

23 July 2024

Reference: Chinese Pharmacopeial Chapter <0261> Water for Pharmaceutical Purposes

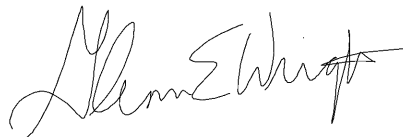
Dear Madam or Sir,

PDA appreciates the opportunity to provide feedback to the Chinese Pharmacopeia on Chapter <0261> Water for Pharmaceutical Purposes. In our attached comments, PDA offers specific comments and feedback that we believe will be helpful in the further development of this important Chapter.

PDA is a non-profit international professional association with over 10,000 individual members, including scientists, industry professionals, and consultants who have an interest in pharmaceuticals, biologics, device manufacturing, and quality. Our comments have been prepared by a committee of PDA members with expertise in the areas covered in the Public Docket on behalf of PDA's Science Advisory Board.

If you have any questions, please do not hesitate to contact me via email at [wright@pda.org](mailto:wright@pda.org).

Sincerely,



Glenn E. Wright  
President and CEO

cc. Josh Eaton, PDA; Carrie Horton, PDA; Jessie Lindner, PDA; Danielle Bretz, PDA

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## PDA Comments to Chinese Pharmacopoeia Guidance: 0261 Water for Pharmaceutical Purposes

<b>General Comments about the Guidance</b>		
<b>Comment</b>	<b>Proposed Change</b>	<b>Rationale</b>
<p>For WFI and Sterile WFI: There is no analysis requirement for Aluminum.</p>	<p>For WFI and Sterile WFI: add Aluminum test with a note of “where it is labeled as intended for use in the manufacture of peritoneal dialysis solutions, hemodialysis solutions, or hemofiltration solutions”.</p>	<p>Proposed changes will account for additional testing necessary to support hemodialysis applications (WFI and Sterile WFI).</p>
<p>Under Appendices 1-2, there are statements that if Conductivity meets requirements, then specific tests are not required. These tests include the following: Acidity or Alkalinity, Heavy Metals, Nitrate, Nitrite, and Ammonia.</p> <p>PDA recommends removing these test requirements along with the clausal statement. If this proposal is not acceptable, refer to the comments below for updates for these sections.</p>	<p>Remove the statements addressing tests not being required if Conductivity meets requirements in Appendices 1-2.</p> <p>Remove the following tests from Appendices 1-2: Acidity or Alkalinity, Heavy Metals, Nitrate, Nitrite, and Ammonia.</p> <p>If this proposal is not acceptable, refer to the comments below for recommended changes on this subject.</p>	<p>If the water meets the conductivity requirements, there is no need to perform a test for acidity or alkalinity, heavy metals, nitrate, nitrite, or ammonia. If any of the other listed tests are not acceptable, the conductivity would not meet requirements.</p> <p>Further, if conductivity does not pass, the water (purified or water for injection) should not be used even if it passes for the subsequent tests of acidity or alkalinity, heavy metals, nitrate, nitrite, and ammonia.</p> <p>This approach will harmonize the recommendations with those found in other current industry and regulatory documents.</p>

**Introduction**

Line Numbers-Original	Line Numbers-Translation	Referenced Text	Comment	Proposed Text	Rationale
10-11	10 - 11	“The source water used for purified water, water for injection and sterile water for injection is usually drinking water.”	PDA proposes to change text to harmonize with the recommendations found in other current industry and regulatory documents.	“The source water used for purified water, water for injection and sterile water for injection is at a <b>minimum drinking water.</b> ”	By updating the wording, the recommendation of this guidance will be more aligned to that found in other current industry and regulatory documents.
15-17	15-16	“The system for water for pharmaceutical purposes should be qualified/validated, systems of daily monitoring, analysis and report should be established, and complete original data should be kept for reference.”	PDA suggests changing recommendation of “daily monitoring” to “routine monitoring and breaking the statement into two sentences.	“The system for water for pharmaceutical purposes should be qualified/validated, and <b>include routine monitoring and analysis based on risk assessment. Reports</b> should be established and complete original data should be kept for reference.”	By making this language change, it will harmonize recommendations with current industry and regulatory documents and allow companies the flexibility to determine monitoring frequency based on the system and associated risks.  By dividing the statement into two sentences, it will clarify the intent for the reader.
18-21	18-21	“Sterilization could be conducted using thermal or chemical processes. The cleaning and disinfection methods used and the removal of cleaning agents and disinfectants after chemical treatments should be qualified/validated.”	PDA encourages the use of the term “sanitization” in place of “sterilization”.  PDA recommends expanding the statement to include other process beyond thermal or chemical processes. Additionally, PDA suggests using the general terminology	“ <b>Sanitization</b> could be conducted using thermal or chemical processes <b>or other suitable processes</b> . Removal of <b>chemical agents used in the process</b> should be qualified/validated.”	Water systems are sanitized at high temperatures but not designed to support sterilization.  By updating the statement to include “or other suitable processes, it will account for other optional processes such as ultra-violet (UV) radiation that can be used for

**Introduction**

Line Numbers-Original	Line Numbers-Translation	Referenced Text	Comment	Proposed Text	Rationale
			of cleaning agents and focusing on the removal of chemical agents being qualified/validated.		sanitization. Generalizing will focus on the critical elements of focusing on removal of chemical solutions without specifying specific category of solutions.

**Drinking Water**

Line Numbers-Original	Line Numbers-Translation	Referenced Text	Comment	Proposed Text	Rationale
25-27	25-27	“Drinking water may be used for washing crude drugs before purification, for preliminarily washing pharmaceutical appliances and may also be used as the solvent for extraction of medicine material crude slices.”	PDA recommends expanding the statement and using more general descriptions of the manufacturing process/equipment.	“Drinking water may be used <b>in the early stages of cleaning pharmaceutical manufacturing equipment and product-contact components. Drinking water is also the minimum quality of water that should be used for the preparation of drug substances and other bulk pharmaceutical ingredients. Where compatible with the processes, the contaminant levels allowed in drinking water are generally considered safe for use in preparing official substances</b>	By updating this statement, it will provide more guidance to the user regarding suitable uses for drinking quality water in the manufacturing process.  This update in language also harmonizes the statement with current industry and regulatory documents.

Drinking Water					
Line Numbers-Original	Line Numbers-Translation	Referenced Text	Comment	Proposed Text	Rationale
				and other drug substances. A risk assessment may be performed to determine suitability of proposed water sources for specific dosage forms or products.”	

Water for Injection					
Line Numbers-Original	Line Numbers-Translation	Referenced Text	Comment	Proposed Text	Rationale
48-52	48-52	<p><b>“Water for injection</b></p> <p>Water for injection is water prepared by distillation of purified water, and its quality should comply with the requirements of the Annex 2 of Water for injection in this General Chapter; or water for injection is prepared through a purification process equivalent to distillation, and its preparation process should meet the requirements of relevant procedures of regulatory authorities, and its quality should meet relevant provisions. It contains no</p>	<p>PDA proposes to update the statement to clarify that drinking water can be used as feed water for the preparation of Water for Injection.</p> <p>Additionally, PDA proposes to clarify the water quality characteristics to assess when determining process equivalency.</p>	<p><b>“Water for injection</b></p> <p>Water for injection is water prepared by distillation of <b>Drinking Water</b> or purified water, and its quality should comply with the requirements of the Annex 2 of Water for injection in this General Chapter; or water for injection is prepared <b>from Drinking Water or Purified Water</b> through a purification process equivalent to distillation <b>in the removal of chemicals, Microorganisms and endotoxins.</b> Its quality should</p>	<p>The feed water to generate Water for Injection (WFI) is ruled as Drinking Water (suitably pretreated) or Purified Water in most international Pharmacopoeias.</p> <p>Regarding the alternative to Distillation, the updated language provides guidance for assessing equivalent processes.</p>

### Water for Injection

Line Numbers-Original	Line Numbers-Translation	Referenced Text	Comment	Proposed Text	Rationale
		additives.”		meet relevant provisions. It contains no additives.	

### Microbial Monitoring

Line Numbers-Original	Line Numbers-Translation	Referenced Text	Comment	Proposed Text	Rationale
75-77	75-77	“Take no less than 1ml purified water or no less than 100ml water for injection, process with the membrane filtration method, culture with R2A agar medium at 30-35°C for no less than 5 days, and carry out the test as described (1105).”	PDA recommends adding a statement allowing the use of other suitable media for culture.	“Take no less than 1ml purified water or no less than 100ml water for injection, process with the membrane filtration method, culture with R2A agar medium at 30-35°C for no less than 5 days, and carry out the test as described (1105). <b>For optimum recovery, other suitable low nutrient media, incubation time and/or temperature may be needed.</b> ”	Other types of media can be suitable for culture in addition to R2A. By adding this statement, it will clarify for the reader that they should select the media type most suitable for their specific process. This update also harmonizes the statement with current industry and regulatory documents.
79-82	79-82	“The microbial limit of purified water shall be no more than 100 cfu/ml, and that of water for injection shall be more than 10 cfu/100ml. On the premise of meeting the limit standards, appropriate alert and action limits should be set	PDA recommends adding missing word “no” in the statement “water for injection shall be more than 10cfu/100 ml”.  PDA also recommends adding clarifying statement and rewording last sentence.	“The microbial limit of purified water shall be no more than 100 cfu/ml, and that of water for injection shall be <b>no</b> more than 10 cfu/100ml. On the premise of meeting the limit standards, appropriate alert and action limits should be set to monitor	Adding in missing word will make the statement/intent accurate and align with the wording used for the purified water microbial limit.  By adding the proposed statement, it will harmonize the language with current

**Microbial Monitoring**

Line Numbers-Original	Line Numbers-Translation	Referenced Text	Comment	Proposed Text	Rationale
		to monitor adverse trends. Tighter alert and action limits may be imposed as needed for high-risk drug products or aseptic processes.”		adverse trends. <b>For aseptic processing, stricter alert levels may need to be applied.”</b>	industry and regulatory documents (e.g., European Pharmacopoeia).  By removing “high-risk” it will eliminate confusion for the reader and focus on aseptic processing of sterile products.
89-92	89-92	“Suitability test for R2A agar Carry out the method described under growth promotion of soya bean casein digest agar in Growth Promotion of the Counting Media, Pseudomonas aeruginosa and Bacillus subtilis as test strain, complies with the microbiological examination of non-sterile products: microbial enumeration tests <1105>. Complies with the requirements.”	PDA proposes to generalize the media reference and expand statement to include use of soya bean casein digest broth.	“Suitability test for <b>the media used (e.g., R2A or other suitable, validated media) should be carried out per</b> the method described under growth promotion of soya bean casein digest agar <b>or soya bean casein digest broth</b> in Growth Promotion of the Counting Media, Pseudomonas aeruginosa and Bacillus subtilis as test strain, complies with the microbiological examination of non-sterile products: microbial enumeration tests <1105>. Complies with the requirements.”	This update in language will harmonize with those found in <1105> and with current industry and regulatory documents. Additionally, the use of a general media reference aligns with the above proposed change for alternate media may needing to be used.

**Appendix 1: Purified Water**

Line Numbers-Original	Line Numbers-Translation	Referenced Text	Comment	Proposed Text	Rationale
101-103	101-103	“If the conductivity is determined according to the procedure for water for injection in the Determination of Conductivity of Water for Pharmaceutical Use (0681), and the conductivity meets the requirements, the pH value and heavy metal test will not be conducted.”	PDA encourages updating statement to be inclusive of all tests that do no need to be performed if the WFI meets the conductivity requirements.	“If the conductivity is determined according to the procedure for water for injection in the Determination of Conductivity of Water for Pharmaceutical Use (0681), and the conductivity meets the requirements, the, <b>acidity or alkalinity</b> , pH value, <b>nitrate, nitrite, ammonia</b> and heavy metal tests will not be conducted.”	Conductivity test limits were developed such that if you passed the conductivity tests, you would pass the existing compendial chemical tests. Therefore, if the water for injection meets the conductivity requirements, there is no need to perform a test for acidity or alkalinity, nitrate, nitrite, or ammonia. If any of the other listed tests are not acceptable, the conductivity would not meet requirements. By updating the statement, it will clarify this for the reader.
111-114	111-114	“If the conductivity is determined according to the procedure for water for injection in the Determination of Conductivity of Water for Pharmaceutical Use (0681), and the conductivity is determined to meet the requirements according to the first step of determination method, nitrate, nitrite and ammonia test will not be conducted.”	PDA recommends removing this statement and instead update the language in lines 101-103 to include the additional tests.	<del>“If the conductivity is determined according to the procedure for water for injection in the Determination of Conductivity of Water for Pharmaceutical Use (0681), and the conductivity is determined to meet the requirements according to the first step of determination method, nitrate, nitrite and ammonia test will not be conducted.”</del>	There is no need for a separate statement because if the water meets the conductivity requirements (regardless of which stage of testing), there is no need to perform a test for nitrate, nitrite or ammonia. Therefore, all applicable tests can be stated in the first statement.



### Appendix 2: Water for Injection

Line Numbers-Original	Line Numbers-Translation	Referenced Text	Comment	Proposed Text	Rationale
144-147	144-147	“If the conductivity is determined according to the procedure for water for injection in the Determination of Conductivity of Water for Pharmaceutical Use (0681), and the conductivity is determined to meet the requirements according to the first step of determination method, nitrate, nitrite and ammonia test will not be conducted.”	PDA recommends updating the language to align with the approach in the proposed changes in lines 101-103 with regards to conductivity acceptance requirements.	“If the conductivity is determined according to the procedure for water for injection in the Determination of Conductivity of Water for Pharmaceutical Use (0681), <b>and the conductivity meets the requirements</b> , nitrate, nitrite and ammonia test will not be conducted.”	By making this change, the verbiage will be aligned with the verbiage used in lines 101-103 and aligned with the proposed verbiage for lines 111-114.

### Appendix 3: Sterile Water for Injection

Line Numbers-Original	Line Numbers-Translation	Referenced Text	Comment	Proposed Text	Rationale
177-213	177-213	“ <b>pH</b> To 100 ml add 0.3 ml of saturated potassium chloride solution. pH 5.0-7.0 (0631). <b>Chlorides, sulfates and calcium</b> Add 50 ml to each of three test tubes, to the first	PDA suggests removing all tests with the exception of oxidizable substances and conductivity.	<b>Oxidizable substances</b> Boil 100 ml with 10 ml of dilute sulphuric acid, add 0.10 ml of potassium permanganate (0.02 mol/L) VS and continue to boil for 10 minutes; the pink colour does not	Sterile Water for injection (SWFI) is made from Water for Injection (WFI) and thus should meet all these requirements. The oxidizable substances and conductivity tests will provide assessment

**Appendix 3: Sterile Water for Injection**

Line Numbers-Original	Line Numbers-Translation	Referenced Text	Comment	Proposed Text	Rationale
		<p>tube add 5 drops of nitric acid and 1 ml of silver nitrate TS; to the second tube add 5 ml of barium chloride TS; and to the third tube add 2 ml of ammonium oxalate TS; no opalescence is produced in any one of the tubes.</p> <p><b>Nitrate</b> To 5 ml in a test tube, cooled in an ice bath, add 0.4 ml of 10 % potassium chloride solution and 0.1 ml of 0.1 % diphenylamine solution in sulfuric acid and shake well. Add 5 ml of sulfuric acid dropwise and allow to stand in a water bath at 50°C for 15 minutes. Any colour produced is not more intense than that of the reference solution (Dissolve a quantity of potassium nitrate in water to produce a solution containing 1 µg of NO<sub>3</sub> per ml. To 0.3 ml add 4.7 ml of nitrate-free water, repeat the operation using the solution instead of the substance being</p>		<p>disappear completely.</p> <p><b>Conductivity</b> Complies with the test for water conductivity (0681).</p>	<p>for potential impurities that may be introduced due to the SWFI packaging. The other chemical tests are covered either through the WFI test requirements or through the conductivity tests.</p> <p>This update in language will harmonize the recommendations with those found in current industry and regulatory documents.</p>

**Appendix 3: Sterile Water for Injection**

Line Numbers-Original	Line Numbers-Translation	Referenced Text	Comment	Proposed Text	Rationale
		<p>examined) (0.000 006%).</p> <p><b>Nitrite</b> To 10 ml in a Nessler cylinder add 1 ml of sulfanilic amide solution (dissolve 1 g in 100 ml of dilute hydrochloric acid ) and 1 ml of 0.1% N-naphthylethylenediamine dihydrochloride solution and shake well. Any colour produced is not more intense than that of the reference solution (Dissolve a quantity of sodium nitrite in water to produce a solution containing 1 µg of NO<sub>2</sub> per ml. To 0.2 ml add 9.8 ml of nitrite-free water, repeat the operation using the solution instead of the substance being examined) (0.000 002%).</p> <p><b>Ammonia</b> To 50 ml add 2 ml of alkaline mercuric potassium iodide TS and allow to stand for 15 minutes. Any colour produced is not more intense than that of a reference prepared by mixing 1.0ml of ammonium chloride solution</p>			

**Appendix 3: Sterile Water for Injection**

Line Numbers-Original	Line Numbers-Translation	Referenced Text	Comment	Proposed Text	Rationale
		<p>(dissolve 31.5 mg of ammonium chloride CRS in ammonia-free distilled water to produce 1000 ml) with 48 ml of ammonia-free distilled water and 2 ml of alkaline mercuricpotassium iodide TS (0.000 02%).</p> <p><b>Carbon dioxide Mix</b> 25 ml with 25 ml of calcium hydroxide TS in a 50 ml stoppered cylinder, shake, allow to stand; no opalescence is produced within 1 hour.</p> <p><b>Oxidizable substances</b> Boil 100 ml with 10 ml of dilute sulphuric acid, add 0.10 ml of potassium permanganate (0.02 mol/L) VS and continue to boil for 10 minutes; the pink colour does not disappear completely.</p> <p><b>Conductivity</b> Complies with the test for water conductivity (0681).</p> <p><b>Non-volatile substances</b></p>			

**Appendix 3: Sterile Water for Injection**

Line Numbers-Original	Line Numbers-Translation	Referenced Text	Comment	Proposed Text	Rationale
		<p>Evaporate 100 ml in an evaporating dish previously dried at 105°C to constant weight, to dryness on a water bath, and dry at 105°C to constant weight, the residue is not more than 1 mg.</p> <p><b>Heavy metals</b> To 100 ml add 19 ml of water, evaporate to 20 ml, cool. Add 2 ml of acetate BS (pH 3.5) and a quantity of water to 25 ml. Add 2 ml of thioacetamide TS, mix and allow to stand for 2 minutes. Any colour produced is not more intense than that of the solution prepared in the same manner using 1.0 ml of lead standard solution mixed with 19 ml of water (0.00001%).”</p>			

## 0261 制药用水

1 制药用水用于药物生产过程和药物制剂的制备。

2 本版药典中所指的制药用水，因其使用的范围不同而分为饮用水、纯化水、  
3 注射用水和灭菌注射用水。一般应根据各生产工序或使用目的与要求选用适宜的  
4 制药用水。药品生产企业应确保制药用水的质量符合预期用途的要求。

5 纯化水、注射用水和灭菌注射用水的原水通常为饮用水。

6 制药用水系统的设计、材质选择、制备过程、储存、分配、使用和维护等均  
7 应符合药品生产质量管理规范的要求。

8 制药用水系统应经过确认/验证，并建立日常监控、检测和报告制度，有完  
9 善的原始记录备查。

10 制药用水系统应定期进行清洗与消毒，消毒可以采用热处理或化学处理等方  
11 法。采用的清洗与消毒方法，以及化学处理后清洗剂与消毒剂的去除应经过确认  
12 /验证。

13 **饮用水** 为天然水经净化处理所得，其质量应符合现行中华人民共和国国家  
14 标准《生活饮用水卫生标准》。

15 饮用水可作为药材净制时的漂洗、制药用具的粗洗用水，一般也可以作为饮  
16 片的提取溶剂。

17 **纯化水** 为饮用水经蒸馏法、离子交换法、反渗透法或其它适宜的方法制备  
18 所得，其质量应符合本通则附 1 纯化水的规定。不含任何附加剂。

19 纯化水中可能存在的元素杂质是药品生产中元素杂质的潜在来源之一，必要  
20 时，可参考 ICH 元素杂质指导原则（Q3D）来评估和控制药品中元素杂质。

21 纯化水在制备、储存和分配过程中，应采取适当的措施确保微生物得到充分  
22 控制和监测。采用本通则“微生物监测”项下方法进行微生物监测。具体可参考  
23 制药用水微生物监测和控制指导原则（指导原则 9209）。

24 纯化水可作为配制普通药物制剂用的溶剂，中药注射剂、眼用制剂等无菌制  
25 剂所用饮片的提取溶剂，口服、外用制剂配制用溶剂或稀释剂，非无菌制剂用器  
26 具的精洗用水，非无菌制剂所用饮片的提取溶剂等。纯化水不得用于注射剂的配  
27 制和稀释。

28 **注射用水** 为纯化水经蒸馏所得，其质量应符合本通则附 2 注射用水的规定；

30 或为通过一个等同于蒸馏的纯化工艺制备所得，其制备工艺应符合监管部门有关  
31 程序要求，其质量应符合有关规定。不含任何附加剂。

32 注射用水中可能存在的元素杂质是药品生产中元素杂质的潜在来源之一，必  
33 要时，可参考 ICH 元素杂质指导原则（Q3D）来评估和控制药品中元素杂质。

34 注射用水在制备、储存和分配过程中，应采取适当的措施确保微生物/细菌  
35 内毒素得到充分控制和监测。采用本通则“微生物监测”项下方法进行微生物监  
36 测。具体可参考制药用水微生物监测和控制指导原则（指导原则 9209）。

37 注射用水的储存方式和储存期限应经过确认/验证，确保水质符合质量要求。

38 注射用水可作为配制注射剂、眼用制剂等的溶剂或稀释剂，以及容器的精洗  
39 等。

40 **灭菌注射用水** 为注射用水按照注射剂生产工艺制备所得，其质量应符合附  
41 3 灭菌注射用水的规定。不含任何附加剂。

42 主要用于注射用无菌药品粉末的溶剂或注射剂的稀释剂。

43 灭菌注射用水灌装规格应与临床需要相适应，避免大规格、多次使用造成的  
44 污染。

45 **微生物监测** 采用下列方法，或经充分验证的等同或更优方法，进行微生物  
46 监测。

47 纯化水取样不少于 1ml，注射用水取样不少于 100ml，经薄膜过滤法处理，  
48 采用 R2A 琼脂培养基，30-35℃ 培养不少于 5 天，依法检查（通则 1105）。可根  
49 据监测数据适当调整检验量，以能够监测到微生物数量变化。

50 纯化水微生物限度标准为不大于 100cfu/ml，注射用水微生物限度标准为不  
51 大于 10cfu/100ml。应在满足限度标准的前提下，设置适当的警戒限度和纠偏限  
52 度，以监测不良趋势。如用于高风险制剂或无菌工艺，可根据需要设定更严格的  
53 警戒限度和纠偏限度。

54 R2A 琼脂培养基处方及制备：酵母浸出粉 0.5g、蛋白胨 0.5g、酶蛋白水解  
55 物 0.5g、葡萄糖 0.5g、可溶性淀粉 0.5g、磷酸氢二钾 0.3g、无水硫酸镁 0.024g、  
56 丙酮酸钠 0.3g、琼脂 15g、纯化水 1000ml。除葡萄糖、琼脂外，取上述成分，混  
57 合，微温溶解，调节 pH 值使加热后在 25℃ 的 pH 值为  $7.2 \pm 0.2$ ，加入琼脂，  
58 加热溶化后，再加入葡萄糖，摇匀，分装，灭菌。

59 R2A 琼脂培养基适用性检查试验 照非无菌产品微生物限度检查：微生物计

60 数法（通则 1105）中“计数培养基适用性检查”的胰酪大豆胨培养基的适用性  
61 检查方法进行，试验菌株为铜绿假单胞菌和枯草芽孢杆菌。应符合规定。

## 62 附 1：纯化水

63 **性状** 本品为无色的澄清液体。

64 **总有机碳** 不得过 0.50 mg/L（通则 0682）。

65 **易氧化物** 取本品 100ml，加稀硫酸 10ml，煮沸后，加高锰酸钾滴定液  
66 (0.02mol/L) 0.10ml，再煮沸 10 分钟，粉红色不得完全消失。

67 以上总有机碳和易氧化物两项可选做一项。

68 **电导率** 应符合规定（通则 0681）。

69 如按制药用水电导率测定法（通则 0681）中注射用水测定法测定电导率，电  
70 导率符合规定，则不再进行酸碱度和重金属检查。

71 **酸碱度** 取本品 10ml，加甲基红指示液 2 滴，不得显红色；另取 10ml，加溴  
72 麝香草酚蓝指示液 5 滴，不得显蓝色。

73 **重金属** 取本品 100ml，加水 19ml，蒸发至 20ml，放冷，加醋酸盐缓冲液（pH  
74 3.5）2ml 与水适量使成 25ml，加硫代乙酰胺试液 2ml，摇匀，放置 2 分钟，与标  
75 准铅溶液 1.0ml 加水 19ml 用同一方法处理后的颜色比较，不得更深（0.000 01%）。

76 如按制药用水电导率测定法（通则 0681）中注射用水测定法测定电导率，电  
77 导率按判定法第一步判定符合规定，则不再进行硝酸盐、亚硝酸盐和氨检查。

78 **硝酸盐** 取本品 5ml 置试管中，于冰浴中冷却，加 10%氯化钾溶液 0.4ml 与  
79 0.1%二苯胺硫酸溶液 0.1ml，摇匀，缓缓滴加硫酸 5ml，摇匀，将试管于 50℃水  
80 浴中放置 15 分钟，溶液产生的蓝色与标准硝酸盐溶液 [取硝酸钾 0.163g，加水  
81 溶解并稀释至 100ml，摇匀，精密量取 1ml，加水稀释成 100ml，再精密量取 10ml，  
82 加水稀释成 100ml，摇匀，即得（每 1ml 相当于 1 μg NO<sub>3</sub>）] 0.3ml，加无硝酸盐  
83 的水 4.7ml，用同一方法处理后的颜色比较，不得更深（0.000 006%）。

84 **亚硝酸盐** 取本品 10ml，置纳氏管中，加对氨基苯磺酰胺的稀盐酸溶液  
85 (1→100) 1ml 与盐酸萘乙二胺溶液 (0.1→100) 1ml，产生的粉红色，与标准亚  
86 硝酸盐溶液 [取亚硝酸钠 0.750g（按干燥品计算），加水溶解，稀释至 100ml，  
87 摇匀，精密量取 1ml，加水稀释成 100ml，摇匀，再精密量取 1ml，加水稀释成  
88 50ml，摇匀，即得（每 1ml 相当于 1 μg NO<sub>2</sub>）] 0.2ml，加无亚硝酸盐的水 9.8ml，  
89 用同一方法处理后的颜色比较，不得更深（0.000 002%）。



90 氨 取本品 50ml, 加碱性碘化汞钾试液 2ml, 放置 15 分钟; 如显色, 与氯化  
91 铵溶液 (取氯化铵 31.5mg, 加无氨水适量使溶解并稀释成 1000ml) 1.5ml, 加无  
92 氨水 48ml 与碱性碘化汞钾试液 2ml 制成的对照液比较, 不得更深 (0.000 03%)。

93 微生物监测 按照制药用水 (通则 0261) 中微生物监测的要求进行。

94 注: ① 本品密闭保存。② 基于风险评估, 必要时, 可测定本品的不挥发物  
95 (可按下述测定方法测定)。

96 取本品 100ml, 置 105℃恒重的蒸发皿中, 在水浴上蒸干, 并在 105℃干燥  
97 至恒重, 遗留残渣不得过 1mg。

## 98 附 2: 注射用水

99 性状 本品为无色的澄明液体。

100 总有机碳 不得过 0.50 mg/L (通则 0682)。

101 电导率 应符合规定 (通则 0681)。

102 如按制药用水电导率测定法 (通则 0681) 中注射用水测定法测定电导率, 电  
103 导率按判定法第一步判定符合规定, 则不再进行硝酸盐、亚硝酸盐和氨检查。

104 硝酸盐 取本品 5ml 置试管中, 于冰浴中冷却, 加 10%氯化钾溶液 0.4ml 与  
105 0.1%二苯胺硫酸溶液 0.1ml, 摇匀, 缓缓滴加硫酸 5ml, 摇匀, 将试管于 50℃水  
106 浴中放置 15 分钟, 溶液产生的蓝色与标准硝酸盐溶液 [取硝酸钾 0.163g, 加水  
107 溶解并稀释至 100ml, 摇匀, 精密量取 1ml, 加水稀释成 100ml, 再精密量取 10ml,  
108 加水稀释成 100ml, 摇匀, 即得 (每 1ml 相当于  $1 \mu\text{gNO}_3$ )] 0.3ml, 加无硝酸盐的  
109 水 4.7ml, 用同一方法处理后的颜色比较, 不得更深 (0.000 006%)。

110 亚硝酸盐 取本品 10ml, 置纳氏管中, 加对氨基苯磺酰胺的稀盐酸溶液  
111 (1→100) 1ml 与盐酸萘乙二胺溶液 (0.1→100) 1ml, 产生的粉红色, 与标准亚  
112 硝酸盐溶液 [取亚硝酸钠 0.750g (按干燥品计算), 加水溶解, 稀释至 100ml,  
113 摇匀, 精密量取 1ml, 加水稀释成 100ml, 摇匀, 再精密量取 1ml, 加水稀释成  
114 50ml, 摇匀, 即得 (每 1ml 相当于  $1 \mu\text{gNO}_2$ )] 0.2ml, 加无亚硝酸盐的水 9.8ml,  
115 用同一方法处理后的颜色比较, 不得更深 (0.000 002%)。

116 氨 取本品 50ml, 加碱性碘化汞钾试液 2ml, 放置 15 分钟; 如显色, 与氯化  
117 铵溶液 (取氯化铵 31.5mg, 加无氨水适量使溶解并稀释成 1000ml) 1.0ml, 加无  
118 氨水 48ml 与碱性碘化汞钾试液 2ml 制成的对照液比较, 不得更深 (0.000 02%)。

119 细菌内毒素 取本品, 依法检查 (通则 1143), 每 1ml 中含内毒素的量应小

120 于 0.25EU。

121 **微生物监测** 按照制药用水（通则 0261）中微生物监测的要求进行。

122 注：① 本品密闭保存。② 基于风险评估，必要时，可测定本品的不挥发物  
123 （可按下述测定方法测定）。

124 取本品 100ml，置 105℃恒重的蒸发皿中，在水浴上蒸干，并在 105℃干燥  
125 至恒重，遗留残渣不得过 1mg。

### 126 附 3：灭菌注射用水

127 **性状** 本品为无色的澄明液体；无臭。

128 **pH 值** 取本品 100ml，加饱和氯化钾溶液 0.3ml，依法测定（通则 0631），pH 值  
129 应为 5.0~7.0。

130 **氯化物、硫酸盐与钙盐** 取本品，分置三支试管中，每管各 50ml，第一管中加  
131 硝酸 5 滴与硝酸银试液 1ml，第二管中加氯化钡试液 5ml，第三管中加草酸铵试液  
132 2ml，均不得发生浑浊。

133 **硝酸盐** 取本品 5ml 置试管中，于冰浴中冷却，加 10%氯化钾溶液 0.4ml 与  
134 0.1%二苯胺硫酸溶液 0.1ml，摇匀，缓缓滴加硫酸 5ml，摇匀，将试管于 50℃水  
135 浴中放置 15 分钟，溶液产生的蓝色与标准硝酸盐溶液 [取硝酸钾 0.163g，加水  
136 溶解并稀释至 100ml，摇匀，精密量取 1ml，加水稀释成 100ml，再精密量取 10ml，  
137 加水稀释成 100ml，摇匀，即得（每 1ml 相当于  $1 \mu\text{gNO}_3$ ）] 0.3ml，加无硝酸盐的  
138 水 4.7ml，用同一方法处理后的颜色比较，不得更深（0.000 006%）。

139 **亚硝酸盐** 取本品 10ml，置纳氏管中，加对氨基苯磺酰胺的稀盐酸溶液  
140 （1→100）1ml 与盐酸萘乙二胺溶液（0.1→100）1ml，产生的粉红色，与标准亚  
141 硝酸盐溶液 [取亚硝酸钠 0.750g（按干燥品计算），加水溶解，稀释至 100ml，  
142 摇匀，精密量取 1ml，加水稀释成 100ml，摇匀，再精密量取 1ml，加水稀释成  
143 50ml，摇匀，即得（每 1ml 相当于  $1 \mu\text{gNO}_2$ ）] 0.2ml，加无亚硝酸盐的水 9.8ml，  
144 用同一方法处理后的颜色比较，不得更深（0.000 002%）。

145 **氨** 取本品 50ml，加碱性碘化汞钾试液 2ml，放置 15 分钟；如显色，与氯化  
146 铵溶液（取氯化铵 31.5mg，加无氨水适量使溶解并稀释成 1000ml）1.0ml，加无  
147 氨水 48ml 与碱性碘化汞钾试液 2ml 制成的对照液比较，不得更深（0.000 02%）。

148 **二氧化碳** 取本品 25ml，置 50ml 具塞量筒中，加氢氧化钙试液 25ml，密塞  
149 振摇，放置，1 小时内不得发生浑浊。

150 **易氧化物** 取本品 100ml, 加稀硫酸 10ml, 煮沸后, 加高锰酸钾滴定液  
151 (0.02mol/L)0.10ml, 再煮沸 10 分钟, 粉红色不得完全消失。

152 **电导率** 应符合规定 (通则 0681)。

153 **不挥发物** 取本品 100ml, 置 105℃恒重的蒸发皿中, 在水浴上蒸干, 并在  
154 105℃干燥至恒重, 遗留残渣不得过 1mg。

155 **重金属** 取本品 100ml, 加水 19ml, 蒸发至 20ml, 放冷, 加醋酸盐缓冲液 (pH  
156 3.5) 2ml 与水适量使成 25ml, 加硫代乙酰胺试液 2ml, 摇匀, 放置 2 分钟, 与标  
157 准铅溶液 1.0ml 加水 19ml 用同一方法处理后的颜色比较, 不得更深(0.000 01%)。

158 **细菌内毒素** 取本品, 依法检查 (通则 1143), 每 1ml 中含内毒素的量应小  
159 于 0.25EU。

160 **其他** 应符合注射剂 (通则 0102) 项下有关的各项规定。

161 **注:** ①本品为溶剂、冲洗剂。②本品密闭保存。

162 起草单位: 北京市药品检验研究院、中国食品药品检定研究院、中国医药设备工  
163 程协会

164 参与单位: RDPAC、江苏省药学会

165 RDPAC 制药用水工作组、中国医药设备工程协会制药用水工作组

166 上海谊康科技有限公司、百特 (中国) 投资有限公司、荣昌生物制药 (烟台) 股  
167 份有限公司、华润双鹤药业股份有限公司、厦门特宝生物工程股份有限公司、江  
168 苏恒瑞医药股份有限公司、正大天晴药业集团股份有限公司、扬子江药业集团有  
169 限公司、齐鲁制药有限公司、湖北葛店人福药用辅料有限责任公司、四川科伦药  
170 业股份有限公司、正大天晴药业集团南京顺欣制药股份有限公司、辰欣药业股份  
171 有限公司、西安杨森制药有限公司、常州四药制药有限公司、苏州百特医疗用品  
172 有限公司、无锡药明生物技术有限公司、费森尤斯卡比华瑞制药有限公司、赛诺  
173 菲 (中国) 投资有限公司、诺和诺德 (中国) 制药有限公司、葛兰素史克 (中国)  
174 投资有限公司 (排名不分先后)

175 **联系电话:**

176 **正文、附 1、附 2:** 010-52779625 **附 3:** 010-67079559

## 0261 Water for Pharmaceutical Purposes

Water for pharmaceutical purposes is used in the production process and the preparation of medicines in pharmaceutical industry.

In this edition of the Pharmacopoeia of the People's Republic of China, water for pharmaceutical purposes includes drinking water, purified water, water for injection and sterile water for injection according to the purposes for which it is to be used.

Water for pharmaceutical purposes should be suitably selected according to manufacturing processes or usage purposes and requirements. The quality of water for pharmaceutical purposes should comply with the requirements of expected usage.

The source water used for purified water, water for injection and sterile water for injection is usually drinking water.

The design, material selection, preparation process, storage, distribution, use and maintenance of the system for water for pharmaceutical purposes shall comply with the requirements of the Good Manufacturing Practice.

The system for water for pharmaceutical purposes should be qualified/validated, systems of daily monitoring, analysis and report should be established, and complete original data should be kept for reference.

The system for water preparation should be cleaned and sterilized periodically. Sterilization could be conducted using thermal or chemical processes.

The cleaning and disinfection methods used and the removal of cleaning agents and disinfectants after chemical treatments should be qualified/validated.

**Drinking water** Drinking water is prepared by purification of natural water and its quality should comply with the current National Standard of the People's Republic of China (“Hygienic Standard of Living Drinking Water”).

Drinking water may be used for washing crude drugs before purification, for preliminarily washing pharmaceutical appliances and may also be used as the solvent for extraction of medicine material crude slices.

**Purified water** Purified water is prepared from drinking water by distillation, ion exchange, reverse osmosis or by means of any other appropriate methods. Its quality should comply with the requirements of the Annex 1 of Purified water in this General Chapter. It contains no additives.

Elemental impurities that may be present in purified water are one of the potential sources of elemental impurities in pharmaceutical manufacturing and, where necessary, can be assessed and controlled in pharmaceutical products by reference to ICH Guideline for Elemental Impurities (Q3D).

Purified water is prepared, stored and distributed with appropriate measures to ensure adequate microbial control and monitoring. Microbial monitoring is performed with the method under “Microbial monitoring” in this General Chapter. For details, please refer to the Guidelines for Microbiological Monitoring and Control of Water for Pharmaceutical Purposes (Guideline 9209).

Purified water may be used as the solvent for preparation of ordinary pharmaceutical

42 preparations, or as the solvent for extraction of medicine material crude slices for preparation  
43 of sterile preparations such as injections or eye drops of traditional Chinese drugs, or as the  
44 solvent or diluent for preparation of oral preparations or preparations for external use, or as the  
45 water for precisely washing appliances for nonsterile preparations, or as the solvent for  
46 extraction of medicine material crude slices for nonsterile preparations. Purified water should  
47 not be used for preparation of injections or as their diluent.

48 **Water for injection** Water for injection is water prepared by distillation of purified water, and  
49 its quality should comply with the requirements of the Annex 2 of Water for injection in this  
50 General Chapter; or water for injection is prepared through a purification process equivalent to  
51 distillation, and its preparation process should meet the requirements of relevant procedures of  
52 regulatory authorities, and its quality should meet relevant provisions. It contains no additives.

53 Elemental impurities that may be present in water for injection are one of the potential sources  
54 of elemental impurities in pharmaceutical manufacturing and, where necessary, can be assessed  
55 and controlled in pharmaceutical products by reference to ICH Guideline for Elemental  
56 Impurities (Q3D).

57 Water for injection is prepared, stored and distributed with appropriate measures to ensure  
58 adequate microorganism/bacterial endotoxin control and monitoring. Microbial monitoring is  
59 performed with the method under “Microbial monitoring” in this General Chapter. For details,  
60 please refer to the Guidelines for Microbiological Monitoring and Control of Water for  
61 Pharmaceutical Purposes (Guideline 9209).

62 The storage condition and shelf life of water for injection should be qualified/validated to ensure  
63 that the water quality meets the quality requirements. Water for injection is used as the solvent  
64 or diluent for preparation of injections, ophthalmic preparations, etc., and as the water for fine  
65 washing of containers.

66 **Sterile water for injection** Sterile water for injection is prepared from water for injection  
67 according to technological conditions under which injections are prepared, and its quality  
68 should comply with the requirements of Annex 3 of Sterile water for injection in this General  
69 Chapter. It contains no additives.

70 It is mainly used as the solvent for sterile powders for injection or as dilute for injections.

71 The specification of filling for Sterile water for injection should meet the clinical requirements.  
72 Contamination due to large volume or multi-use must be avoided.

73 **Microbial monitoring** Microbial monitoring is performed with the following method, or a fully  
74 validated equivalent or superior method.

75 Take no less than 1ml purified water or no less than 100ml water for injection, process with the  
76 membrane filtration method, culture with R2A agar medium at 30-35°C for no less than 5 days,  
77 and carry out the test as described (1105). The test quantity may be adjusted appropriately based  
78 on monitoring data to enable detection of changes in the number of microorganisms.

79 The microbial limit of purified water shall be no more than 100 cfu/ml, and that of water for  
80 injection shall be more than 10 cfu/100ml. On the premise of meeting the limit standards,  
81 appropriate alert and action limits should be set to monitor adverse trends. Tighter alert and  
82 action limits may be imposed as needed for high-risk drug products or aseptic processes.

83 Prescription and preparation of R2A agar: yeast extract 0.5g, protease peptone 0.5g, casein  
84 hydrolysate 0.5g, glucose 0.5g, starch 0.5g, dipotassium hydrogen phosphate 0.3g, magnesium

85 sulfate, anhydrous 0.024g, sodium pyruvate 0.3g, agar 15g, purified water 1000ml. Mix the  
86 above ingredients in water except for glucose and agar, warming slightly until the substances  
87 are dissolved. Adjust the pH so that after heating it is  $7.2 \pm 0.2$  at  $25^{\circ}\text{C}$ . Add agar, heat until  
88 melted and then add glucose, shake thoroughly, dispense and sterilize.

89 Suitability test for R2A agar Carry out the method described under growth promotion of soya  
90 bean casein digest agar in Growth Promotion of the Counting Media, Pseudomonas aeruginosa  
91 and Bacillus subtilis as test strain, complies with the microbiological examination of non-sterile  
92 products: microbial enumeration tests <1105>. Complies with the requirements.

### 93 **Appendix 1: Purified water**

94 **Description** A clear, colourless liquid, odourless.

95 **Total organic carbon** Not more than 0.50 mg per L <0682>.

96 **Oxidizable substances** Boil 100 ml with 10 ml of dilute sulphuric acid, add 0.10 ml of  
97 potassium permanganate (0.02 mol/L) VS and continue to boil for 10 minutes; the pink colour  
98 does not disappear completely.

99 Total organic carbon and Oxidizable substances may be used alternatively.

100 **Conductivity** Complies with the test for water conductivity (0681).

101 If the conductivity is determined according to the procedure for water for injection in the  
102 Determination of Conductivity of Water for Pharmaceutical Use (0681), and the conductivity  
103 meets the requirements, the pH value and heavy metal test will not be conducted.

104 **Acidity or alkalinity** To 10 ml add 2 drops of methyl red IS, no red colour is produced. To  
105 another 10 ml add 5 drops of bromothymol blue IS, no blue colour is produced.

106 **Heavy metals** To 100 ml add 19 ml of water, evaporate to 20 ml, cool. Add 2 ml of acetate BS  
107 ( pH 3.5 ) and a quantity of water to 25 ml. Add 2 ml of thioacetamide TS, mix and allow to  
108 stand for 2 minutes. Any colour produced is not more intense than that of the solution prepared  
109 in the same manner using 1.0 ml of lead standard solution mixed with 19 ml of water  
110 (0.00001%).

111 If the conductivity is determined according to the procedure for water for injection in the  
112 Determination of Conductivity of Water for Pharmaceutical Use (0681), and the conductivity  
113 is determined to meet the requirements according to the first step of determination method,  
114 nitrate, nitrite and ammonia test will not be conducted.

115 **Nitrate** To 5 ml in a test tube, cooled in an ice bath, add 0.4 ml of 10 % potassium chloride  
116 solution and 0.1 ml of 0.1 % diphenylamine solution in sulfuric acid and shake well. Add 5 ml  
117 of sulfuric acid dropwise and allow to stand in a water bath at  $50^{\circ}\text{C}$  for 15 minutes. Any colour  
118 produced is not more intense than that of the reference solution (Dissolve a quantity of  
119 potassium nitrate in water to produce a solution containing  $1\ \mu\text{g}$  of  $\text{NO}_3$  per ml. To 0.3 ml add  
120 4.7 ml of nitrate-free water, repeat the operation using the solution instead of the substance  
121 being examined) (0.000 006%).

122 **Nitrite** To 10 ml in a Nessler cylinder add 1 ml of sulfanilic amide solution (dissolve 1 g in 100  
123 ml of dilute hydrochloric acid ) and 1 ml of 0.1% N-naphthylethylenediamine dihydrochloride  
124 solution and shake well. Any colour produced is not more intense than that of the reference  
125 solution (Dissolve a quantity of sodium nitrite in water to produce a solution containing  $1\ \mu\text{g}$   
126 of  $\text{NO}_2$  per ml. To 0.2 ml add 9.8 ml of nitrite-free water, repeat the operation using the solution

127 instead of the substance being examined) (0.000 002%).

128 **Ammonia** To 50 ml add 2 ml of alkaline mercuric potassium iodide TS and allow to stand for  
129 15 minutes. Any colour produced is not more intense than that of a reference prepared by mixing  
130 1.5 ml of ammonium chloride solution (dissolve 31.5 mg of ammonium chloride CRS in  
131 ammonia-free distilled water to produce 1000 ml) with 48 ml of ammonia-free distilled water  
132 and 2 ml of alkaline mercuricpotassium iodide TS (0.00003%).

133 **Microbial monitoring** Carry out in accordance with the requirements of microbial monitoring  
134 in Water for pharmaceutical purposes (0261).

135 Note: ① Preserve in well closed containers. ② Based on risk assessment, if necessary, the non-  
136 volatile substances of this product can be measured (which can be determined with the  
137 following method).

138 Evaporate 100 ml in an evaporating dish previously dried at 105°C to constant weight, to  
139 dryness on a water bath, and dry at 105°C to constant weight, the residue is not more than 1 mg.

## 140 **Appendix 2: Water for injection**

141 **Description** A clear, colourless liquid; odourless.

142 **Total organic carbon** Not more than 0.50 mg per L (0682).

143 **Conductivity** Complies with the test for water conductivity (0681).

144 If the conductivity is determined according to the procedure for water for injection in the  
145 Determination of Conductivity of Water for Pharmaceutical Use (0681), and the conductivity  
146 is determined to meet the requirements according to the first step of determination method,  
147 nitrate, nitrite and ammonia test will not be conducted.

148 **Nitrate** To 5 ml in a test tube, cooled in an ice bath, add 0.4 ml of 10 % potassium chloride  
149 solution and 0.1 ml of 0.1 % diphenylamine solution in sulfuric acid and shake well. Add 5 ml  
150 of sulfuric acid dropwise and allow to stand in a water bath at 50°C for 15 minutes. Any colour  
151 produced is not more intense than that of the reference solution (Dissolve a quantity of  
152 potassium nitrate in water to produce a solution containing 1 µg of NO<sub>3</sub> per ml. To 0.3 ml add  
153 4.7 ml of nitrate-free water, repeat the operation using the solution instead of the substance  
154 being examined) (0.000 006%).

155 **Nitrite** To 10 ml in a Nessler cylinder add 1 ml of sulfanilic amide solution (dissolve 1 g in 100  
156 ml of dilute hydrochloric acid ) and 1 ml of 0.1% N-naphthylethylenediamine dihydrochloride  
157 solution and shake well. Any colour produced is not more intense than that of the reference  
158 solution (Dissolve a quantity of sodium nitrite in water to produce a solution containing 1 µg  
159 of NO<sub>2</sub> per ml. To 0.2 ml add 9.8 ml of nitrite-free water, repeat the operation using the solution  
160 instead of the substance being examined) (0.000 002%).

161 **Ammonia** To 50 ml add 2 ml of alkaline mercuric potassium iodide TS and allow to stand for  
162 15 minutes. Any colour produced is not more intense than that of a reference prepared by mixing  
163 1.0ml of ammonium chloride solution (dissolve 31.5 mg of ammonium chloride CRS in  
164 ammonia-free distilled water to produce 1000 ml) with 48 ml of ammonia-free distilled water  
165 and 2 ml of alkaline mercuricpotassium iodide TS (0.00002%).

166 **Bacterial endotoxins** Carry on the test for bacterial endotoxins (1143): less than 0.25 EU per  
167 ml.

168 **Microbial monitoring** Carry out in accordance with the requirements of microbial monitoring

169 in Water for pharmaceutical purposes (0261).

170 Note: ① Preserve in well closed containers. ② Based on risk assessment, if necessary, the non-  
171 volatile substances of this product can be measured (which can be determined with the  
172 following method).

173 Evaporate 100 ml in an evaporating dish previously dried at 105°C to constant weight, to  
174 dryness on a water bath, and dry at 105°C to constant weight, the residue is not more than 1 mg.

### 175 **Appendix 3: Sterile water for injection**

176 **Description** A clear, colourless liquid; odourless.

177 **pH** To 100 ml add 0.3 ml of saturated potassium chloride solution. pH 5.0-7.0 (0631).

178 **Chlorides, sulfates and calcium** Add 50 ml to each of three test tubes, to the first tube add 5  
179 drops of nitric acid and 1 ml of silver nitrate TS; to the second tube add 5 ml of barium chloride  
180 TS; and to the third tube add 2 ml of ammonium oxalate TS; no opalescence is produced in any  
181 one of the tubes.

182 **Nitrate** To 5 ml in a test tube, cooled in an ice bath, add 0.4 ml of 10 % potassium chloride  
183 solution and 0.1 ml of 0.1 % diphenylamine solution in sulfuric acid and shake well. Add 5 ml  
184 of sulfuric acid dropwise and allow to stand in a water bath at 50°C for 15 minutes. Any colour  
185 produced is not more intense than that of the reference solution (Dissolve a quantity of  
186 potassium nitrate in water to produce a solution containing 1 µg of NO<sub>3</sub> per ml. To 0.3 ml add  
187 4.7 ml of nitrate-free water, repeat the operation using the solution instead of the substance  
188 being examined) (0.000 006%).

189 **Nitrite** To 10 ml in a Nessler cylinder add 1 ml of sulfanilic amide solution (dissolve 1 g in 100  
190 ml of dilute hydrochloric acid ) and 1 ml of 0.1% N-naphthylethylenediamine dihydrochloride  
191 solution and shake well. Any colour produced is not more intense than that of the reference  
192 solution (Dissolve a quantity of sodium nitrite in water to produce a solution containing 1 µg  
193 of NO<sub>2</sub> per ml. To 0.2 ml add 9.8 ml of nitrite-free water, repeat the operation using the solution  
194 instead of the substance being examined) (0.000 002%).

195 **Ammonia** To 50 ml add 2 ml of alkaline mercuric potassium iodide TS and allow to stand for  
196 15 minutes. Any colour produced is not more intense than that of a reference prepared by mixing  
197 1.0ml of ammonium chloride solution (dissolve 31.5 mg of ammonium chloride CRS in  
198 ammonia-free distilled water to produce 1000 ml) with 48 ml of ammonia-free distilled water  
199 and 2 ml of alkaline mercuricpotassium iodide TS (0.000 02%).

200 **Carbon dioxide** Mix 25 ml with 25 ml of calcium hydroxide TS in a 50 ml stoppered cylinder,  
201 shake, allow to stand; no opalescence is produced within 1 hour.

202 **Oxidizable substances** Boil 100 ml with 10 ml of dilute sulphuric acid, add 0.10 ml of  
203 potassium permanganate (0.02 mol/L) VS and continue to boil for 10 minutes; the pink colour  
204 does not disappear completely.

205 **Conductivity** Complies with the test for water conductivity (0681).

206 **Non-volatile substances** Evaporate 100 ml in an evaporating dish previously dried at 105°C to  
207 constant weight, to dryness on a water bath, and dry at 105°C to constant weight, the residue is  
208 not more than 1 mg.

209 **Heavy metals** To 100 ml add 19 ml of water, evaporate to 20 ml, cool. Add 2 ml of acetate BS  
210 (pH 3.5) and a quantity of water to 25 ml. Add 2 ml of thioacetamide TS, mix and allow to



211 stand for 2 minutes. Any colour produced is not more intense than that of the solution prepared  
212 in the same manner using 1.0 ml of lead standard solution mixed with 19 ml of water  
213 (0.00001%).

214 **Bacterial endotoxins** Carry on the test for bacterial endotoxins (1143): less than 0.25 EU per  
215 ml.

216 **Other requirements** Complies with the general requirements for injections (0102).

217 **Note:** ① This product is a solvent and rinsing agent. ② Preserve in well closed containers.

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234 **Contact number:**

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