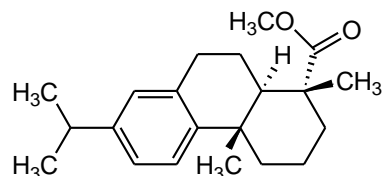
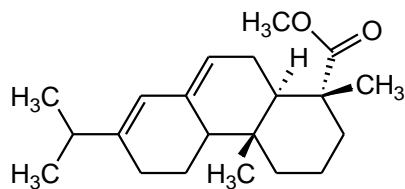
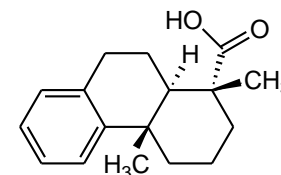
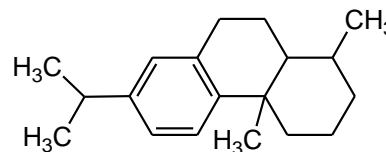
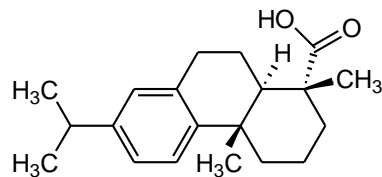


SECONDARY PACKAGING

- **Tubs** for Nested Syringes (eg Tyvek)
- **Carton / Paper** (may also from label):



Example Structures of abietic acids / abietates (& Vanillin)



2. Container/Closure Systems for SVP's: Information Relevant to the Design of an E&L Study

1. Vials:



VIALS for Liquid Drug Products or Reconstitution Solution



- If it is a **GLASS VIAL with RUBBER CLOSURE**: Sources of Impurities, coming from packaging:
 - **Glass**: Metals (may not be necessary to be studied in EXT Study, if glass composition is available, direct assessment in LEA study)
 - **Rubber Closure**:
 - ✓ Typically, higher migration when solution is in contact (inverted)
 - ✓ Migration will be determined by:
 - Solubility of leachables in **Drug Product** Solution
 - Potential Diffusion of Compounds through rubber, into solution
 - Temperature
 - ✓ VOC, SVOC and NVOC & some metals may cause a **Safety Issue**
 - ✓ VOC, SVOC, NVOC, Silicone Oil and some Metals may also be **Reactive**
e.g. with reconstituted DP: also potential Performance & **Quality Issue!**
 - ✓ Also, Ions may need to be “checked off”...

LYO-CAKE VIAL



- Sources of impurities, coming from packaging

- **Glass:** Metals (may not be necessary to be studied in EXT Study, if glass composition is available, direct assessment in LEA study)

- **Rubber Closure:**

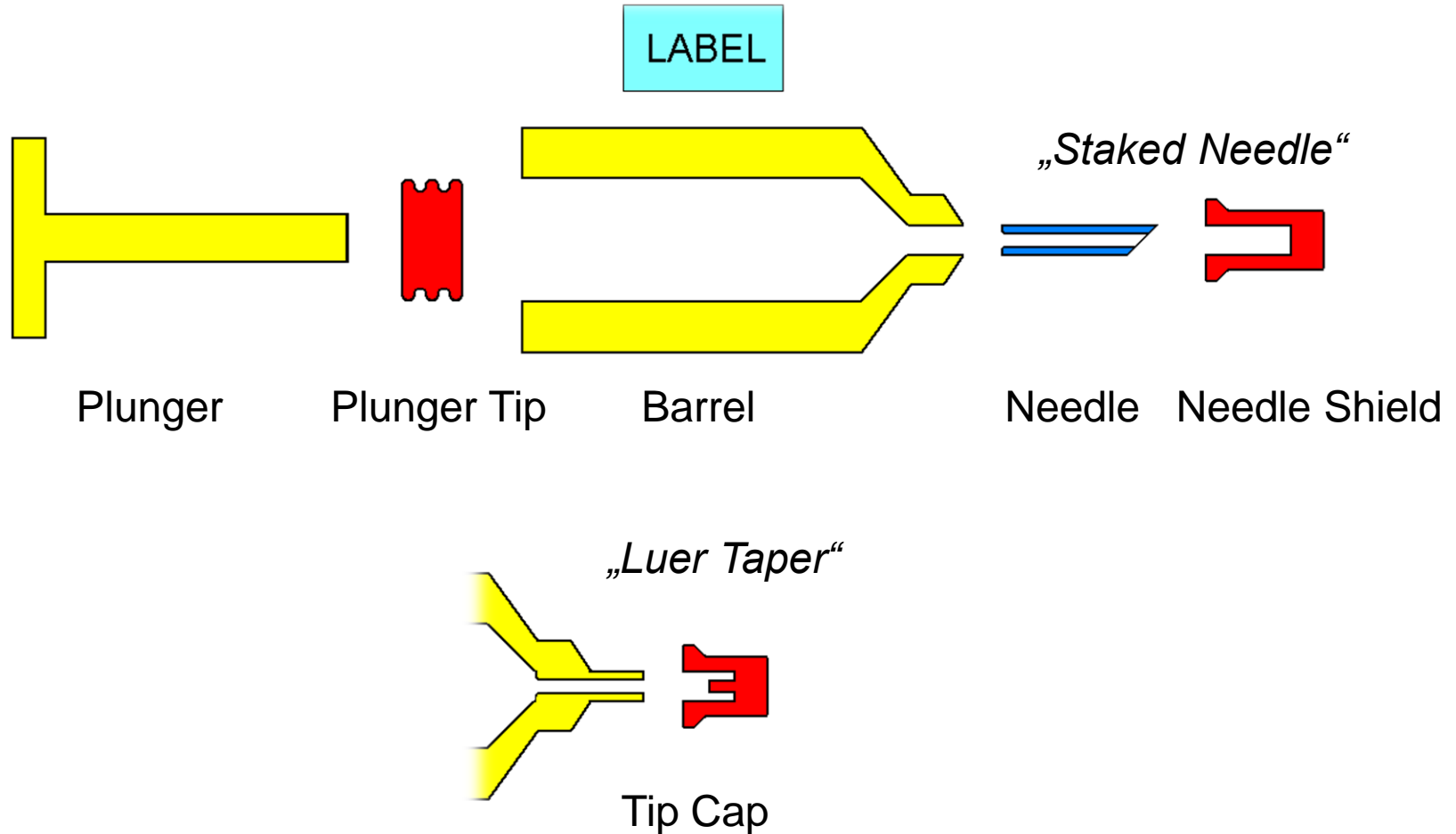
- ✓ **No Direct Contact** between DP and Closure (upright)
- ✓ **HOWEVER:** Release of Volatile (VOC) and Semi-Volatile (SVOC) Compounds from the Rubber Closure vial desorption and subsequent adsorption of compounds onto Lyo-Cake!
- ✓ **Lyo-cake acts as adsorbent for VOC and SVOC compounds! Released Compounds are concentrated over time onto the Lyo Cake**
- ✓ **Regardless** if vial is in **upright or inverted** position (contact / no contact with DP)
- ✓ VOC and SVOC may also be **Reactive** with DP: also potential Performance & Quality Issue!
- ✓ Also NVOC, Metals and Ions need to be "checked off", because of short term contact with Reconstituted DP.

2. Pre-Filled Syringe:



2. Container/Closures for SVP's

PRE-FILLED SYRINGE: COMPOSING PARTS



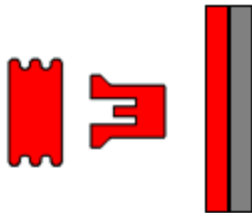
2. Container/Closures for SVP's



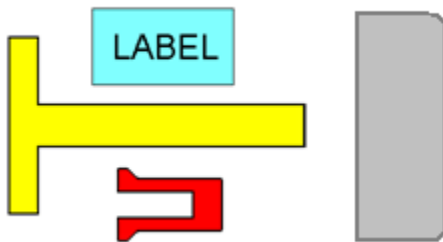
BARREL –
Glass, COC/COP, PP, Silicone Oil, ...



NEEDLE –
Metals, Tungsten (W), Needle Glue, ...



RUBBER SEALINGS (Plunger Tip, Tip Cap, Disks) -
Rubber, Silicone, ...



SECONDARY (Needle Shield, Label, Stem, ...) –
Rubber, Label Adhesive, ...

Pre-Filled Syringes



- **BARREL:** *Metals* (may not be necessary to be studied in EXT Study, if glass composition is available, direct assessment in LEA study)
Silicone Oil residues may cause protein aggregation
- **Rubber Plunger (*very similar to rubber stopper for vial*):**
 - ✓ Typically, higher migration when solution is in contact
 - ✓ Migration will be determined by:
 - Solubility of leachables in **Drug Product Solution**
 - Potential **Diffusion of Compounds through rubber**, into solution
 - **Temperature**
 - ✓ VOC, SVOC and NVOC may cause a safety issue
 - ✓ VOC, SVOC, NVOC, Silicone Oil and some Metals may also be **Reactive** with reconstituted DP: also potential Performance & Quality Issue!
 - ✓ Also, Ions may need to be “checked off”...
 - ✓ Coated versus Non-Coated plungers



Pre-Filled Syringes

- **GLUE for staked needle:** Glue residues may for protein denaturation
- **TUNGSTEN Residues:** May cause protein aggregation
- **NEEDLE SHIELD:**
 - **No Direct Contact** between DP and Needle Shield
 - **HOWEVER:** Release of **Volatile (VOC)** and **Semi-Volatile (SVOC)** Compounds from the Needle shield into the content of the PFS is possible!
 - VOC and SVOC may also be **Reactive with DP:** also potential Performance & Quality Issue!
 - Typically **No NVOC, Metals and Ions** investigation is necessary for a Needle Shield.

2. Container/Closures for SVP's

Concern for

- Glass PFS
- Polymer PFS

Concern for – Glass PFS

Concern for – Glass PFS

Concern for – Glass PFS

Concern for - Polymer PFS

Concern for - Polymer PFS

(COATED) RUBBER

MONOMER
REMAINDERS &
POLYMER
FRAGMENTS

FILLERS: *Clay, Talc,
Carbonates...*

ANTIOXIDANTS:
*BHT, Irganox 1010,
Irgafos 128, ...*

CURING AGENTS:
*S,S-Donors, Phenol-
Formaldehyde...*

ACTIVATORS:

ZnO / Stearic Acid

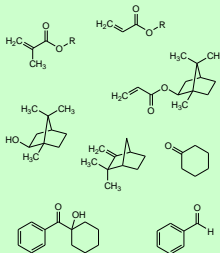
ACCELERATORS:

*Carbamates,
Sulfenamides...*

OTHERS: *Pigments,
Stabilizers, Release
agents...*

NEEDLE GLUE

- POLYMER
FRAGMENTS
- UV CURING-
ACTIVATORS



TUNGSTEN (W)

A Tungsten pin is
used in the
production of glass
pre-filled syringes
to keep the syringe
hub open
(cavity where the
staked needle is
glued in)

GLASS BARREL

Major: Silica (SiO₂)

Alkali (e.g. Na₂O)

Minor:

K (K₂O),

B (B₂O₃),

Ca (CaO),

Al (Al₂O₃)

Colored glass:
Fe₂O₃, TiO₂, CuO,
Mn³⁺

Sulfate (from
dealkalization)

Silicone oil
(provides lubricity)

COC/COP/PP BARREL

POLYMER
FRAGMENTS
SOLVENTS

ANTIOXIDANTS:
*BHT, Irganox 1010,
...*

ACID
SCAVENGERS:
Stearate,...

LUBRICANTS:
FA Esters, ...

WAXES

SLIP ADDITIVES:
*Erucamide,
Oleamide, ...*

PLASTICIZERS

RELEASE AGENTS

PIGMENTS

Optional: Silicone Oil

LABEL (ADHESIVE)

Permeable Plastic
Materials (e.g. PP, ...)

INK and ADHESIVE
constituents in a
complex composition,
but at low
concentrations

POLYMER
FRAGMENTS,
SOLVENTS,
PHOTO-INITIATORS,
STABILIZERS,
TACKIFIERS, ...

e.g. *Acrylates, PVA,
NR, Benzophenone,
Irgacure 184, Irgacure
651, Irganox 1010,
Dehydroabietic Acid,
DCHP, TBPP,
Siloxanes, ...*

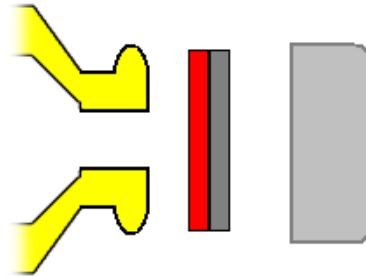
Potential Concern:
**SECONDARY
PACKAGING**

Piston / Needle
Shield / Tip Cap

3. Cartridges



2. Container/Closures for SVP's



Sealing Disk



➤ Sealing Disk:

- ✓ Typically, a sealing disk is a **two-layered** system
- ✓ The **inner layer** has **product contact** (primary contact), should be the focus of the investigation
- ✓ **“One Sided” extraction** mimics the product contact, avoids contribution of the outer layer
- ✓ **Complete Extraction** of the 2 layered sealing disk can be considered as **“Worst Case”**
- ✓ Both approaches can be taken and have found regulatory acceptance

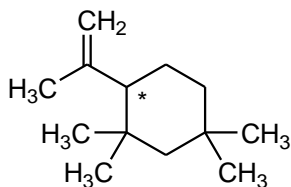


- **BARREL: *Metals*** (may not be necessary to be studied in EXT Study, if glass composition is available, direct assessment in LEA study)
Silicone Oil residues may cause protein aggregation
- **Cartridge Plunger (same as for PFS!):**
 - ✓ Typically, higher migration when solution is in contact (inverted)
 - ✓ Migration will be determined by:
 - Solubility of leachables in Reconstitution Solution (typically inorganic aqueous solution (*typically low solubility for most non-polar organic compounds*))
 - Potential Diffusion of Compounds through rubber, into solution
 - Temperature
 - ✓ VOC, SVOC and NVOC may cause a safety issue
 - ✓ VOC, SVOC, NVOC, Silicone Oil and some Metals may also be **Reactive** with reconstituted DP: also potential Performance & Quality Issue!
 - ✓ Also, Ions may need to be “checked off” ...

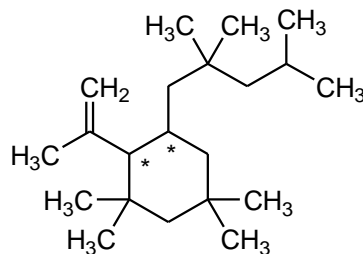
RUBBER OLIGOMERS: MAY NEED MORE ATTENTION

$C_{13}H_{24}$ and $C_{21}H_{40}$ Oligomers

- Considered as
 - Cyclic aliphatic hydrocarbon compounds
 - One double bond
- No experimental data / Literature data is known about **toxicity of these compounds**
- **Structure Activity Relationship Assessment (SAR)**: compound of low tox. risk.



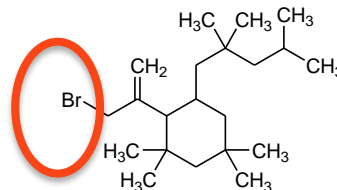
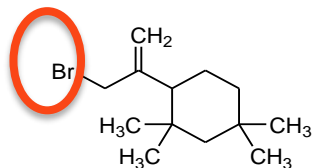
C13 oligomer



C21 oligomer

Halogenated Rubber Oligomers – Compounds of high concern

$C_{13}H_{23}Br$ / $C_{13}H_{23}Cl$ and $C_{21}H_{39}Br$ / $C_{21}H_{39}Cl$ Oligomers



- Considered as
 - **HALOGENATED** Cyclic Aliphatic Hydrocarbon compounds (Allyl Halide)
 - **Alkylating Agents**
 - One double bond
- Structure Activity Relationship (SAR) Assessment:

CARCINOGENICITY IN HUMANS IS PLAUSIBLE

- As no experimental data / Literature data is known about the toxicity of these compounds, a lot of Pharma companies:
 - Rely on the result of a SAR assessment to perform a tox evaluation
 - Conclude that these compounds are of High Concern

For potential Mutagenic/Carcinogenic compounds:

SCT: 0.15 µg/day (PQRI OINDP)

SCT: 1.5 µg/day (PQRI-PODP; ICH guideline on Genotoxic Impurities)

The low SCT/TTC levels for the Halogenated Oligomers mean:

- Low associated AET levels
- High level of method optimization to obtain these levels (certainly with LVP)
- e.g. SIM mode for GC/MS
- Can only be performed with appropriate analytical standards with known purity
 - *Method Selectivity*
 - *Accuracy*
 - *Sensitivity*
 - *Precision*
 - ...

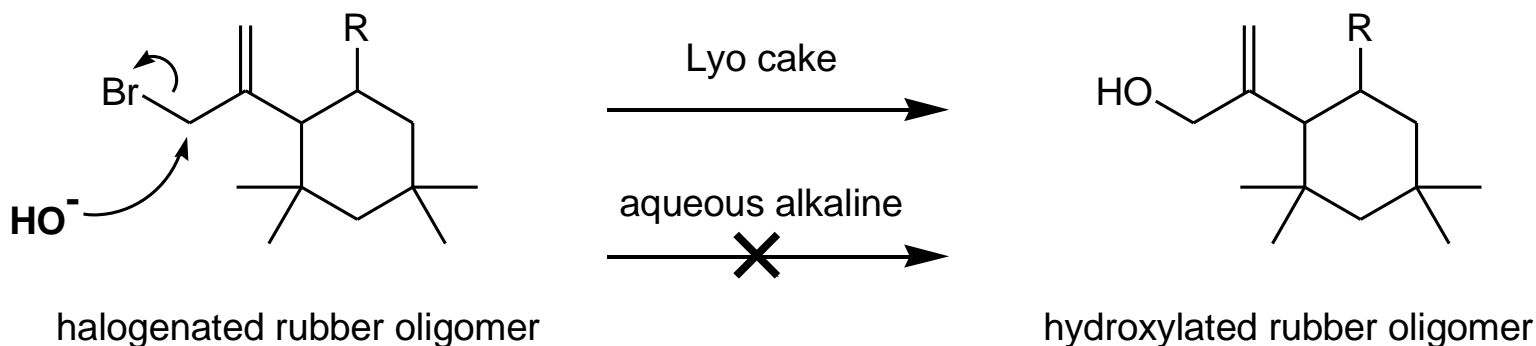
Halogenated Rubber Oligomers – Reactive Leachables

Cresol containing drug products, Bromocresol may be formed in the presence of Bromobutyl Stoppers (Mechanism is unknown)



Halogenated Rubber Oligomers – Reactive Leachables

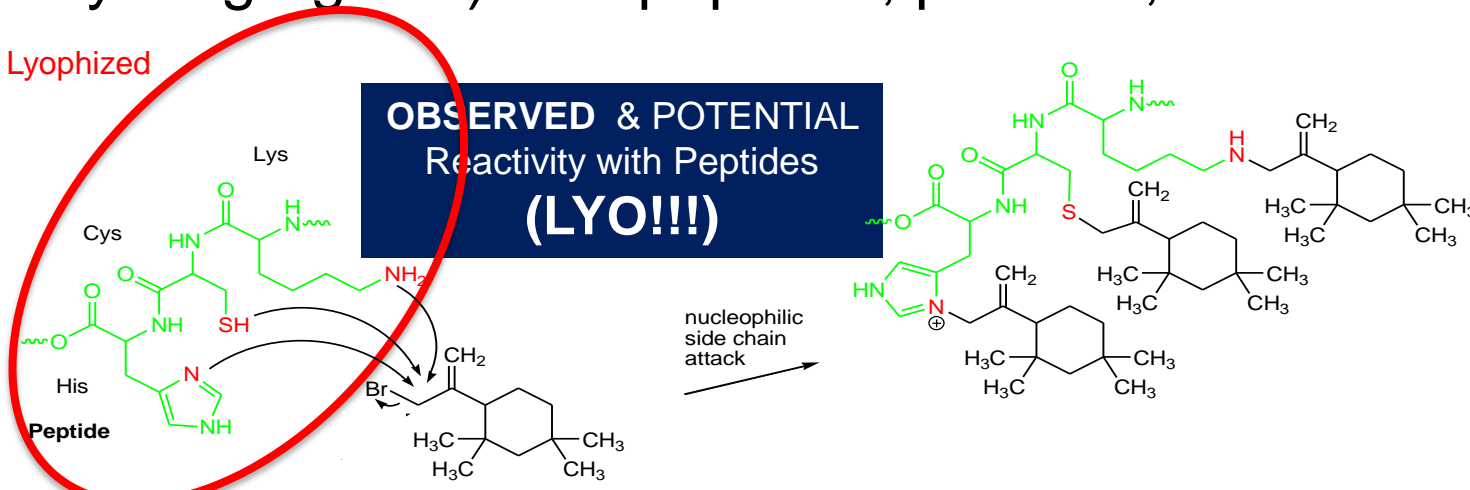
Formation of $C_{13}H_{23}OH$ out of $C_{13}H_{23}Br$ in Lyo Products



Observed Reactivity of C₁₃H₂₃Br and C₂₁H₃₉Br

(as alkylating agents) with peptides, proteins, and nucleic acids

Example of Lyophilized peptide



With different nucleophilic groups

Guidance for

Immunogenicity

Therapeutic Protein Products

Interactions between therapeutic protein products and the container closure may negatively affect product quality and immunogenicity. These interactions are more likely with prefilled syringes of therapeutic protein products. These syringes are composed of multiple surfaces and materials that interact with the therapeutic protein product over a prolonged time period and thus have the potential to alter product quality and immunogenicity. The following are other container closure considerations pertinent to immunogenicity:

U.S. Department of Health and Human Services
Food and Drug Administration
Center for Drug Evaluation and Research (CDER)
Center for Biologics Evaluation and Research (CBER)

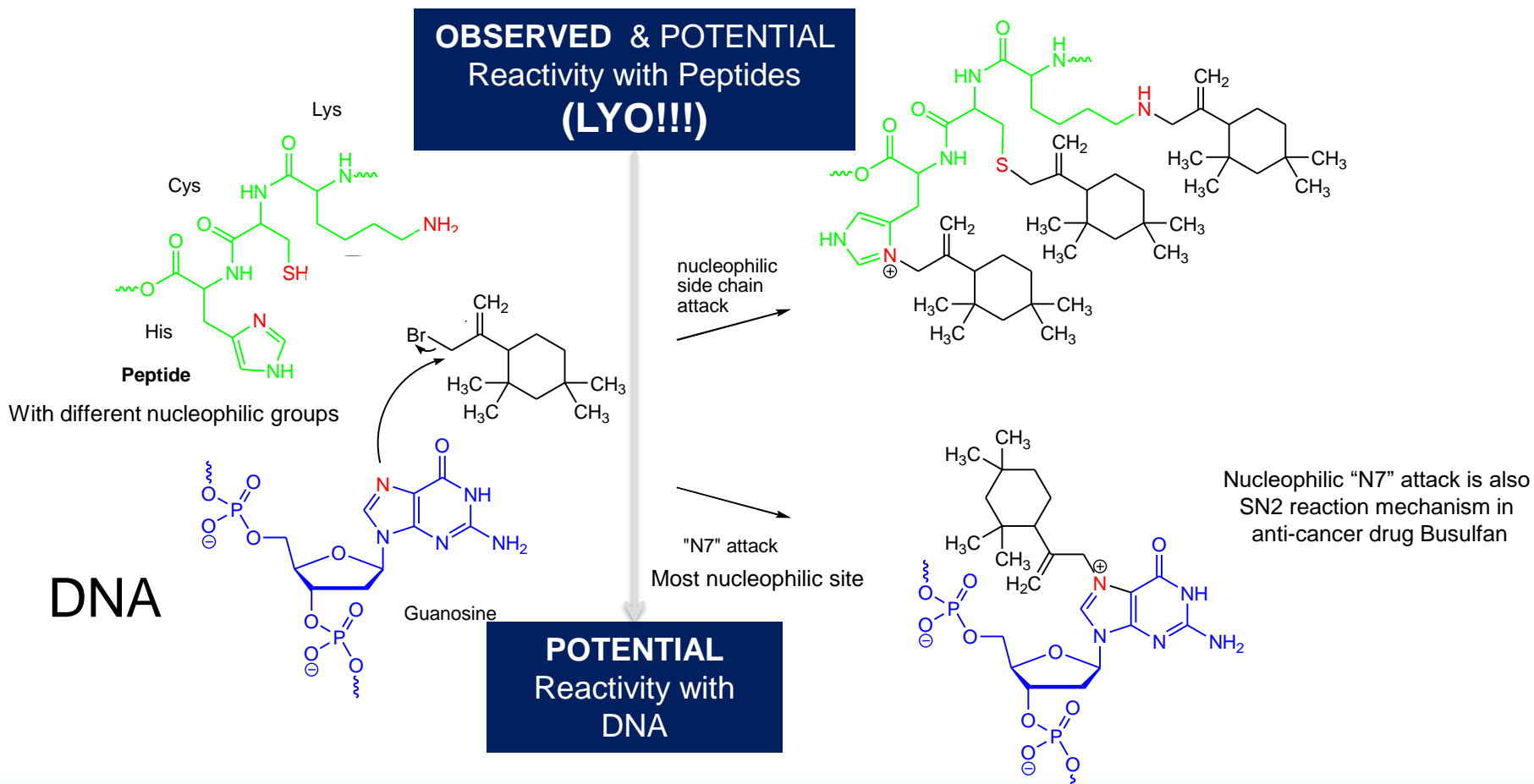
August 2014
Clinical/Medical

“... Interactions are more likely with prefilled syringes of therapeutic protein products...”

“... Materials that interact with the therapeutic protein product over a prolonged time and thus have the potential to alter product quality and immunogenicity...”

FDA Guidance for Industry, 2014

Observed Reactivity of $C_{13}H_{23}Br$ and $C_{21}H_{39}Br$
(as alkylating agents) with peptides, proteins, and nucleic acids





ANY QUESTIONS?