

Supplementary material

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Freeze-drying in practice

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How to determine the reconstitution volume?

Note: Reconstitution volume \neq Filling volume (solid content needs to be considered)

Two practical approaches:

1. Measuring the loss on drying during freeze-drying
 - Weigh selected filled and semi-stoppered vials before and after freeze-drying
 - Mass difference can be accounted to water loss on drying
 - Mass loss can be converted to the reconstitution volume by division by density of water
2. Calculate the total amount of water that could be lost on drying
 - Determine the density of your formulated solution to be freeze-dried
 - Calculate the exact total solid content based on composition
 - Calculate the theoretically filled mass (by multiplication with formulation density)
 - Subtract the total solid content from the theoretically filled mass
 - Calculated mass difference can be converted to the reconstitution volume by division by density of water

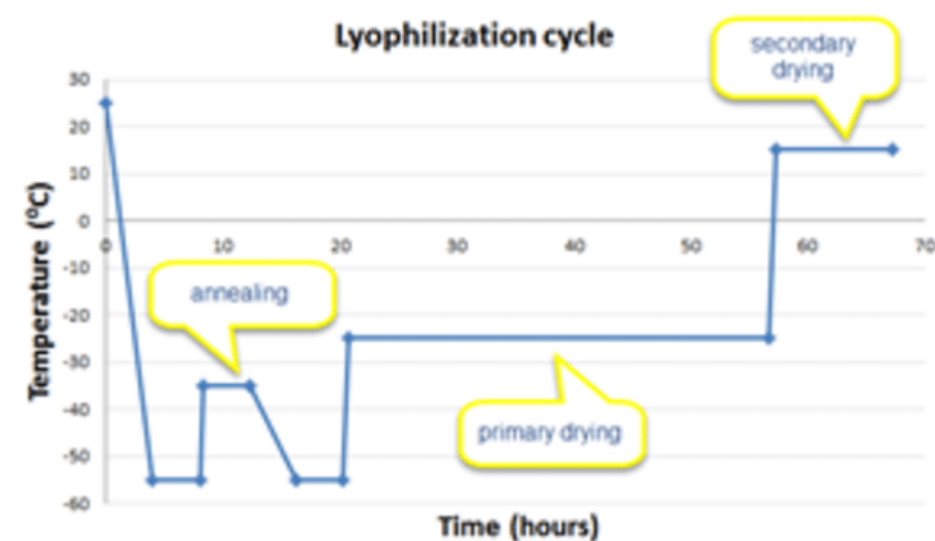


Freezing – Annealing/Thermal treatment

Annealing = hold step at $T_s > T_g'$ to allow for (complete) crystallization of potentially crystalline components

- Mainly used in formulations with crystalline bulking agents (e.g., Mannitol or Glycine)
- Allows for crystallization of potentially crystalline excipients in the freezing step and prevents crystallization during (primary) drying and has been shown to increase chemical stability
- Only partial crystallization of potentially crystalline excipients may impair product stability after lyo
- Literature recommendation (Tang, Pikal, Pharm. Res., 2004):
 - Apply regular freezing procedure
 - Allow for complete solidification by hold times of 1-2h
 - Bring product temperature to 10 °C – 20 °C above T_g' , but well below T_{eu}
 - Allow for complete solidification afterwards again before starting with primary drying
 - Example annealing step for Mannitol/Glycine: $T_s = -20$ °C for ≥ 2 h

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Annealing in amorphous formulations:

Luthra SA, Hodge IM, Pikal MJ. Investigation of the Impact of Annealing on Global Molecular Mobility in Glasses: Optimization for Stabilization of Amorphous Pharmaceuticals. J Pharm Sci. 2008;97(9):3865–82.

T. Kharatyan et al. Quantitative Analysis of Glassy Relaxation and Ostwald Ripening during Annealing Using Freeze-Drying Microscopy. Pharmaceutics. 2022;14(6), 1176.



Lyo cake appearance I

Typical cake defects

- **Collapse / Meltback:**

- Collapse: Viscous flow resulting in loss of microstructure established by the freezing process
- Meltback: poorly defined term mostly referring to melting of frozen matrix or collapse



Figure 1. Collapsed cake: total collapse (left) and partial collapse (center). The vial on the right shows no evidence of collapse.

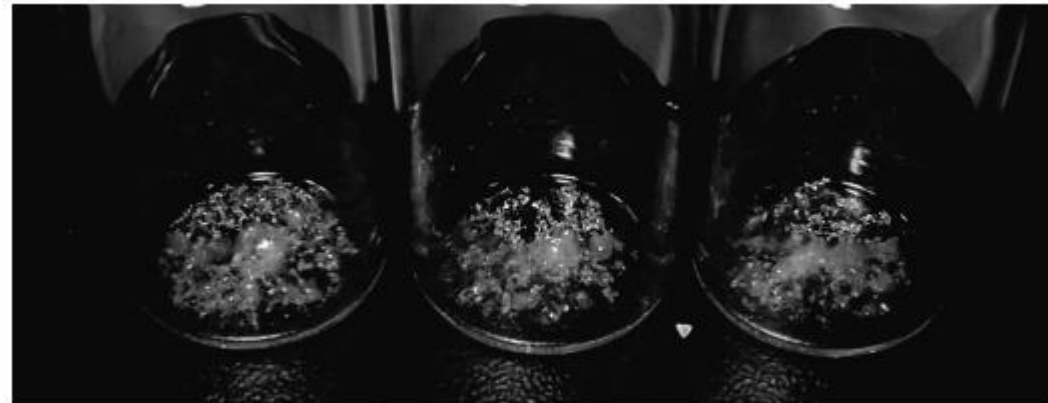


Figure 3. Meltback: could also be a form of collapsed cake.

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Lyo cake appearance II

Typical cake defects

- **Lifted cake**

- Potentially takes place during primary drying
- May be caused by separation of the cake from the inner vial wall → low resistance path for water flow relative to flow through the partially dried solids

- **Cake shrinkage & cracked cake**

- May be related to amount of unfrozen water in amorphous matrix
- Unfrozen water content removed during (secondary) drying
- Causes stress to build up in the cake due to volume contraction
- Release of stress either by cake contraction or cracking



Figure 8. Lifted cake.



Figure 9. Cake shrinkage: this shrinkage is not associated with collapse.



Figure 10. Cracked cake.

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Lyo cake appearance III

Typical cake defects

- **Bubble/Foam formation**