# Theory 2

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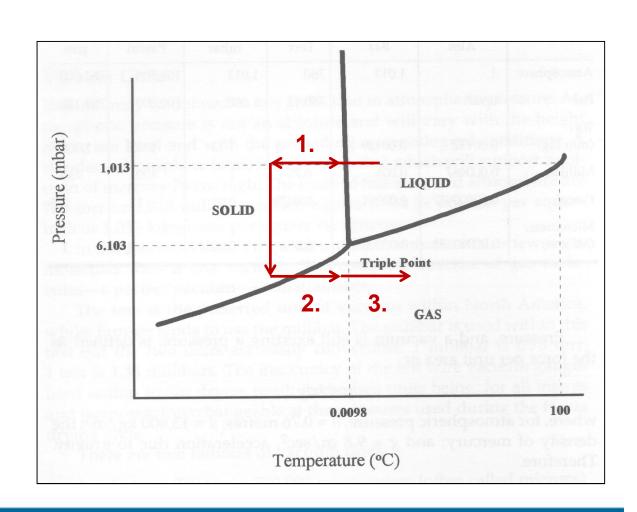


- Basic principles of freeze drying processes
  - Physical understanding
  - Critical process parameters
- Primary packaging components
- Development and composition of a (biological) formulation
- Analytical characterization:
  - Product attributes for designing lyophilization cycles
  - Solid state characterization after lyophilization



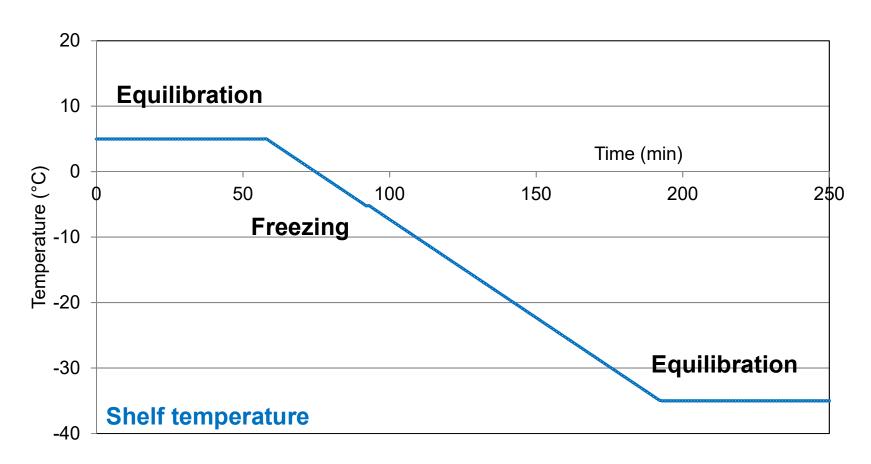
# Basic principles

- Drying by sublimation of ice (as well as desorption)
- Phases:
  - 1. Freezing phase
    - ca. 2-5 h
  - 2. Primary drying
    - ca. 5 h 5 d
  - 3. Sekundärtrocknung
    - bis 10 h



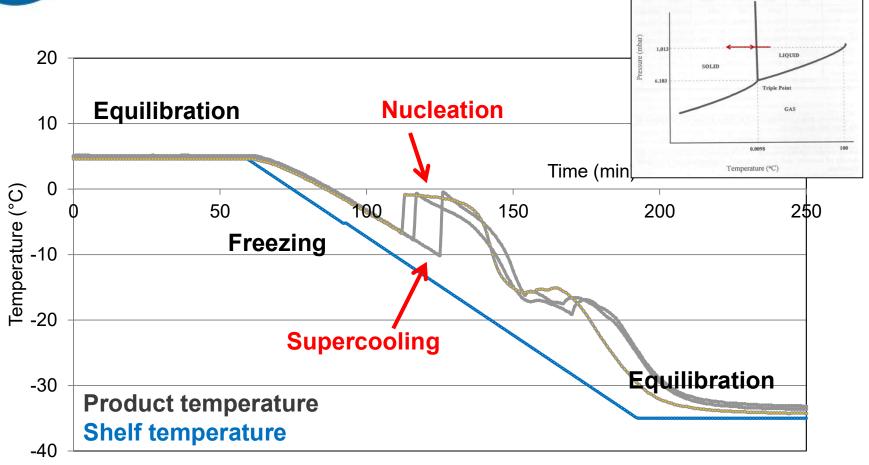


# Freezing



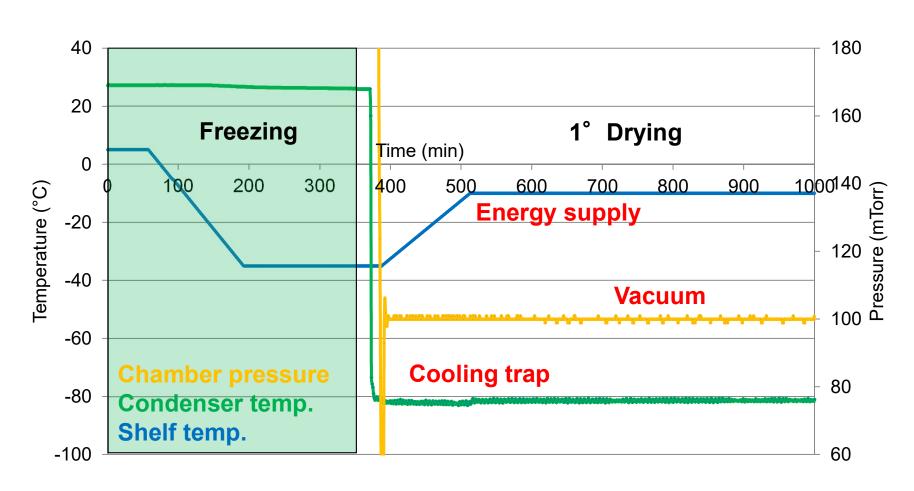


# Freezing - Nucleation



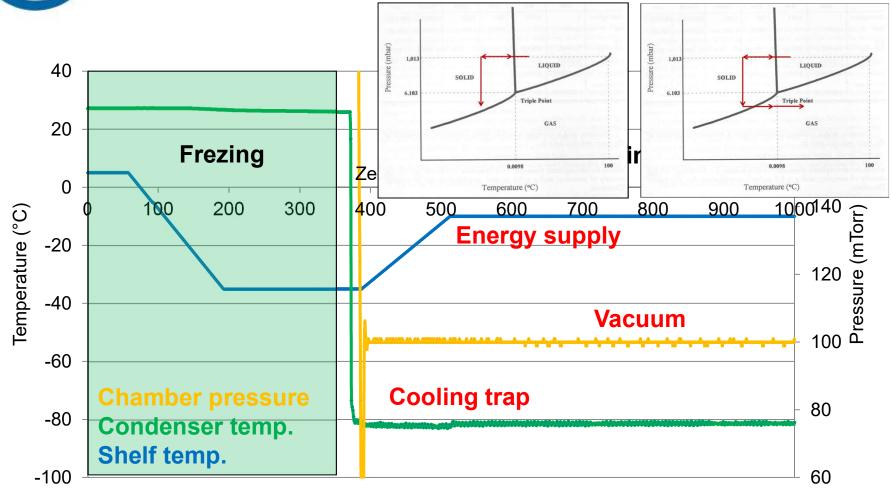


# Primary Drying



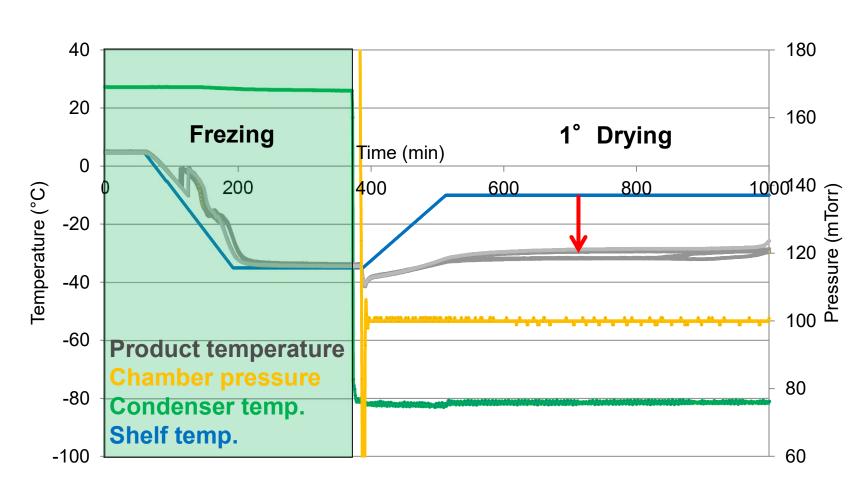


# **Primary Drying**



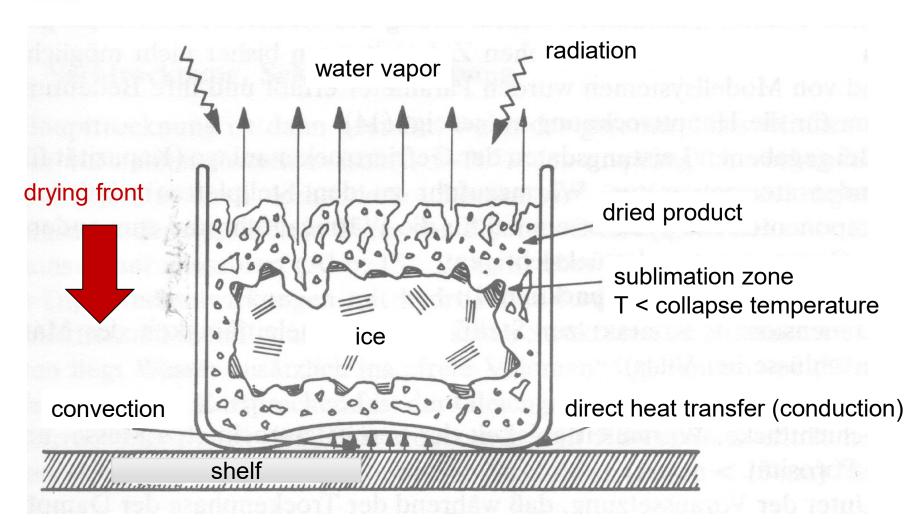


# Primary Drying - Sublimation



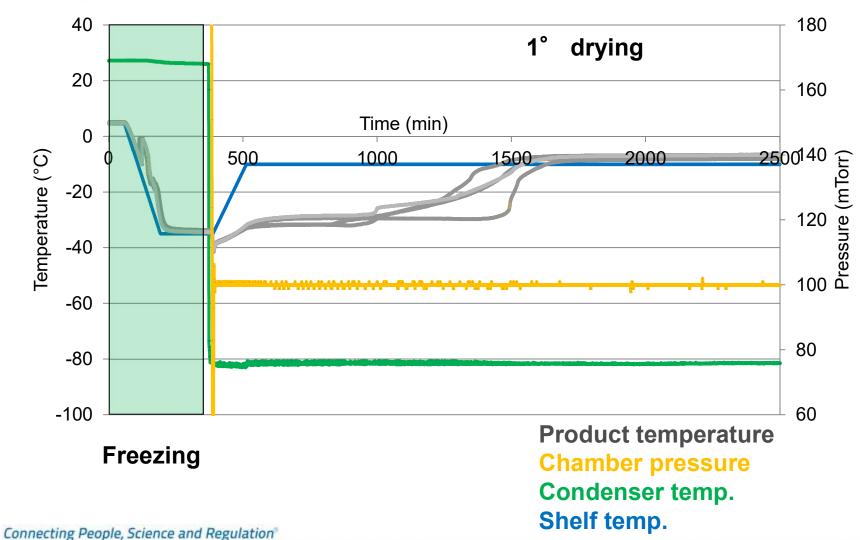


# Primary Drying - Sublimation



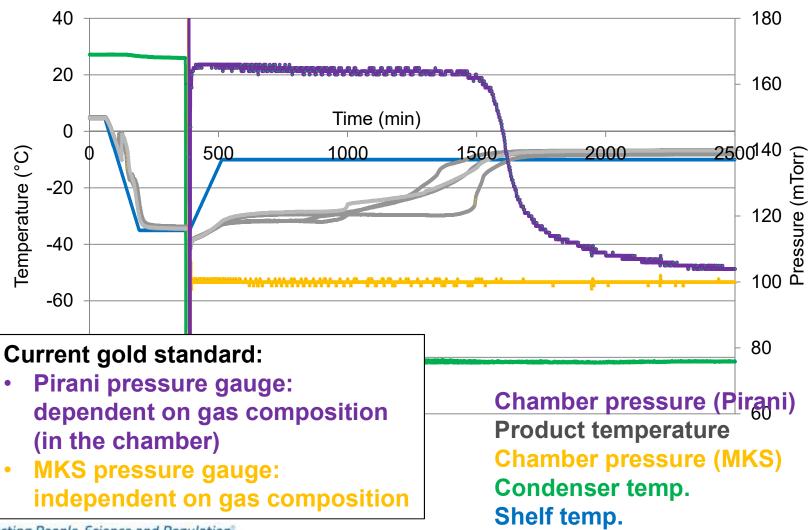


## End of primary drying: Product temperature





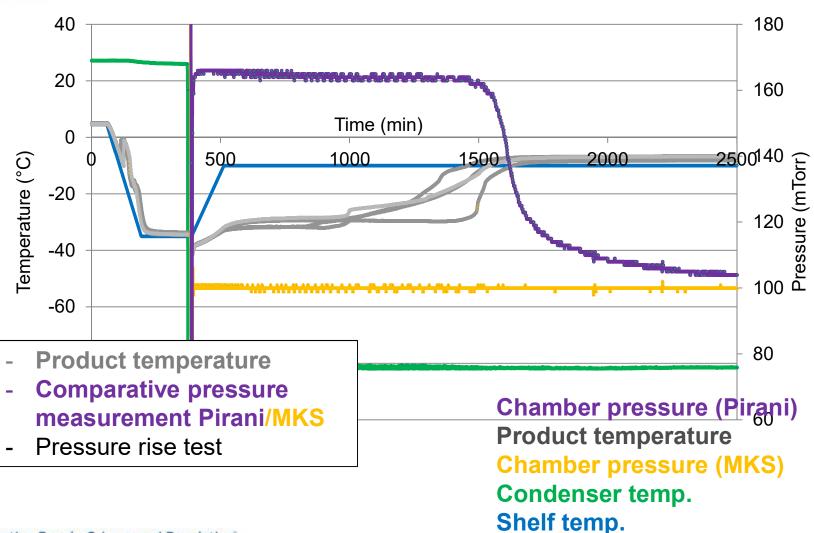
### End of primary drying: Pressure gauges



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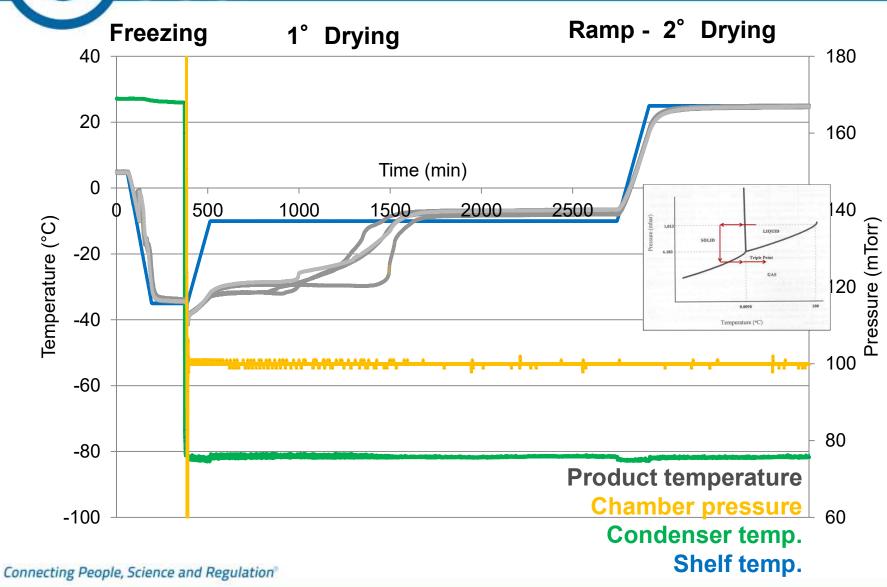


## End of primary drying - Options

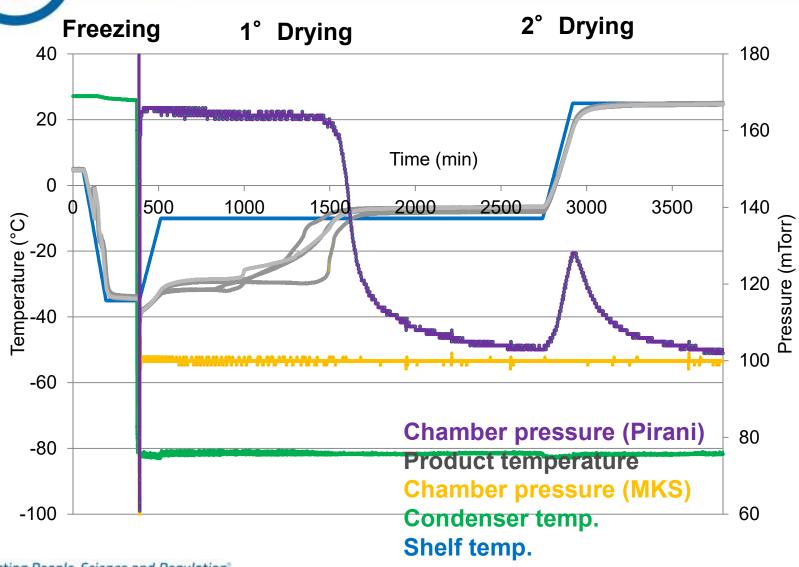


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# Secondary drying - Desorption

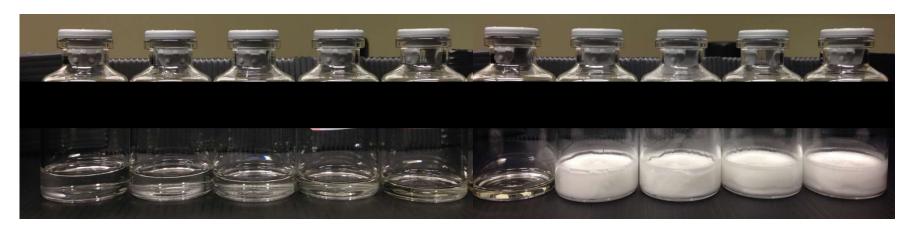


# Secondary drying - Desorption





# Progress of drying







# Primary packaging





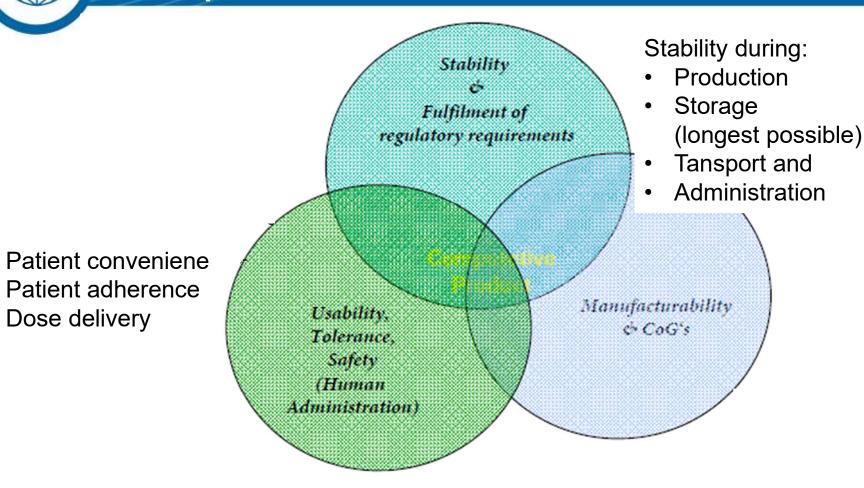
Vial (different coatings)

Cartridge

Syringe (Dual chamber syringe)



# Requirements of a formulation



Caveat for proteins: Influence on undesirable adverse events and clinical efficiency, immunogenicity and pharmacokinetic profile through product specific degradation products.

Dose delivery



# Design of a formulation

variable constant

**Buffer** system

(His/HisHCI, Citrat, Acetat)

**Phosphat** 

Stabilizers during freezing/ thawing (Sucrose, Trehalose)

- tonicity adjusting agents at the same time

**Antioxidant** 

(Methionine)

**Preservatives** (Multi-does-vials)

Benzylalcohol

Lyo/cryoprotectants and bulking agents (Sucrose, Mannitol, ...)

Liquid vs. Lyo IV vs. SC

light chain

heavy chain

**Surfactants** (Polysorbate 20, 80, Poloxamer)

**Viscosity** reduction e.g. **Arginine-HCI** 

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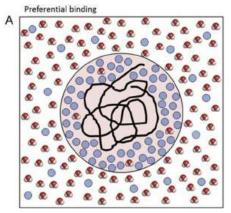


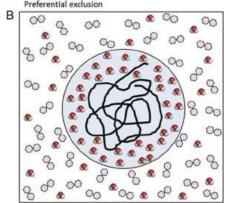
# Lyo/cryo-protective excipients

#### **Cryoprotectant**

#### **Stabilizes during the freezing process**

- Excipients are preferentially excluded from the surface of the protein. This is an thermo-dynamically unfavored state. As the unfolded state of the protein would enhance this state, the protein is stabilized.
- (Timasheff 1993).

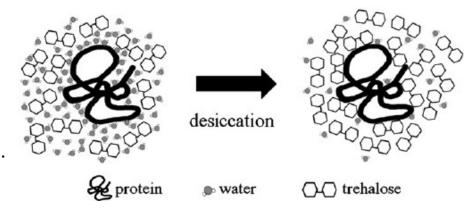




#### Lyoprotectant

#### **Stabilizes during the drying process**

 Water stablizes a protein in liquid solution by hydrogen bonding. The excipient replaces the hydrogen bonds of water during drying and thus stabilizes the protein.





# Lyo/cryoprotective excipients

Crystalline excipients	Amorphous excipients
Ordered crystal structure	Glassy state
Eutectic temperature (defined melting point)	Glas transition temperature  Characterization by differential Characterization by differential calorimetry
<ul> <li>Bulking agent</li> <li>High eutectic temperature :</li> <li>Elegant cake appearance</li> <li>Fast drying</li> </ul>	<ul><li>Stabilzation of e.g. proteins</li><li>Acceptable bulking agent at the same time</li></ul>
<ul> <li>In many cases no stabilization (e.g. for most proteins)</li> <li>Different morphologies dependent on excipient (Mannitol → Annealing)</li> <li>Glass breakage (Mannitol at high fill)</li> </ul>	<ul> <li>Low glass transition temperatures</li> <li>→ Cake structure?</li> </ul>
Glycin, Mannitol, NaCl,	Sucrose, Trehalose, PVP, Dextran,



# Examples



#### Kadcyla 100 / 160mg

20 mg/mL ado-trastuzumab emtansine 10 mM sodium succinate pH 5.0 60 mM D-Sucrose 0.02% Polysorbate

#### Herceptin 150 / 400 mg

25 mg/mL Trastuzumab 5 mM L-Histidine/-HCl, pH 6.0 60 mM D-Trehalose 0.01 % Polysorbat 20





## Analytical characterization

#### Product attributes for designing lyophilization cycles

- Differential scanning calorimetry: T<sub>g</sub>, T<sub>g</sub>, T<sub>eut</sub>
- Freeze drying microscopy: T<sub>collapse</sub>

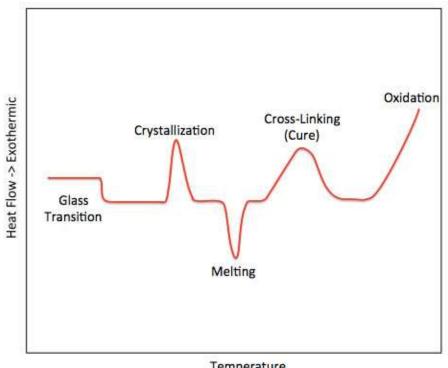
#### Solid state characterization after lyophilization

- Residual moisture (Karl Fischer, NIR)
- Reconstitution time
- Thermodynamic state (Xray powder diffraction)
- Specific surface area (BET)
- Cake appearance at different levels (visual inspection, 3D scanning, PDMS embedding, SEM, μCT)

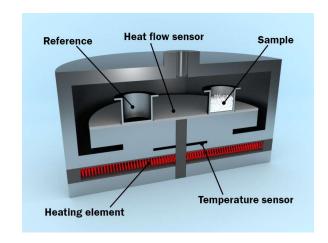
Other quality attributes of active compound



## Differential Scanning Calorimetry (e.g. T<sub>g'</sub>)



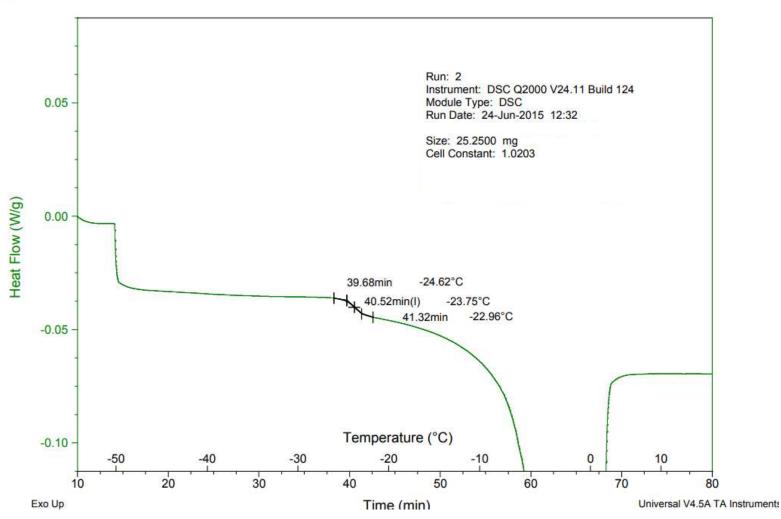
Temperature



- Thermal analysis to detect physical transformation such as phase transitions (e.g. glass transition temperature T<sub>g'</sub>/T<sub>g</sub>, crystallization/melting point T<sub>eut</sub> ...)
- Measurement of the difference in the amount of heat required to increase the temperature of a sample compared to a reference with well-defined heat capacity as a function of temperature
- Both the sample and reference are maintained at nearly the same temperature throughout the experiment

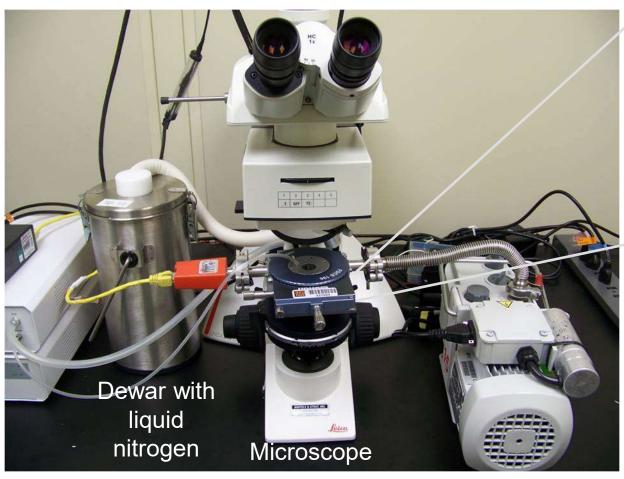


## Differential Scanning Calorimetry (e.g. $T_{g'}$ )





# Freeze drying microscopy (T<sub>collapse</sub>)



Cryostage

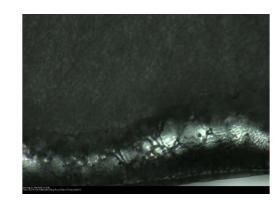
Vacuum pump



# Freeze drying microscopy (T<sub>collapse</sub>)







(Intact) frozen sample

Onset of collapse

Complete collapse

$$\rightarrow$$
 T<sub>g</sub>' < T<sub>collapse</sub> !!



## Residual moisture – Water content



# Gravimetric analysis

cabinet or IR

Destructive

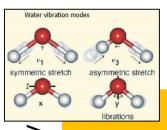


Quantitative water

CH<sub>3</sub>OH + SO<sub>2</sub> + RN 

→ (RNH)SO<sub>3</sub>CH<sub>3</sub>

- Destructive
- Extraction versus direct



N N

spectroscopy

- multivariate calibration



## Karl-Fischer Titration

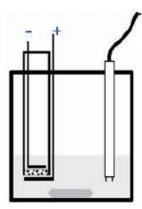


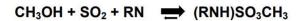
- Two media are needed: Titrating agent and working medium consisting of the three components sulfur dioxide, alcohol, and organic base or/and water free vehicle.
- End-point detection occurs either by color change or potentiometrically via an indicator electrode (free I<sub>2</sub>/I- redox couple).

#### Volumetric Karl Fischer Titration

lodine is added by a burette during titration.
Suitable for samples where water is present as a major component: 100 ppm - 100%







$$H_2O + I_2 + (RNH)SO_3CH_3 + 2 RN \implies (RNH)SO_4CH_3 + 2 (RNH)I$$

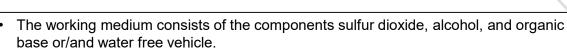
Redox reaction

#### **Coulometric Karl Fischer Analysis**

lodine is generated electrochemically during titration.

Suitable for samples where water is present in trace amounts:

1 ppm - 5%



Two electrodes are needed: One for Iodine generation (anode), and one for potentiometric end-point detection via the indicator electrode (free  $I_2/I$ - redox couple).



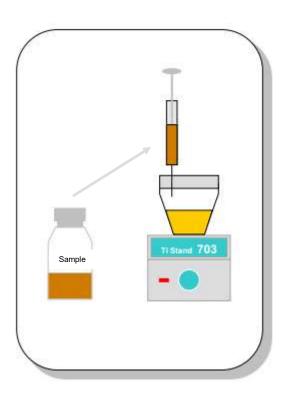


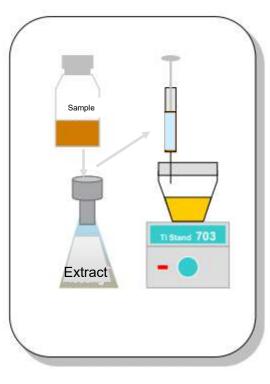
# Karl-Fischer Titration

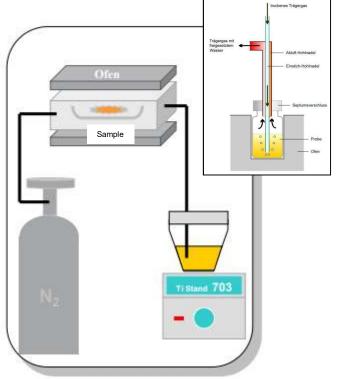
**Direct Titration** 

Liquid Extraction

Evaporation





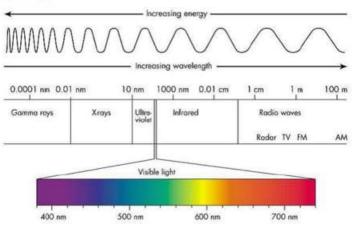


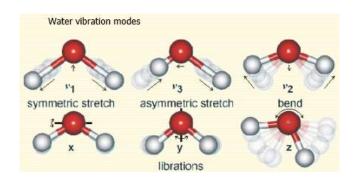
Highly dependent on the sample and its heat sensitivity.

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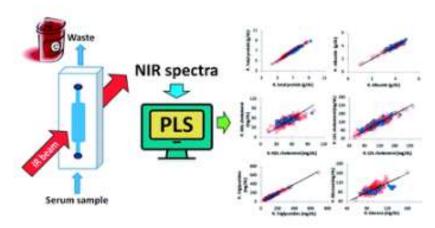


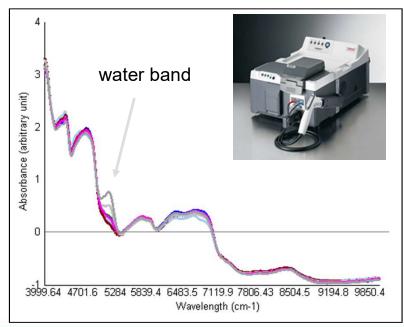
# Residual moisture - NIR





- Molecule vibrations (overtone and combinations)
- Near infrared: ~760–2500 nm or 13.000–4.000 cm-1







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#### Solid state characterization after lyophilization

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## Reconstitution time



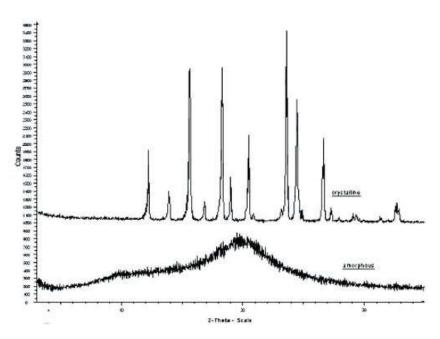




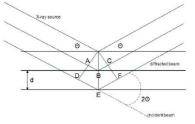
- → Water ideally flows along the side wall
- → Avoid foaming if samples contain surfactants
- → In case of long reconstitution times, shaking systems may be considered



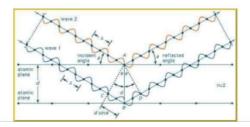
## Xray powder diffraction - Morphology



The constructive and destructive interference can be measured as different intensities in the X-ray beam at given angles.



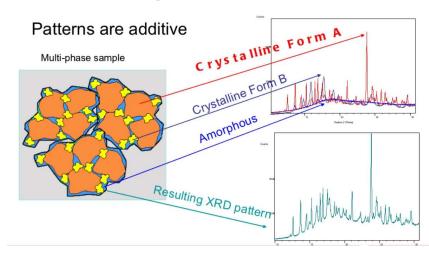
- A crystalline powder contains many small crystallites, ideally randomly oriented
- Diffraction occurs when crystallites are oriented such that specific atomic planes are in the correct relationship with the incoming x-rays



#### Bragg's law: nλ=2dsinθ

Constructive interference is detected when the path-length difference is equal to an integer number of wavelengths

#### Mixture analysis





# Specific surface area (BET)

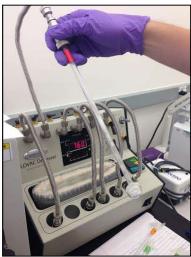
S.Brunauer, P.Emmett, E.Teller Adsorption of Gases in Multimolecular Layers, J. Am. Chem. Soc., 1938, 60 (2), pp 309–319



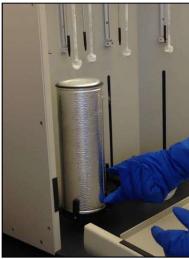


- Physical adsorption of a gas on the surface of the solid.
- Physical adsorption results from relatively weak forces (van der Waals forces)
   between the adsorbed gas molecules and the adsorbent surface area of the test
   powder. Thus, the determination is usually carried out at the temperature of liquid N2.
- Traditionally nitrogen is used as adsorbate gas.
- Based on the BET theory, the amount of adsorbed gas corresponds to a monomolecular layer on the surface.
- The amount of adsorbed gas is correlated to the total surface area of the particles including pores.









Sample preparation: degasing under vacuum and elevated temperature followed by measurement in liquid N2.



## Visual inspection

Patel et al: Lyophilized Drug Product Cake Appearance: What Is Acceptable?
Patel S, Nail S, Pikal M, Geidobler R, Winter G, Hawe A, Davagnino J, Rambhatla Gupta S.
J Pharm Sci. 2017 Jul;106(7):1706-1721. doi: 10.1016/j.xphs.2017.03.014.

Cosmetic defects versus impact on product quality?



Intact cake



light collapse/melt-back



severe collapse/melt-back



complete collapse/melt-back



crack



dents



splashing



fogging

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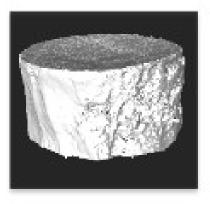
# 3D scanning



Dex0/Suc100

Dex60/Suc40

Dex100/Suc0









## PDMS embedding

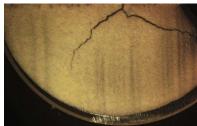
PDA Journal of Pharmaceutical Science and Technology



#### An Improved Method for Visualizing the Morphology of Lyophilized Product Cakes

Philippe Lam and Thomas W. Patapoff

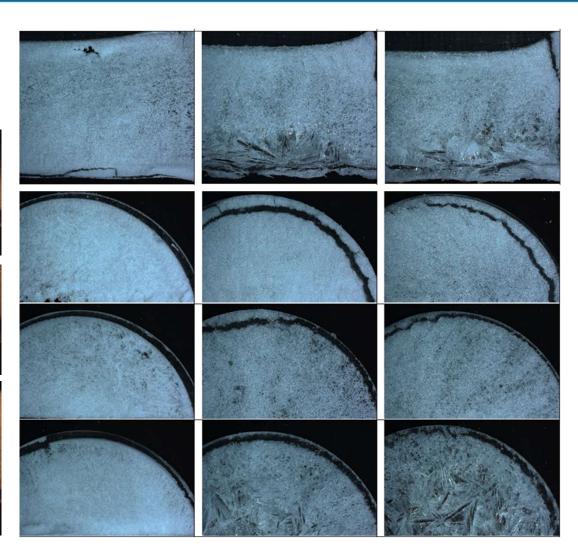
PDA J Pharm Sci and Tech 2011, 65 425-430





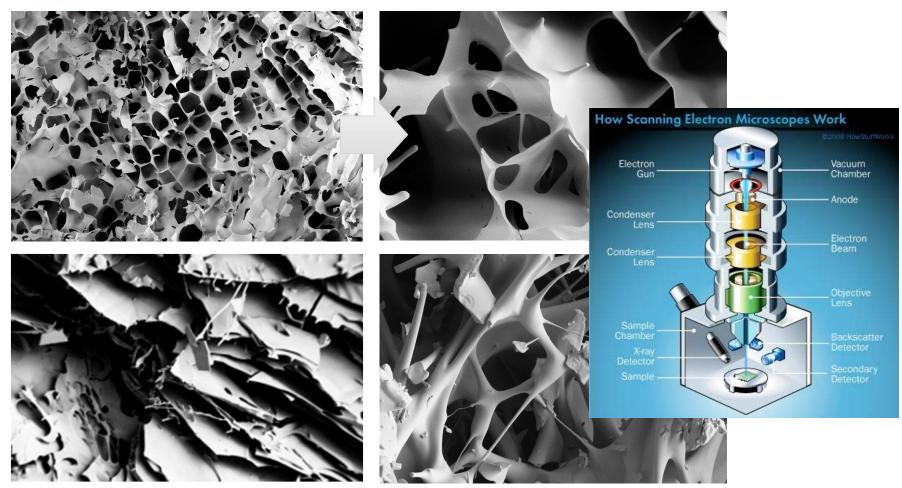






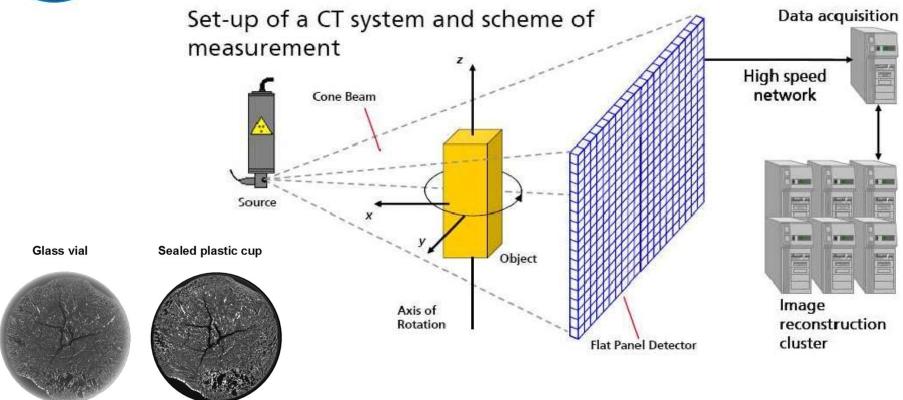


## Scanning electron microscopy (SEM)





## Micro-computated tomography (μCT)



- A micro-focus x-ray source illuminates the object and a planar x-ray detector collects magnified projection images.
- Based on hundreds of angular views acquired while the object rotates, a computer synthesizes a stack of virtual cross section slices through the object.
- · You can then scroll through the cross sections, interpolating sections along different planes, to inspect the internal structure.
- Selecting simple or complex volumes of interest, you can measure 3D morphometric parameters and create realistic visual models.



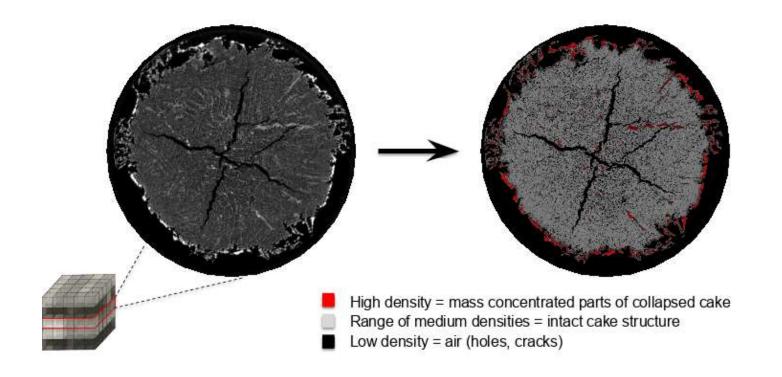
## Micro-computated tomography (µCT)

#### Global cake characterization

 $\mu$ -CT - Interpretation of reconstructed volume

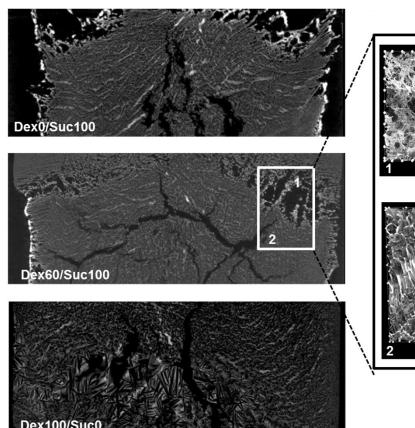
Reconstructed  $\mu$ -CT slice

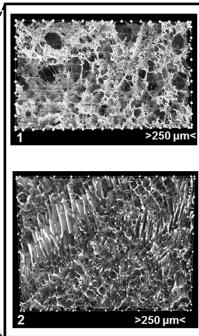
Histogram based coloration

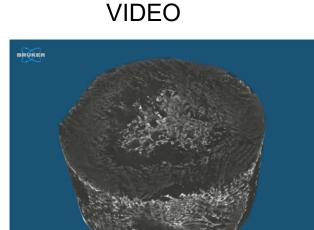




## Micro-computated tomography (μCT)







Pros and cons and applicability of different imaging techniques summarized in Häuser et al: Imaging techniques to characterize cake appearance of freeze-dried Products. J Pharm Sci. 2018.

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