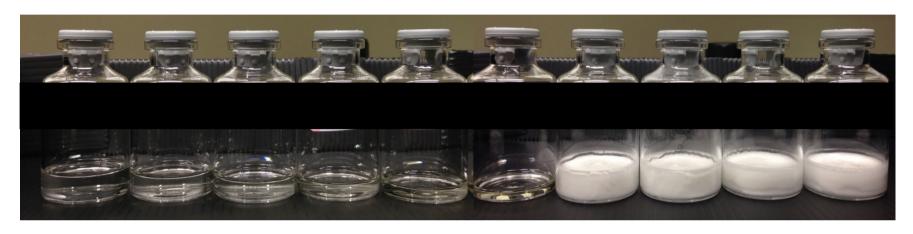


Progress of drying

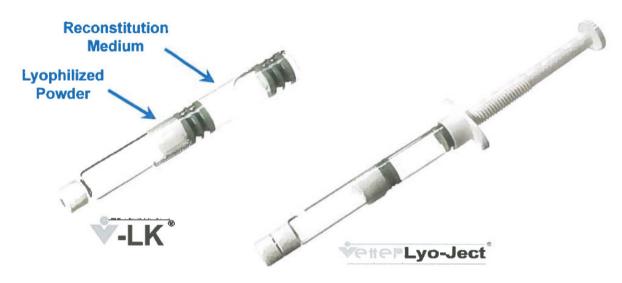






Primary packaging





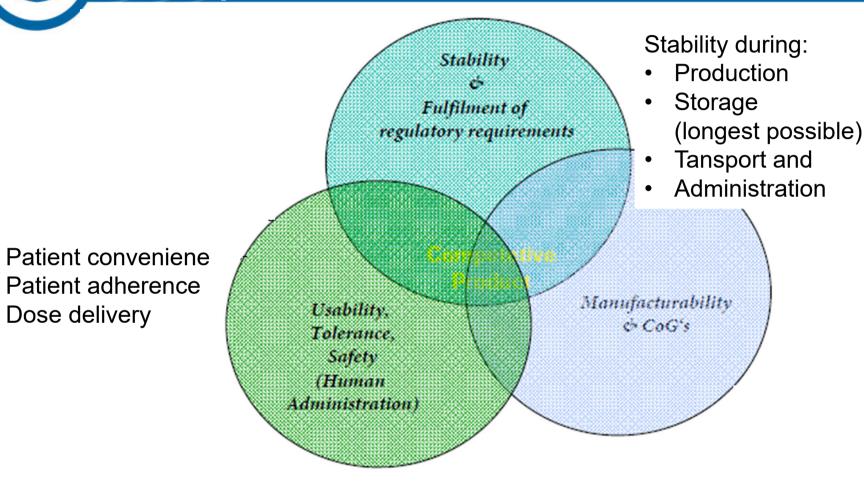
Vial (different coatings)

Cartridge

Syringe (Dual chamber syringe)



Requirements of a formulation



Caveat for proteins: Influence on undesirable adverse events and clinical efficiency, immunogenicity and pharmacokinetic profile through product specific degradation products.

Dose delivery



Design of a formulation

variable constant

Buffer system (His/HisHCI, Citrat, Acetat)

Phosphat

Stabilizers during freezing/ thawing (Sucrose, Trehalose)

- tonicity adjusting agents at the same time

Antioxidant (Methionine)

Preservatives (Multi-does-vials)

Benzylalcohol

Lyo/cryoprotectants and bulking agents (Sucrose, Mannitol, ...)

Liquid vs. Lyo IV vs. SC

light chain

heavy chain

Surfactants (Polysorbate 20, 80, Poloxamer)

Viscosity reduction e.g. Arginine-HCI

Connecting People, Science and Regulation®

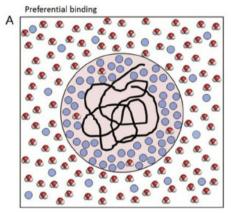


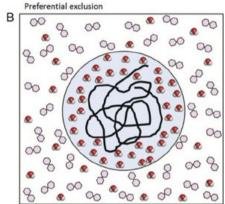
Lyo/cryo-protective excipients

Cryoprotectant

Stabilizes during the freezing process

- Excipients are preferentially excluded from the surface of the protein. This is an thermo-dynamically unfavored state. As the unfolded state of the protein would enhance this state, the protein is stabilized.
- (Timasheff 1993).

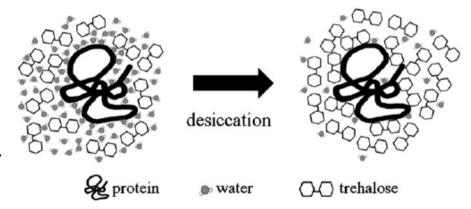




Lyoprotectant

Stabilizes during the drying process

 Water stablizes a protein in liquid solution by hydrogen bonding. The excipient replaces the hydrogen bonds of water during drying and thus stabilizes the protein.





Lyo/cryoprotective excipients

Crystalline excipients	Amorphous excipients
Ordered crystal structure	Glassy state
Eutectic temperature (defined melting point)	Glas transition temperature Characterization by differential Characterization ch
 Bulking agent High eutectic temperature : Elegant cake appearance Fast drying 	Stabilzation of e.g. proteinsAcceptable bulking agent at the same time
 In many cases no stabilization (e.g. for most proteins) Different morphologies dependent on excipient (Mannitol → Annealing) Glass breakage (Mannitol at high fill) 	 Low glass transition temperatures → Cake structure?
Glycin, Mannitol, NaCl,	Sucrose, Trehalose, PVP, Dextran,



Examples



Kadcyla 100 / 160mg

20 mg/mL ado-trastuzumab emtansine 10 mM sodium succinate pH 5.0 60 mM D-Sucrose 0.02% Polysorbate

Herceptin 150 / 400 mg

25 mg/mL Trastuzumab 5 mM L-Histidine/-HCl, pH 6.0 60 mM D-Trehalose 0.01 % Polysorbat 20





Analytical characterization

Product attributes for designing lyophilization cycles

- Differential scanning calorimetry: T_g, T_g, T_{eut}
- Freeze drying microscopy: T_{collapse}

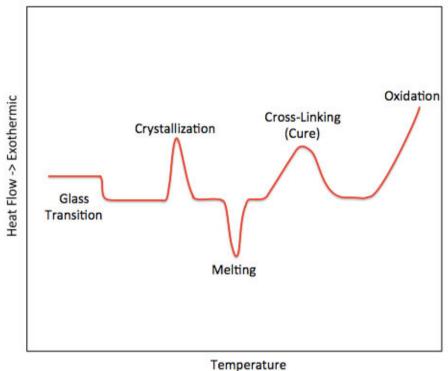
Solid state characterization after lyophilization

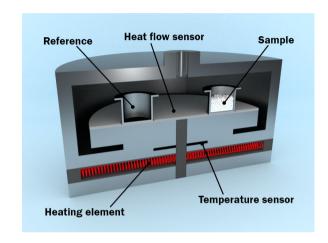
- Residual moisture (Karl Fischer, NIR)
- Reconstitution time
- Thermodynamic state (Xray powder diffraction)
- Specific surface area (BET)
- Cake appearance at different levels (visual inspection, 3D scanning, PDMS embedding, SEM, µCT)

Other quality attributes of active compound



Differential Scanning Calorimetry (e.g. T_{a'})

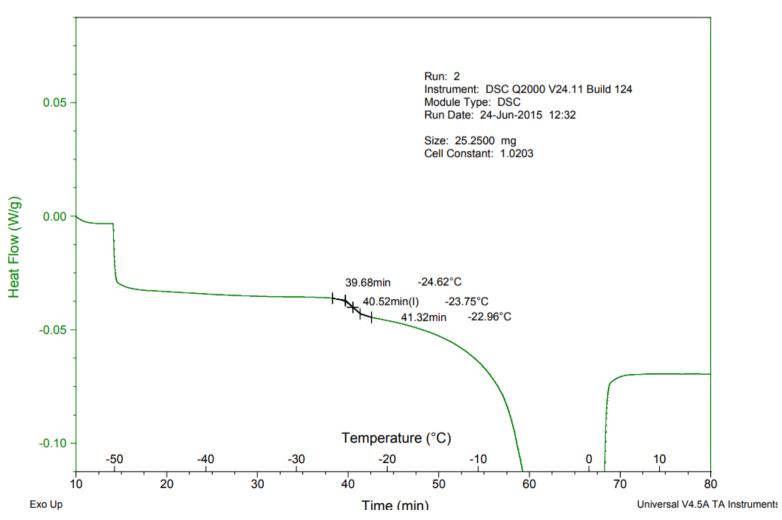




- Thermal analysis to detect physical transformation such as phase transitions (e.g. glass transition temperature T_{a'}/T_a, crystallization/melting point T_{eut} ...)
- Measurement of the difference in the amount of heat required to increase the temperature of a sample compared to a reference with well-defined heat capacity as a function of temperature
- Both the sample and reference are maintained at nearly the same temperature throughout the experiment



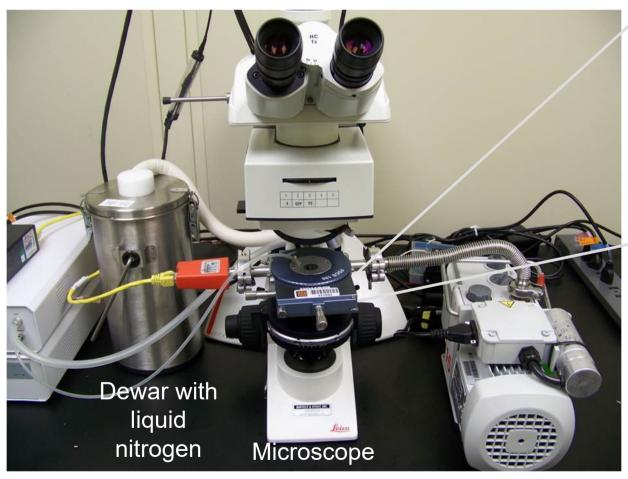
Differential Scanning Calorimetry (e.g. T_{g'})



Connecting People, Science and Regulation®



Freeze drying microscopy (T_{collapse})





Cryostage

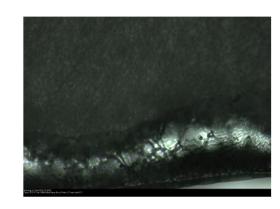
Vacuum pump



Freeze drying microscopy (T_{collapse})







(Intact) frozen sample

Onset of collapse

Complete collapse

$$\rightarrow$$
 T_g, $<$ T_{collapse} !!



Residual moisture – Water content



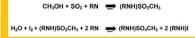
Gravimetric analysis

- Loss of mass in drying
- Targets any volatile components
- Destructive
- Does not account for 'hidden' water

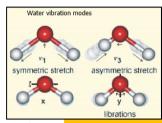


Karl-Fischer

 Quantitative water determination by titration



- Destructive
- Volumetric versus coulometric
- Extraction versus direct measurement



NIR spectroscopy

- Fingerprinting of molecule vibrations by near infrared
- Non-destructive
- High throughput (can be automated)
- Model generation and multivariate calibration techniques needed (e.g. principal components and partial least square analysis)



Karl-Fischer Titration

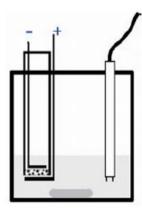


- Two media are needed: Titrating agent and working medium consisting of the three components sulfur dioxide, alcohol, and organic base or/and water free vehicle.
- End-point detection occurs either by color change or potentiometrically via an indicator electrode (free l₂/l- redox couple).

Volumetric Karl Fischer Titration

lodine is added by a burette during titration.
Suitable for samples where water is present as a major component: 100 ppm - 100%





$$CH_3OH + SO_2 + RN \implies (RNH)SO_3CH_3$$

$$H_2O + I_2 + (RNH)SO_3CH_3 + 2RN \implies (RNH)SO_4CH_3 + 2(RNH)I$$

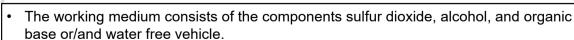
Redox reaction

Coulometric Karl Fischer Analysis

lodine is generated electrochemically during titration.

Suitable for samples where water is present in trace amounts:

1 ppm - 5%



• Two electrodes are needed: One for lodine generation (anode), and one for potentiometric end-point detection via the indicator electrode (free l₂/I- redox couple).



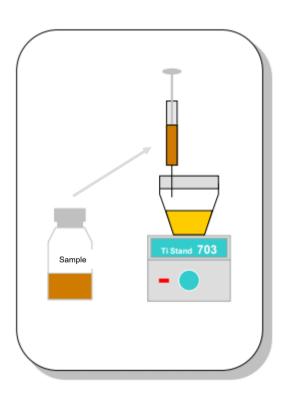


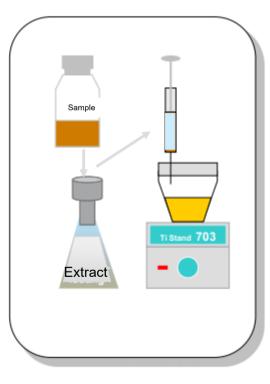
Karl-Fischer Titration

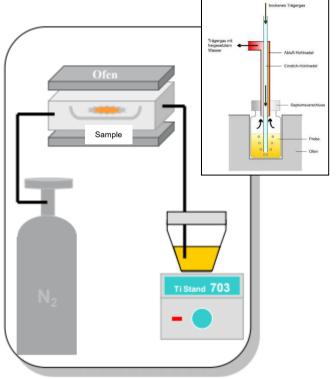
Direct Titration

Liquid Extraction

Evaporation





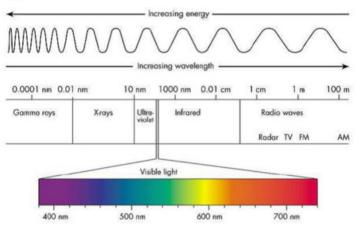


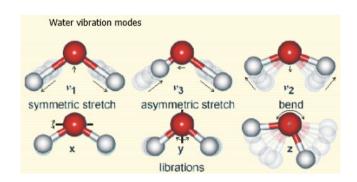
Highly dependent on the sample and its heat sensitivity.

Connecting People, Science and Regulation®

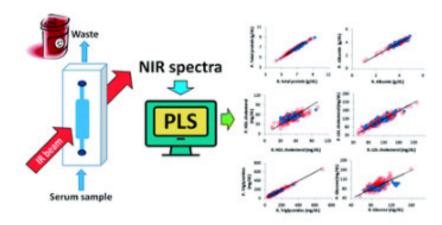


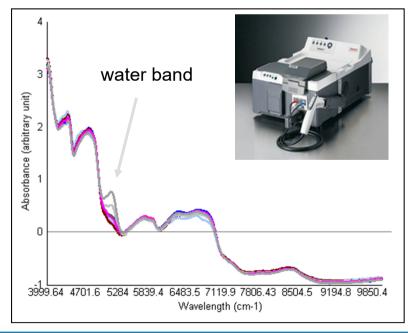
Residual moisture - NIR





- Molecule vibrations (overtone and combinations)
- Near infrared: ~760–2500 nm or 13.000–4.000 cm-1







Analytical characterization

Product attributes for designing lyophilization cycles

- Differential scanning calorimetry: T_g, T_g, T_{eut}
- Freeze drying microscopy: T_{collapse}

Solid state characterization after lyophilization

- Residual moisture (Karl Fischer, NIR)
- Reconstitution time
- Thermodynamic state (Xray powder diffraction)
- Specific surface area (BET)
- Cake appearance at different levels (visual inspection, 3D scanning, PDMS embedding, SEM, μCT)

Other quality attributes of active compound



Reconstitution time



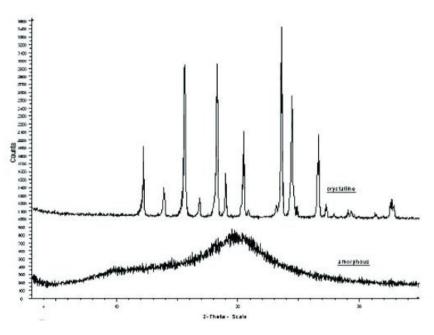




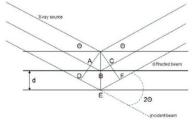
- → Water ideally flows along the side wall
- → Avoid foaming if samples contain surfactants
- → In case of long reconstitution times, shaking systems may be considered



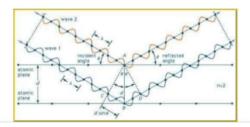
Xray powder diffraction - Morphology



The constructive and destructive interference can be measured as different intensities in the X-ray beam at given angles.



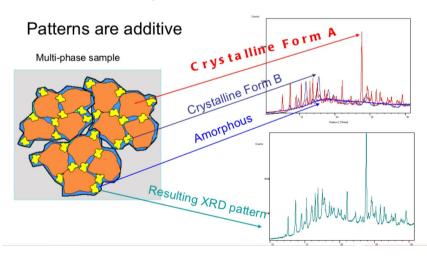
- A crystalline powder contains many small crystallites, ideally randomly oriented
- Diffraction occurs when crystallites are oriented such that specific atomic planes are in the correct relationship with the incoming x-rays



Bragg's law: nλ=2dsinθ

Constructive interference is detected when the path-length difference is equal to an integer number of wavelengths

Mixture analysis





Specific surface area (BET)

S.Brunauer, P.Emmett, E.Teller Adsorption of Gases in Multimolecular Layers, J. Am. Chem. Soc., 1938, 60 (2), pp 309–319





- Physical adsorption of a gas on the surface of the solid.
- Physical adsorption results from relatively weak forces (van der Waals forces)
 between the adsorbed gas molecules and the adsorbent surface area of the test
 powder. Thus, the determination is usually carried out at the temperature of liquid N2.
- Traditionally nitrogen is used as adsorbate gas.
- Based on the BET theory, the amount of adsorbed gas corresponds to a monomolecular layer on the surface.
- The amount of adsorbed gas is correlated to the total surface area of the particles including pores.









Sample preparation: degasing under vacuum and elevated temperature followed by measurement in liquid N2.



Visual inspection

Patel et al: Lyophilized Drug Product Cake Appearance: What Is Acceptable?
Patel S, Nail S, Pikal M, Geidobler R, Winter G, Hawe A, Davagnino J, Rambhatla Gupta S.
J Pharm Sci. 2017 Jul;106(7):1706-1721. doi: 10.1016/j.xphs.2017.03.014.

Cosmetic defects versus impact on product quality?



Intact cake



light collapse/melt-back



severe collapse/melt-back



complete collapse/melt-back



crack



dents



splashing



fogging

Connecting People, Science and Regulation®



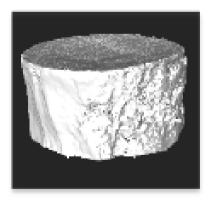
3D scanning



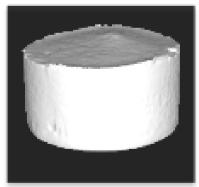
Dex0/Suc100

Dex60/Suc40

Dex100/Suc0









PDMS embedding



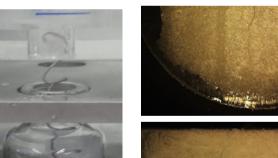


Philippe Lam and Thomas W. Patapoff

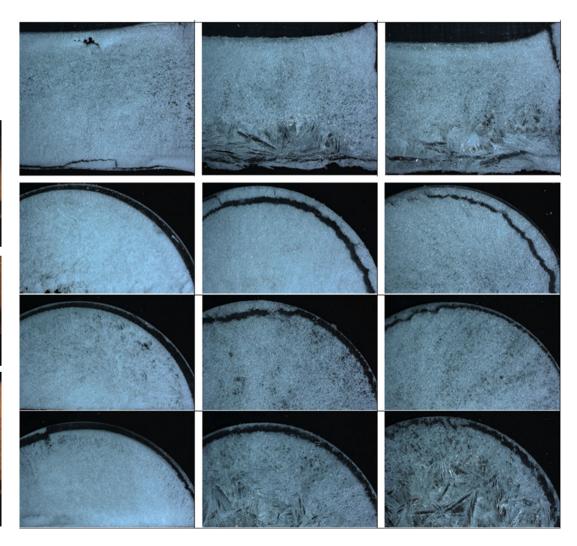
PDA J Pharm Sci and Tech 2011, 65 425-430



PDA



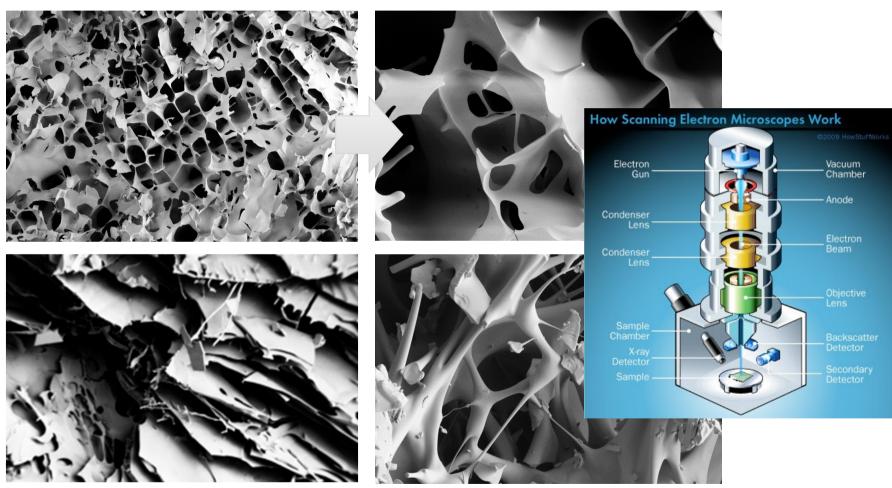




Connecting People, Science and Regulation®

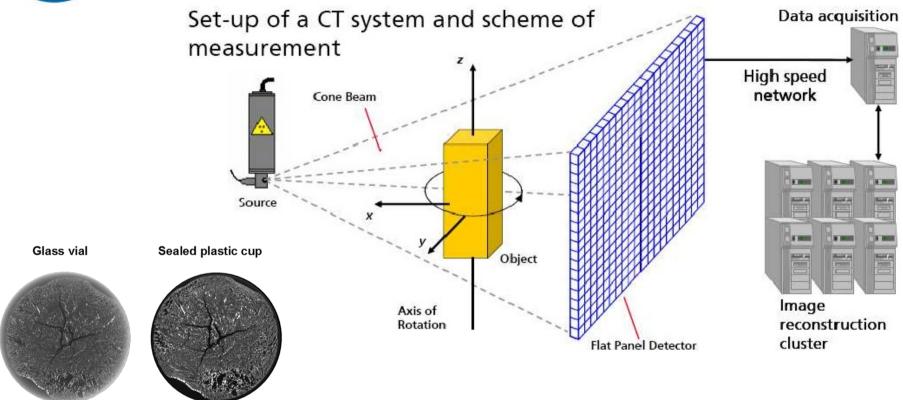


Scanning electron microscopy (SEM)





Micro-computated tomography (μCT)



- A micro-focus x-ray source illuminates the object and a planar x-ray detector collects magnified projection images.
- Based on hundreds of angular views acquired while the object rotates, a computer synthesizes a stack of virtual cross section slices through the object.
- · You can then scroll through the cross sections, interpolating sections along different planes, to inspect the internal structure.
- Selecting simple or complex volumes of interest, you can measure 3D morphometric parameters and create realistic visual models.



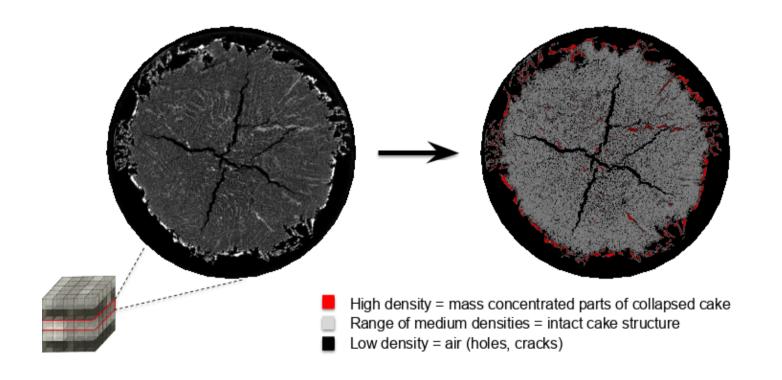
Micro-computated tomography (μCT)

Global cake characterization

 μ -CT - Interpretation of reconstructed volume

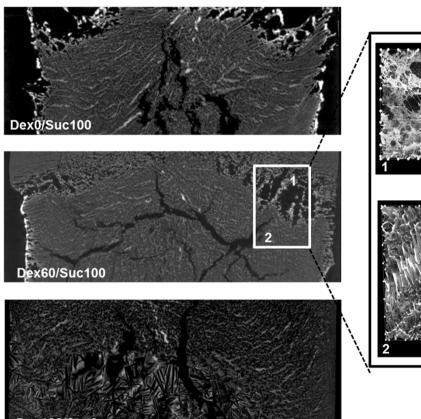
Reconstructed μ -CT slice

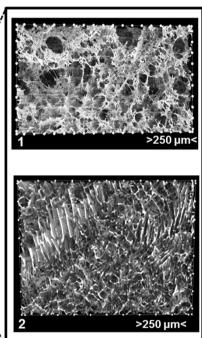
Histogram based coloration



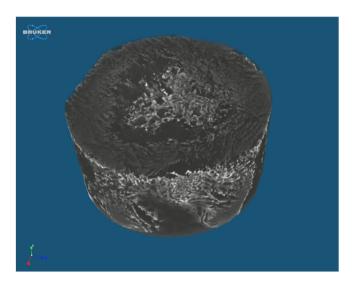


Micro-computated tomography (μCT)









Pros and cons and applicability of different imaging techniques summarized in Häuser et al: Imaging techniques to characterize cake appearance of freeze-dried Products. J Pharm Sci. 2018.